REDACTED APPLICATION

REGULATION 13 PERMIT APPLICATION FOR THE CONSTRUCTION OF A **COAL TO GASOLINE PLANT IN** MINGO COUNTY, WEST VIRGINIA

Prepared for:

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Project No. 0101-08-0324

December 2008

APOTESTA

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Attachment R, Authority Forms and Attachments not applicable to this submission: Attachment S, Title V Permit Revision Information.

Regulation 13 Permit Application - Coal to Gasoline Plant (0101-08-0324), December 2008

SECTION I - III

GENERAL APPLICANT INFORMATION

22. Provide a Detailed Process Flow Diagram(s) showing each proposed or modified emissions unit, emission point and control device as Attachment F.

23. Provide a Process Description as Attachment G.

Also describe and quantify to the extent possible all changes made to the facility since the last permit review (if applicable).

All of the required forms and additional information can be found under the Permitting Section of DAQ's website, or requested by phone.

35A. Certification of Information. To certify this permit application, a Responsible Official (per 45CSR§13-2.22 and 45CSR§30-2.28) or Authorized Representative shall check the appropriate box and sign below.

Certification of Truth, Accuracy, and Completeness

I, the undersigned \boxtimes Responsible Official / \Box Authorized Representative, hereby certify that all information contained in this application and any supporting documents appended hereto, is true, accurate, and complete based on information and belief after reasonable inquiry I further agree to assume responsibility for the construction, modification and/or relocation and operation of the stationary source described herein in accordance with this application and any amendments thereto, as well as the Department of Environmental Protection, Division of Air Quality permit issued in accordance with this application, along with all applicable rules and regulations of the West Virginia Division of Air Quality and W.Va. Code § 22-5-1 et seq. (State Air Pollution Control Act). If the business or agency changes its Responsible Official or Authorized Representative, the Director of the Division of Air Quality will be notified in writing within 30 days of the official change.

Compliance Certification

Except for requirements identified in the Title V Application for which compliance is not achieved, I, the undersigned hereby certify that, based on information and belief formed after reasonable inquiry, all air contaminant sources identified in this application are in
compliance with all applicable read) remembers

ATTACHMENT A

BUSINESS CERTIFICATE

2008

WEST VIRGINIA STATE TAX DEPARTMENT

2010

BUSINESS REGISTRATION CERTIFICATE

ISSUED TO: RANSGAS DEVELOPMENT SYSTEMS, LLC 630 1ST AVE STE 30G **NEW YORK, NY 10016-3799**

BUSINESS REGISTRATION ACCOUNT NUMBER: 2218-0756

This certificate is issued for the registration period beginning: **July 1, 2008 June 30, 2010** This certificate is valid until:

This business registration certificate is issued by the West Virginia State Tax Commissioner in accordance with Chapter 11, Article 12 of the West Virginia Code.

The person or organization identified on this certificate is registered to conduct business in the State of West Virginia at the location above.

This certificate is not transferrable and must be displayed at the location for which issued. ENGAGING IN BUSINESS WITHOUT CONSPICUOUSLY POSTING A WEST VIRGINIA BUSINESS REGISTRATION CERTIFICATE IN THE PLACE OF BUSINESS IS A CRIME AND MAY SUBJECT YOU TO FINES PER W. VA. CODE § 11-9.

TRAVELING/STREET VENDORS: Must carry a copy of this certificate in every vehicle operated by them. CONTRACTORS, DRILLING OPERATORS, TIMBER/LOGGING OPERATIONS: Must have a copy of this certificate displayed at every job site within West Virginia.

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ATTACHMENT B

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AREA MAP

Driving Directions from Charleston, West Virginia to Site

- \triangleright Take US Route 119 South (Corridor G) toward Logan, West Virginia (approximately 50 miles).
- \triangleright Veer right onto the Logan exit (Route 73).
- \triangleright At the end of the exit ramp, turn left at the stop light onto Route 73 toward Logan.
- \triangleright Go approximately 2 miles to the Logan Boulevard until Route 44 South is directly in front of you at the stop light.
- \triangleright Go through the stop light and travel approximately 17 miles on Route 44 South to the intersection of Route 44 and Route 52 (the top of Horsepen Mountain).
- \triangleright Turn left onto 52 South toward Gilbert. Stay straight approximately 8 miles.
- \triangleright Turn right onto Gilbert Creek (County Route 13). From Gilbert Creek, follow the signs that say "Twisted Gun Golf Course".
- \triangleright Turn right onto Ben Creek (County Route 10), still following signs to Twisted Gun Golf Course.
- Stop at the Cobra Natural Resources' guard shack at the top of the mountain. Site visits need to be coordinated with Cobra's as this is an active mining site.

ATTACHMENT C

INSTALLATION AND START UP SCHEDULE

ATTACHMENT C

INSTALLATION AND START UP SCHEDULE

Construction of the facility will begin after receipt of Construction Permit from WVDEP and other necessary regulatory approvals (2010 anticipated). Operations will commence approximately 42 months after the beginning of construction.

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ATTACHMENT D

REGULATORY DISCUSSION

ATTACHMENT D

REGULATORY DISCUSSION

The facility proposed herein, or portions of the facility, may be subject to the following state and federal rules and regulations based on a review of potential air quality regulations. This facility is being designed as a minor source of emissions; therefore, is not subject to Prevention of Significant Deterioration (PSD) or National Emissions Standards for Hazardous Air Pollutants (NESHAPS). Additionally, there is discussion concerning the major process emissions points and the applicable requirements.

1. **STATE REGULATIONS**

45CSR2 – "To Prevent and Control Particulate Air Pollution from Combustion of Fuel in A. Indirect Heat Exchangers"

Sets emission limits on particulate matter mass and opacity from indirect heat exchangers such as the proposed startup boiler. Opacity is generally restricted to no more than 10% while the mass limit is set by a multiplier and the unit's total design heat input (BTU/hr). The start-up boiler will be subject to this rule.

45CSR2A – "Testing, Monitoring, Recordkeeping and Reporting Requirements Under $B.$ 45CSR2"

Provides guidance for complying with the requirements of 45CSR2.

 C_{\cdot} 45CSR4 – "To Prevent and Control the Discharge of Air Pollutants into the Open Air Which Causes or Contributes to an Objectionable Odor or Odors"

The proposed facility will control the discharge of objectionable odors.

45CSR5 - "To Prevent and Control Air Pollution from the Operation of Coal Preparation D. Plants, Coal Handling Operations and Coal Refuse Disposal Areas"

45CSR5 requires the facility to maintain fugitive dust control systems on coal processing equipment. Controls are proposed herein for the coal system.

 $E.$ 45CSR7 – "To Prevent and Control Particulate Matter Air Pollution from Manufacturing Processes and Associated Operations"

Sets emission limits on particulate matter mass and opacity from manufacturing processes. Opacity is generally restricted to no more than 20% while the mass limit is a function of source type and process weight rate as established in Table 45-7A of the rule. Proposed manufacturing process units are defined as type "a" and type "d" units in accordance with 45CSR7. Rule 7 will apply to the limestone handling as a type "a" source and ash handling and process heaters as a type "d" source.

 F_{\cdot} 45CSR7A - "Compliance Test Procedures for 45CSR7 - To Prevent and Control Particulate Matter Air Pollution from Manufacturing Process Operations"

Provides guidance for complying with the requirements of 45CSR7.

G. 45CSR10 - "To Prevent and Control Air Pollution from the Emission of Sulfur Oxides"

Sets emission limits on sulfur dioxide from fuel burning units, manufacturing processes, and combustion of process gas streams. The rule establishes emission limits based on a multiplier and the total design heat input (BTU/hr) for combustion sources. Manufacturing process units generating sulfur dioxide emissions are restricted to an in-stack sulfur dioxide concentration of no more than 2,000 ppm.

H. 45CSR10A – "Testing, Monitoring, Recordkeeping and Reporting Requirements Under 45CSR10"

Provides guidance for complying with the requirements of 45CSR10.

 \mathbf{I} . 45CSR13 - "Permits for Construction, Modification, Relocation and Operation of Stationary Sources of Air Pollutants, Notification Requirements, Temporary Permits, General Permits, and Procedures for Evaluation"

The applicant of the proposed facility is required to obtain a permit prior to the start of construction. This application is being submitted based on the requirements of 45CSR13 to obtain said permit.

 J_r 45CSR16 – "Standards of Performance for New Stationary Sources"

45CSR16 formally adopts NSPS of 45CFR60 which are the federal standards discussed below.

 K_{\cdot} 45CSR20 – "Good Engineering Practice as Applies to Stack Heights"

Facility stack heights will meet the requirements 45CSR20.

L. 45SCR30 – "Requirements for Operation Permits"

> Requires permitting under Title V of the Clean Air Act as needed. This facility is designed to be a minor source under Title V and is deferred from obtaining a Title V Permit at this time. The facility will pay operating fees under Title V since it is subject to NSPS Standards.

45CSR31 – "Confidential Information" M.

> This application contains confidential information. This claim of confidentiality is made in accordance with the requirements of 45CSR31.

$2.$ **FEDERAL REGULATIONS**

40CFR60 Subpart Y - "Standards of Performance for Coal Preparation Plants" A.

Requires written notification of construction and startup, operation of air pollution control equipment, and performance testing and recordkeeping. The proposed coal processing equipment up to the feed hoppers is subject to Subpart Y.

 \mathbf{B} . 40CFR60 Subpart OOO – "Standards of Performance for Nonmetallic Mineral Processing Plants"

Requires written notification of construction and startup, operation of air pollution control equipment, and performance testing and record keeping. The proposed limestone processing equipment up to the feed hoppers is subject to Subpart OOO.

C. 40CFR60 Subpart RRR – "Standards of Performance for Volatile Organic Compounds Emissions from Synthetic Organic Chemical Manufacturing Industry (SOCMI) Reactor Processes"

This facility makes methanol as an intermediate product which is then formed into the final products. Methanol is listed as a material in the regulation.

Subpart NNN - "Standards of Performance for Volatile Organic Compound (VOC) D. Emissions From Synthetic Organic Chemical Manufacturing Industry (SOCMI) **Distillation Operations"**

This facility makes methanol as an intermediate product which is then formed into the final products. Methanol is listed as a material in the regulation.

E. Subpart Kb-Standards of Performance for Volatile Organic Liquid Storage Vessels (Including Petroleum Liquid Storage Vessels) for Which Construction, Reconstruction, or Modification Commenced After July 23, 1984

There are three (3) gasoline tanks and one (1) methanol tank proposed for the site each with two (2) million gallon capacity.

The storage tanks will be subject to this rule. The engineering design for the tanks will incorporate the requirements as stated in Section 60.112b and will include a fixed roof tank with an internal floating roof. The tanks will also be subject to the monitoring,

recordkeeping, reporting, and testing requirements as stated in sections 60.113b., 60.115b., and 60.116b.

 $F₁$ Subpart Dc-Standards of Performance for Small Industrial-Commercial-Institutional **Steam Generating Units**

The start-up boiler will be subject to this rule.

 G_c Subpart VVa—Standards of Performance for Equipment Leaks of VOC in the Synthetic Organic Chemicals Manufacturing Industry for Which Construction, Reconstruction, or Modification Commenced After November 7, 2006

The fugitive emissions of volatile organic compounds occur at several sections of the facility. This facility makes methanol and other products as an intermediate product or within the final product which are listed in the rule.

This regulation applies to pumps, compressors, pressure relief devised, sampling connections, open-ended valves or lines, and flanges or other connectors that operate in VOC service at synthetic organic chemical manufacturing facilities. This facility meets the definition of a SOCMI facility and the portions of the facility in VOC service are subject to the rule. The rule establishes leak definitions for the sources and sets requirements for leak detection and repair.

H. 40CFR60, Subpart XX - Standards of Performance for Bulk Gasoline Terminals

There are two (2) gasoline loading racks at the facility for loading gasoline to either tank trucks or tank rail cars for shipping to market. The following regulation has been identified as being applicable to this source when loading trucks.

40CFR60, Subpart XX—Standards of Performance for Bulk Gasoline Terminals

Sets emission standards for that apply to the loading racks at a bulk gasoline terminal which deliver liquid product into gasoline tank trucks. The proposed system is to be designed to meet the requirements of MACT level control as stated in AP-42, 5.2 Transportation And Marketing Of Petroleum Liquids. The loading and control of the system shall be designed to meet the applicable requirements contained in 60.502, and monitoring, recordkeeping, reporting, and testing as required in sections 60.502, 60.503, and 60.505.

The limit set by section 60.502(b) is a not to exceed of 35 milligrams of total organic compounds per liter of gasoline loaded. The system will be designed to meet this requirement.

I. Part 60 - Standards of Performance for New Stationary Sources Subpart A - General Provisions Part 60.18-General Control Device Requirements.

The flare will be subject to this section. The engineering design for the flare will incorporate these requirements.

$3₁$ **SPECIFIC POINTS OF INTEREST**

 $A₁$ MTG Start-Up/Regeneration Gas Heater Operation (E1)

> The emissions from the operation of the MTG Start-up/Regeneration Gas Heater are emitted through emission point E1. This unit is fired by syngas produced within the plant and the unit is not used to destroy a waste stream from the process. The unit is not in continuous operation as it only operates during process start-up and regeneration of the catalysts.

> 45CSR2 – "To Prevent and Control Particulate Air Pollution from Combustion of Fuel in Indirect Heat Exchangers"

> The purpose of the rule is to prevent and control particulate matter air pollution from indirect heat exchangers. Process heaters are excluded from the definition of indirect heat exchangers.

> 45CSR4 – "To Prevent and Control the Discharge of Air Pollutants into the Open Air Which Causes or Contributes to an Objectionable Odor or Odors"

There are no anticipated objectionable odors from the operation of this source.

45CSR7 – "To Prevent and Control Particulate Matter Air Pollution from Manufacturing Processes and Associated Operations"

This rule establishes emissions limitations for smoke and particulate matter which are discharged from process heaters. This heater (30.0 MM Btu/hr) is in the process of the MTG unit within the facility and is part of the chemical change manufacturing facility classified as Type "d".

For Type "d" operations the allowable particulate emissions rate from the total operation is based on the process weight rate in pounds per hour and Table 45-7A. The process weight rate is the methanol charged to the system which is 526,000 pound per hour. Table 47-7A provides a maximum allowable of 21.2 pounds per hour for the sources. The proposed emission rate of 0.223 pounds per hour meets the requirements under this rule.

45CSR10 - "To Prevent and Control Air Pollution from the Emission of Sulfur Oxides"

The purpose of this rule is to prevent and control air pollution from the emissions of sulfur dioxides. By definition, the process heater is considered Type "b" fuel burning unit (45CSR10-2.8.b.).

For Type 'b' fuel burning units the emission limit is the product of 3.2 and the total design heat inputs for such units in million B.T.U.'s per hour (45CSR10-3.1.e.). Therefore, the allowable sulfur dioxide emissions rate from the total maximum heat input (30 MM Btu/hr) is 96.0 pounds per hour. The syngas fuel used in this system is sulfur-free; therefore, there are no anticipated emissions of sulfur from the combustion of syngas in the system.

40 CFR 60, Subpart Dc: Standards of Performance for Small Industrial-Commercial-**Institutional Steam Generating Units**

This is a process heater and is not a steam generating unit; therefore, the unit is not subject to 40CFR60, Subpart Dc.

40 CFR 60, Subpart RRR: Standards of Performance for Volatile Organic Compound Emissions From Synthetic Organic Chemical Manufacturing Industry (SOCMI) Reactor Process.

The unit burns syngas from the front end of the plant and is not a control device for destruction of volatile organic compounds or total organic compounds; therefore, is not subject to 40CFR60, Subpart RRR.

40 CFR 60, Subpart NNN: Standards of Performance for Volatile Organic Compound Emissions From Synthetic Organic Chemical Manufacturing Industry (SOCMI) Distillation Process.

The unit burns syngas from the front end of the plant and is not a control device for destruction of volatile organic compounds or total organic compounds; therefore, is not subject to 40CFR60, Subpart NNN.

B. MTG Start-Up/Regeneration Gas Heater Operation (E2)

The emissions from the operation of the MTG Start-up/Regeneration Gas Heater are emitted through emission point E2. This unit is fired by syngas produced within the plant and the unit is not used to destroy a waste stream from the process. The unit is not in continuous operations as it only operates during start-up and regeneration of the catalysts.

45CSR2 – "To Prevent and Control Particulate Air Pollution from Combustion of Fuel in Indirect Heat Exchangers"

The purpose of the rule is to prevent and control particulate matter air pollution from indirect heat exchangers. Process heaters are excluded from the definition of indirect heat exchangers.

45CSR4 – "To Prevent and Control the Discharge of Air Pollutants into the Open Air Which Causes or Contributes to an Objectionable Odor or Odors"

There are no anticipated objectionable odors from the operation of this source.

45CSR7 – "To Prevent and Control Particulate Matter Air Pollution from Manufacturing Processes and Associated Operations"

This rule establishes emissions limitations for smoke and particulate matter which are discharged from process heaters. This heater (120 MM Btu/hr) is in the process of the MTG unit within the facility and is part of the chemical change manufacturing facility classified as Type "d".

For Type "d" operations the allowable particulate emissions rate from the total operation is based on the process weight rate in pounds per hour and Table 45-7A. The process weight rate is the methanol charged to the system which is 526,000 pound per hour. Table 47-7A provides a maximum allowable of 21.2 pounds per hour for the sources. The proposed emission rate of 0.89 pounds per hour meets the requirements under this rule.

45CSR10 – "To Prevent and Control Air Pollution from the Emission of Sulfur Oxides"

The purpose of this rule is to prevent and control air pollution from the emissions of sulfur dioxides. By definition, the heater is considered Type "b" fuel burning unit $(45CSR10-2.8.b.).$

For Type 'b' fuel burning units the emission limit is the product of 3.2 and the total design heat inputs for such units in million B.T.U.'s per hour (45CSR10-3.1.e.). Therefore, the allowable sulfur dioxide emissions rate from the total maximum heat input (120 MM Btu/hr) is 384.0 pounds per hour. The syngas fuel used in this system is sulfur-free; therefore, there are no anticipated emissions of sulfur from the combustion of syngas in the system.

40 CFR 60, Subpart Dc: Standards of Performance for Small Industrial-Commercial-**Institutional Steam Generating Units**

This is a process heater and is not a steam generating unit; therefore, the unit is not subject to 40CFR60, Subpart Dc.

40 CFR 60, Subpart RRR: Standards of Performance for Volatile Organic Compound Emissions from Synthetic Organic Chemical Manufacturing Industry (SOCMI) Reactor Process.

The unit burns syngas from the front end of the plant and is not a control device for destruction of volatile organic compounds or total organic compounds; therefore, is not subject to 40CFR60, Subpart RRR.

40 CFR 60, Subpart NNN: Standards of Performance for Volatile Organic Compound Emissions from Synthetic Organic Chemical Manufacturing Industry (SOCMI) Distillation Process.

The unit burns syngas from the front end of the plant and is not a control device for destruction of volatile organic compounds or total organic compounds; therefore, is not subject to 40CFR60, Subpart NNN.

C_{\cdot} MTG Start-Up/Regeneration Gas Heater Operation (E3)

The emissions from the operation of the MTG Start-up/Regeneration Gas Heater are emitted through emission point E3. This is a continuous operating unit, is a process heater, and is fueled with syngas from the process. This is not a control device or a unit which is used to destroy a waste product from the plant. This unit heats an intermediate product prior to further processing.

45CSR2 – "To Prevent and Control Particulate Air Pollution from Combustion of Fuel in Indirect Heat Exchangers"

The purpose of the rule is to prevent and control particulate matter air pollution from indirect heat exchangers. Process heaters are excluded from the definition of indirect heat exchangers.

45CSR4 – "To Prevent and Control the Discharge of Air Pollutants into the Open Air Which Causes or Contributes to an Objectionable Odor or Odors"

There are no anticipated objectionable odors from the operation of this source.

45CSR7 – "To Prevent and Control Particulate Matter Air Pollution from Manufacturing Processes and Associated Operations"

This rule establishes emissions limitations for smoke and particulate matter which are discharged from process heaters. This heater (4.0 MM Btu/hr) is in the process of the MTG unit within the facility and is part of the chemical change manufacturing facility classified as Type "d".

For Type "d" operations the allowable particulate emissions rate from the total operation is based on the process weight rate in pounds per hour and Table 45-7A. The process weight rate is the methanol charged to the system which is 526,000 pound per hour. Table 47-7A provides a maximum allowable of 21.2 pounds per hour for the sources. The proposed emission rate of 0.03 pounds per hour meets the requirements under this rule.

45CSR10 – "To Prevent and Control Air Pollution from the Emission of Sulfur Oxides"

The purpose of this rule is to prevent and control air pollution from the emissions of sulfur dioxides. By definition, the process heater is considered Type "b" fuel burning unit (45CSR10-2.8.b.).

For Type 'b' fuel burning units the emission limit is the product of 3.2 and the total design heat inputs for such units in million B.T.U.'s per hour (45CSR10-3.1.e.). Therefore, the allowable sulfur dioxide emissions rate from the total maximum heat input (4.0 MM Btu/hr) is 12.8 pounds per hour. The syngas fuel used in this system is sulfur-free; therefore, there are no anticipated emissions of sulfur from the combustion of syngas in the system.

The rule exempts fuel burning units having a design heat input under ten (10) million BTU's per hour from section 3 (Sulfur Dioxide Weight Emission Standards for Fuel Burning Units), and sections 6 (Registration), 7 (Permits) and 8 (Testing, Monitoring, Recordkeeping and Reporting).

40 CFR 60, Subpart Dc: Standards of Performance for Small Industrial-Commercial-**Institutional Steam Generating Units**

This is a process heater and is not a steam generating unit; therefore, the unit is not subject to 40CFR60, Subpart Dc.

40 CFR 60, Subpart RRR: Standards of Performance for Volatile Organic Compound Emissions From Synthetic Organic Chemical Manufacturing Industry (SOCMI) Reactor Process.

The unit burns syngas from the front end of the plant and is not a control device for destruction of volatile organic compounds or total organic compounds; therefore, is not subject to 40CFR60, Subpart RRR.

40 CFR 60, Subpart NNN: Standards of Performance for Volatile Organic Compound Emissions From Synthetic Organic Chemical Manufacturing Industry (SOCMI) **Distillation Process.**

The unit burns syngas from the front end of the plant and is not a control device for destruction of volatile organic compounds or total organic compounds; therefore, is not subject to 40CFR60, Subpart NNN.

D. CO2 Purification System Operation (C1)

> The CO2 Purification System prepares CO2 by catalytic stripping for use as blanket gas for the process. The process streams are from the AGR and the gasses from the regeneration of the MTG catalysts. The AGR vents to the CO2 Purification System on a continual basis under normal operations and during startup once that portion of the facility has been started. The MTG catalysts regeneration is a discontinuous operation and only discharges to the CO2 Purification System during regeneration of the catalysts.

> 45CSR4 – "To Prevent and Control the Discharge of Air Pollutants into the Open Air Which Causes or Contributes to an Objectionable Odor or Odors"

There are no anticipated objectionable odors from the operation of this source.

45CSR7 – "To Prevent and Control Particulate Matter Air Pollution from Manufacturing Processes and Associated Operations"

The purpose of the rule is to prevent and control particulate matter air pollution from manufacturing processes and associated operations. Emission point C1 is not anticipated to contain particulate matter.

45CSR10 – "To Prevent and Control Air Pollution from the Emission of Sulfur Oxides"

The purpose of this rule is to prevent and control air pollution from the emissions of sulfur dioxides. Section 10-4, Standards for Manufacturing Process Source Operations contains limits for sulfur dioxide from the manufacturing process in section 4.1 at a limit of an in-stack sulfur dioxide concentration exceeding 2,000 parts per million by volume (ppmv).

The conservative estimate of SO_x contained in the C1 is estimated at 10 ppmy which is the resulting SO_x from the Acid Gas Removal operation. There is no sulfur coming from the regeneration of the catalysts. This facility will meet the requirement contained in this rule.

 E_{\cdot} Flare Operation (B2, C2, E5, and G)

> The flare is the control device for emissions during start-up and normal operation. The flare controls emissions from several sources throughout the process of start-up and normal operation: B2 (raw syngas during startup); C2 (raw syngas during startup); and E5 (tailgas flaring). The largest flow of material to the flare is during startup of the process when the raw syngas is being flared to allow the facility to build up to the proper operating temperatures and pressures. These include emissions points identified as B2 and C2. After the start-up is completed neither B2 nor C2 are venting to the flare. E5 is a discontinuous emission under normal operations when the front end of the plant is down and the tailgas cannot be recycled. Emissions identified as G are for the pilot flame only. These emissions (B2, C2, and E5) occur at the same emissions point (the flare); however, they are not additive in the fact that they do not occur at the same time in the process sequence. B2, C2 and E5 cannot vent to the flare at the same time. G is additive as it occurs on a continual basis to keep the flare operational to allow destruction of materials which would vent to it either during startup or normal operation.

45CSR6 - "Control of Air Pollution from Combustion of Refuse"

45CSR6 set particulate emission standards and opacity requirements for the operation of activities involving incineration of refuse which includes the operation of a flare. This facility employs the use of a flare at start-up of operation and during normal operations. The flare is typically being used to flare syngas which results from gasification or MTG

off gas which occurs when the front end of the plant is down and the MTG process is in operation.

The rule sets the allowable emissions standard based on a formula:

Emissions (lb/hr) = $F \times$ Incinerator Capacity (tons/hr)

Where, the factor, F, is as indicated in Table I below:

Table I Factor F, for Determining Maximum Allowable Particulate Emissions.

Incinerator Capacity	Factor F
Less than $15,000$ lbs/hr	5.43
B. 15,000 lbs/hr or greater	מד ה 2. I 2

The feed to the flare varies depending on the operation of the plant. The maximum anticipated flow to the flare is during start-up of the facility. Flare feed at startup is anticipated to be 140 tons per hour (280,000 lbs/hr) which would yield an allowable emission value of 380.8 pounds per hour. The feed is syngas that is being produced but cannot be processed further down the line into methanol and therefore it is sent to the flare. The anticipated emissions value at this time is 0.223 lbs per hour which meets the requirement of this rule.

Emissions points which are considered flared emissions either during startup or normal operations along with the respective emissions values during operation are as follows:

The rule establishes an opacity limit under sections 4.3 at 20% with exceptions under 4.4 where 4.3 shall not apply to smoke which is less than forty percent (40%) opacity, for a period or periods aggregating no more than eight (8) minutes per start-up, or six (6) minutes in any sixty (60)-minute period for stoking operations periods of and 4.4.

There are no anticipated opacity issues with the operation of the flare. The flare will be specifically designed for the process and will meet the requirements for particulate matter emissions and opacity as stated within 45CSR6.

40 CFR 60, Subpart RRR: Standards of Performance for Volatile Organic Compound Emissions From Synthetic Organic Chemical Manufacturing Industry (SOCMI) Reactor Process.

The process is a synthetic organic chemical manufacturing facility and is subject to 40CFR60, Subpart RRR. The emissions from the process during startup of the gasifiers and the operations of the regeneration system are controlled by the flare. The flare must meet: Section 60.702(b) standards; Section 60.703, monitoring of emissions and operations under section $60.703(b)$; Section $60.704(c)$ for test methods and procedures the flare must meet the requirements of 40CFR60, Subpart A, General Provisions, Section 60.18; and Section 60.705(a), (b), (b).(3) for reporting and recordkeeping requirements. The flare will be designed to meet the requirements for this section. This includes the requirement for no visible emissions except for a period of 5 minutes during any 2 consecutive hours. This visible emissions requirement is more stringent than the requirement in 45CRS6.

40 CFR 60, Subpart NNN: Standards of Performance for Volatile Organic Compound Emissions From Synthetic Organic Chemical Manufacturing Industry (SOCMI) **Distillation Process.**

The process is a synthetic organic chemical manufacturing facility and is subject to 40CFR60, Subpart NNN. The emissions from the process during startup of the gasifiers and the operations of the regeneration system are controlled by the flare. The flare must meet: Section 60.662(b) standards; Section 60.663, monitoring of emissions and operations under section $60.663(b)$; Section $60.664(d)$ for test methods and procedures the flare must meet the requirements of 40CFR60, Subpart A, General Provisions, Section 60.18; and Section 60.665(a), (b), (b)(3) for reporting and recordkeeping requirements. The flare will be designed to meet the requirements for this section.

F. Startup Boiler Operation (F)

> The emissions from the operation of the startup boiler are emitted through emission point F.

> 45CSR2 – "To Prevent and Control Particulate Air Pollution from Combustion of Fuel in Indirect Heat Exchangers"

> This rule establishes emissions limitations for smoke and particulate matter which are discharged from fuel burning units. By definition of this rule the natural gas fueled startup boiler is classified as Type b fuel burning unit (45CSR2-2.10.b.).

> For Type 'b' fuel burning units the emission limit is the product of 0.09 and the total design heat inputs for such units in million B.T.U.'s per hour, provided however that no more than six hundred (600) pounds per hour of particulate matter shall be discharged into the open air from all such units. Therefore, the allowable particulate emissions rate

from the total maximum heat input (81.84 MM Btu/hr) is 7.36 pounds per hour. The proposed emissions of 0.61 pounds per hour meet the requirements under this rule.

45CSR4 - "To Prevent and Control the Discharge of Air Pollutants into the Open Air Which Causes or Contributes to an Objectionable Odor or Odors"

There are no anticipated objectionable odors from the boiler as the boiler is fueled by natural gas.

45CSR7 – "To Prevent and Control Particulate Matter Air Pollution from Manufacturing Processes and Associated Operations"

The purpose of the rule is to prevent and control particulate matter air pollution from manufacturing processes and associated operations. The startup boiler (fuel burning unit) is exempt from the requirements of this rule.

45CSR10 – "To Prevent and Control Air Pollution from the Emission of Sulfur Oxides"

The purpose of this rule is to prevent and control air pollution from the emissions of sulfur dioxides. By definition, the startup boiler is considered Type b fuel burning unit $(45CSR10-2.8.b.).$

For Type 'b' fuel burning units the emission limit is the product of 3.1 and the total design heat inputs for such units in million B.T.U.'s per hour (45CSR10-3.1.e.). Therefore, the allowable sulfur dioxide emissions rate from the total maximum heat input (81.84 MM Btu/hr) is 253.7 pounds per hour. The proposed emissions of 0.31 pounds per hour meet the requirements under this rule.

As stated under 45CSR10-10.3., the owner or operator of a fuel burning unit(s) which combusts natural gas, wood or distillate oil, alone or in combination, shall be exempt from the requirements of Section 8. The startup boiler is exempt from 45CSR10-8. Testing, Monitoring, Recordkeeping and Reporting.

40 CFR 60, Subpart Dc: Standards of Performance for Small Industrial-Commercial-**Institutional Steam Generating Units**

The startup boiler is subject to 40 CFR 60, Subpart Dc under the applicability requirements of $§60.40c(a)$. Subpart Dc does not have an emission standard for combusting natural gas and there are no monitoring requirements for boilers combusting natural gas. The startup boiler is subject to the Reporting and Recordkeeping requirements of $\S60.48c(a)$, (a)(1), (a)(3), (g) and (i).

ATTACHMENT E

PLOT PLAN

 $\mathcal{L}^{\text{max}}_{\text{max}}$ and $\mathcal{L}^{\text{max}}_{\text{max}}$

ATTACHMENT F

PROCESS FLOW DIAGRAM

XREF Files: IMAGE Files:
File: S:\C3D-Projects\08-0324 - TRANSGAS\A08-0324-02.dwg
Plot Date/Time: May 28, 2009 - 10:01am Plotted By: MBSankoff

XREF Files: MAGE Files:
File: S:\C3D-Projects\08-0324 - TRANSGAS\A08-0324-02.dwg
Plot Date/Time: May 28, 2009 - 10:01am Plotted By: MBSankoff

ATTACHMENT G

PROCESS DESCRIPTION

Attachment G General Process Description

The proposed facility will produce gasoline and LPG from coal through the processes of gasification of the coal to produce a syngas, modify the syngas to methanol and then unitize the methanol as the feed to the methanol to gas process. The following is a description of the start-up and operation of the process. A more detailed version of this information is provided for additional detail for the agency review and understanding of the process in preparing the permit for the facility.

Start-up of the Plant

Start-up of the operation will sequence through the following steps:

PDQ Start-up Overall Sequence

The start-up overall sequence gives a summary of the overall operating procedures, for the gasification (one stream), water treatment unit and the raw gas treatment units (one stream). The start-up will be described from "cold, air-filled" to "fully operational". Prior to start the PDO-Gasifier with feedstock the water/steam system has to be filled with boiler feed and the quench water system with fresh water. Both systems are to be warmed up. The downstream Units as Sour Shift and Gas Cooling, Sulfur Removal and Sulfur Recovery/Tail gas Treatment are to be transferred into hot /cold condition ready to receive washed and water saturated raw gas and Sour Gas. The consumers of the treated syngas are ready for syngas take over. All Utilities as Cooling Water, HP and LP Nitrogen systems, Oxygen system, Instrument Air and Chemical Dosing Stations are in operation, The Start-up Boiler is started and the steam system is ready for take over. The Startup Boiler operation is scheduled approx. 4 times /year, each 96 hours. For emissions see Emission Point F. Waste Water Treatment, Condensate System and Boiler Feed Water Preparation are in operation.

The Coal Milling and Drying system consists of five identical 100 % process lines, four in operation, one in stand by, feeding two Coal Pressurization and Feeding lines. In this description only one mill is described. In case 2 Gasifiers are in operation 4 Coal Milling & Drying Units are in operation. In this case for Coal Milling & drying the Emission Points A1/1 & A1/2 & A1/3 & A1/4 are active.

Main start-up steps are as followed: starting of the auxiliary systems, filling of raw coal and lime-stone bunkers; Inertizing of the mill and the gas circulating system; adjusting the oxygen concentration; if necessary supplying N2 to the system to reduce the oxygen content below 8% on volume; pre-heating the system, if necessary the start-up burner of the inert gas generator can be ignited to support the tracing of the pulverized coal bag house. In this situation the emissions at Emission Point $A1/x$ are similar as for stand by operation (emission values approx. 10% of operation emissions); and start-up of the mill and the coal feeders.

The shutdown procedure basically follows the reversed order of the start-up procedure. It is possible to operate each line on hot or inerted stand-by.

Coal Pressurization and Feeding

The coal pressurization and feeding units are started through the following sequence: Checking that all manholes, blinds, etc. are securely placed, block valves are in correct position, instrument isolation valves are opened, instrument air is turned on, nitrogen purges are started, and radioactive level instruments are unblocked. The main start-up steps are listed below: Start operation of the Blow Back System of Feed Dust Bunker Filter; Feed Dust Bunker is filled with specified coal dust; Start supply of CO2 or N2 to the aeration cones of Feed Dust Bunker; Start supply of CO2 or N2 to the Aeration cones of Feed Vessel, Start-up; Vessel and Lock Hoppers are in operation; The Start-up Vessel for coal dust recycle is inert and ready for use: Adjust the differential pressure controller between Feed Vessel and Gasifier; and Transfer Coal Dust Feed Stock via Lock Hoppers into the Feed Bin.

These systems feed to the gasifier section of the plant.

Gasifier Section

For startup to occur the pre-conditions must be met: liquid carrying tanks and vessels are filled with water/boiler feed water; all water circulations, the chemical dosing system and the vacuum pump are to be started; the waste water treatment unit is ready to receive process water; the oxygen system is tightness tested and under operation pressure with oxygen; and the flare system is ready for use.

Warming-up of the Sour CO-Shift Catalyst

The sour CO-Shift catalysts achieve the maximum activity in the sulphided state. Therefore, prior to the first start-up, they are to be treated with sulfur. Normal preheating of the catalyst is carried out in a dry inert gas (nitrogen) which is recycled via the start-up compressor and preheated by MP-Steam and the electrical Start-up Heater. The shift reaction will start at a temperature of around 200°C. The catalyst normally operates between 230°C and 500°C. Temperatures up to 550°C are acceptable.

Preparation of the Sour Water Stripping

The Stripper has to be pressurized by nitrogen and warmed up by steam. It is in hot standby condition under total reflux.

Start-up of the H2S/CO2 Removal Unit

The aim of the start-up operation of the Acid Gas Removal Unit is to cool down the methanol solvent inventory to a level of about -35 °C for absorption of the acid gas components H2S and CO2. Before any gas enters the Unit, absorption section has to be pressurized with nitrogen. The normal gas path to the downstream synthesis unit is closed

and the gas path to the flare is opened. The pressure in the absorption section will be controlled on flow control to the flare (Emission Point C2). Once the normal operating temperatures are established, levels have stabilized and the operating pressures in the vessels of the regeneration section are set, the unit is ready for take over of feed gas. The H2S stripped off from the solvent in the thermal desorber is sent as acid gas to the Sulfur Recovery Unit, the CO2 produced in the flash regeneration section of the Acid Gas Removal Unit is sent to the CO2 purification section before it can be pressurized and reused as carrier or sluicing gas in the gasification unit or sent to the atmosphere (Emission Point C1).

Preparation of the Sulfur Recovery Unit

The SRU represents a typical Claus unit comprising one thermal and two catalytic stages. The SRU can be operated by air and/or by O2 enriched air. Feedstock is acid gas generated in the H2S/CO2 removal plant, a small stream of sour gases of the sour water stripping unit and flash gases yielded in the slurry flash vessels and different other venting points. The thermal and the two catalytic stages have to be warmed up to operation temperature. All sulfur carrying lines and vessels are to be carefully warmed up by steam tracing. The unit can remain in hot stand-by position until sour gas is available.

Preparation of the Tail Gas Treatment Section

The Tail Gas Treatment Section has to be heated up to 200 °C by using nitrogen and the electrical tail gas heater. The unit is now ready for Claus tail gas intake.

Warming-up/Start-up of PDQ Gasification

Warming up of the PDQ Gasification has to be carried out in the following sequence: Warming-up of the Gasifier Steam system has to be carried out; The water quench part of the Gasifier, the water circulation of the Venturi Scrubber and the Wash Column and the Slurry Expansion and Filtrate Preheating will be warmed-up; Ignition of the Start-up Burner; Pressurizing of the Gasifier up to start-up pressure for the first Coal Dust Burner; Line up the raw gas route; Start coal dust recycle of the first burner to the start-up vessel. For emissions and duration of emissions at the start-up vessel filter 1121-F004 see Emission Point A2/1; A2/2; Start Coal Sluicing and Feeding to keep sufficient coal in the Feed Vessel. For emission during expansion of the sluice vessels see Emission Point B1/1;B1/2; Start the Coal Dust Burners with the required coal dust quantity; Retract the Start-up burner; Increase the gasifier pressure to normal operating pressure. The produced gas is routed via the Scrubber 1161-C001 to the flare. For gasifier start-up emissions see Emission Point B2; Start sending raw gas to the cooling train of the CO Shift Conversion, the CO2/H2S Removal and the pressure controller downstream of the CO2/H2S Removal to the flare; Start slag sluicing program, (Slag cooling, slag sluice support pump, slag extractor and slag disposal.) Adjust Slurry Expansion flow and keep Gasifier base level constant via the level control valve; Start the Belt Filter Press; Start filter cake disposal; Adjust the CO2/H2S Wash Unit. Route the yielded sour gas to the sour gas flare first, until the sour gas quality (H2S content) is suitable to be used in the Sulfur recovery Unit; Start sending raw gas stepwise via the CO Shift Reactor; Start-up the Sulfur Recovery Unit with sour gas; Start-up the Tail Gas Treatment Section with tail gas.

Shut Down Depressurization of the PDQ Unit

After tripping of the coal dust burner depressurization of the PDQ unit can be initialized. The depressurization time is related to the cooling-down time of the quench water circulation and the water circulation around the venture scrubber / wet scrubber to prevent excess of water evaporation. Forced temperature reduction can be reached by cooling of the expanded quench water to the clarification unit in the cooling water cooler down-steam of the third expansion vessel. The clarified cooled quench water has to be routed back to the PDQ quench circuit bypassing the pre-heater of the first quench Vessel.

Start-up Methanol Plant

Clean synthesis gas must be available for the start-up downstream of the acid gas removal. The synthesis gas is fed via the make up gas (MUG) compressor. The steam pressure will be adjusted to the required temperature profile of the methanol reaction. Above 10 bar methanol will start to condense in the HP methanol separator and so synthesis gas will be removed from the system by the reaction. The feed rate from the MUG compressor has to be increased to compensate the losses of H2 and CO. If the liquid level in the HP methanol separator is sufficient high then commission the let down system to the methanol flash drum. If the concentration of Nitrogen and /or CH4 in the loop reaches the flow sheet values start the purge from the synthesis loop. The purge is routed to the hydrogen recovery unit.

Start-up Methanol to Gasoline Plant

Start-up of the MTG unit from cold conditions requires that the system be nearly oxygen free. During start-up of the methanol to gasoline section there are flue gas emissions from the fired heaters (Emission Points E1 and E2) used during heating up the DME Reactor and the MTG Reactor. The columns and reflux accumulators of the product fractionation section will be filled with pre-stored gasoline. For pressurizing of the Stabilizer prior to feed in, fuel gas will be used. During Start-up of the product fractionation section no emissions are expected. Fuel gas produced in the Stabilizer will be routed to the fuel gas system (OSBL).

During Start-up of the heavy gasoline treatment section emissions due to use of the HGT Charge Heater (Emission Point E3) appear. Gas purged from the HGT Section during take over of heavy gasoline from product fractionation section and stabilization of the heavy gasoline section will be routed to the fuel gas system (OSBL).

Normal Operation of the Plant

Once the start-up sequence is completed the process operations are continued with the entire process being active. The coal feed and limestone are sent to the preparation facilities from the coal and limestone yard, the material is then fed through the preparation/grinding facilities and fed to the PDQ gasification systems. The products of the gasification system are the syngas and the ash material. The gasification system is basically a high temperature partial oxidation process for converting the coal into a syngas of carbon monoxide and hydrogen. The ash material is mostly a slag/aggregate and a smaller fraction of fly ash. The ash materials are stored in the storage building (slag/aggregate) or in the storage silo (fly ash). The ash materials will be shipped off site.

The syngas passes through the remaining part of the syngas treatment units which include the CO Shift (enriches the syngas into a hydrogen rich stream), the Acid Gas Removal (removes the acid gases CO2, H2S, COS, mercury, and HCN), Mercury Removal Systems (removal of remaining mercury with the majority of mercury removed in the Acid Gas Removal System), CO Purification (purifies the CO stream), Sulfur Recovery Unit (processes H2S, CO2 and NH3 containing sour gas from stripping of process condensates, H2S, COS, CO2 and HCN containing acid gas, NH3 and HCN are decomposed to sulfur containing components are converted to elemental sulfur which is a product of the plant), Sour Water Stripping (removal of H2S, CO2, NH3 and HCN), and then to the Methanol Production Unit which transfers the syngas to methanol which is the intermediate product of the plant. Air separation units are required to provide nitrogen and oxygen for the process steps. Each step of the process is necessary to take the original produced syngas and transfer it to the next process at the grade required to continue down the line for further processing to maximize production of methanol. Differing temperatures and pressures are required throughout the processes for proper operation of each step of the system.

The methanol is then sent to the Methanol to Gasoline Treatment (MGT) unit to be converted to gasoline and LPG products and the remainder of the material is converted to water.

Portions of the process also produce steam (high pressure and low pressure). The steam is used in the processes to provide heat as needed, to provide steam for steam driven sections of the process, or converted into electrical power as needed to drive electrical system. There will be a cooling tower to allow for the process water to be cooled and reentered into the process. Electric power, as produced, will be utilized at the facility. Additional needs for electric will be supplied by utility connections. There are no sales of electric power.

The plant will feed approximately 346 tons per hour, 8,304 tons per day, and 3,030,960 tons per year into the system and produce 18,000 barrels of gasoline per day and 6,570,000 barrels per year with a barrel being 42 gallons. Raw materials for the plant will be trucked into the site. Product gasoline and LPG will be removed from the site via the preferred method of loading to railcar and some allowance is made for trucking for the delivery to markets.

ATTACHMENT H

MATERIAL SAFETY DATA SHEETS

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Categories

LPG Regulators Peco Filters Natural Gas & Sperryn Signs & Markers Cast Iron Products Algas SDI Vaporisers Gas Installation **LPG Leisure Adaptors** Gas Controls & Pilots LPG Specialists' Adaptors Rail (Gas) Cocks Rochester Gauges Infrared Heating Red Dragon - Weed Flamers Gas Pex Tube Camping Lanterns Brass Fittings - SAE Brass Fittings - BSP **Tubing Tool Range** Industry Specialty Tools Rock LP Gas Products

Please note: This MSDS current at publication date below. This site does not undertake to keep most recent version.

Source: Energy Safety Handbook - LP Gas Edition **MATERIAL SAFETY DATA SHEET This material safety data sheet (MSDS).**

1. Is produced by Australian Liquefied Petroleum Gas Association Ltd. Pty. for use by its members.

2. Has been produced following the principles and recommendations outlined in the Worksafe Australia Guidance Note for completion of a material Safety Data Sheet, Second Edition, Sydney, February 1990.

MSDS No: 0000/1

Date issued: December 1991

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HYDROGEN SULFIDE

ATTACHMENT I EMISSION UNIT TABLE

For Emission Units (or Sources) use the following numbering system:1S, 2S, 3S,... or other appropriate des

² For Emission Points use the following numbering system:1E, 2E, 3E, ... or other appropriate designation.

³

^s New, modification, removal
⁴ For <u>C</u>ontrol Devices use the following numbering system: 1C, 2C, 3C,... or other appropriate designation.

ATTACHMENT J

EMISSION POINTS DATA SUMMARY SHEET

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WVDEP-OAQ Revision 03-2007 DRAFT

WVDEP-OAQ Revision 03-2007 DRAFT

WVDEP-OAQ Revision 03-2007 DRAFT

The EMISSION POINTS DATA SUMMARY SHEET provides a summation of emissions by emission unit. Note that uncaptured process emission unit emissions are not typically considered to be fugitive and must be accounted for on the a activities.

T
Please add descriptors such as upward vertical stack, downward vertical stack, horizontal stack, relief vent, rain cap, etc.
2. Letinate by 1991 if youthout a sertionary - Otherwise, association accuracy shad terms unati

Indicate by "C" if venting is continuous. Otherwise, specify the average short-term venting rate with units, for intermittent venting (ie., 15 min/hr). Indicate as many rates as needed to clarify

3 List all regulated air pollutants. Speciate VOCs. including all HAPs. Follow chemical name with Chemical Abstracts Service (CAS) number. LIST Acids, CO, CS2, VOCs, H2S, Inorganics, Lead, Organics, O3, NO, NO2, SO₂, SO₃

4 Give maximum potential emission rate with no control equipment operating. If emissions occur for less than 1 hr, then record emissions per batch in minutes (e.g. 5 lb VOC/20 minute batch).

5 Give maximum potential emission rate with proposed control equipment operating. If emissions occur for less than 1 hr, then record emissions per batch in minutes (e.g. 5 lb VOC/20 minute batch).

 6° Indicate method used to determine emission rate as follows: MB = material balance; ST = stack test (give date of test); EE = engineering estimate; O = other (specify).

 $\boldsymbol{7}$ Provide for all pollutant emissions. Typically, the units of parts per million by volume (ppmy) are used. If the emission is a mineral acid (sulfuric, nitric, hydrochloric or phosphoric) use units of milligram per dry cubi if the pollutant is SO₂, use units of ppmv (See 45CSR10).

frequency of venting (e.g., 5 min/day, 2 days/wk).

WVDEP-OAQ Revision 03-2007

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WVDEP-OAQ Revision 03-2007

 1 Give at operating conditions. Include inerts.
 2 Release height of emissions above ground level.

WVDEP-OAQ Revision 03-2007

ATTACHMENT K

FUGITIVE EMISSIONS DATA SUMMARY SHEET

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FUGITIVE EMISSIONS DATA SUMMARY SHEET

The FUGITIVE EMISSIONS SUMMARY SHEET provides a summation of fugitive emissions. Fugitive emissions are those emissions, which could not reasonably pass through a stack, chimney, vent or other functionally equivalent opening. Note that uncaptured process emissions are not typically considered to be fugitive, and must be accounted for on the appropriate EMISSIONS UNIT DATA SHEET and on the EMISSION POINTS DATA SUMMARY SHEET.

Please note that total emissions from the source are equal to all vented emissions, all fugitive emissions, plus all other emissions (e.g. uncaptured emissions).

VOCs, H₂S, Inorganics, Lead, Organics, O₃, NO, NO₂, SO₃, etc. DO NOT LIST CO₂, H₂, H₂O, N2, O₂, and Noble Gases.
VOCs, H₂S, Inorganics, Lead, Organics, O₃, NO, NO₂, SO₃, etc. DO NOT LIST CO₂, H₂

batch).
⁴ Indicate method used to determine emission rate as follows: MB = material balance; ST = stack test (give date of test); EE = engineering estimate; O = other (specify).

Revision 12/01/2000

ATTACHMENT L

EMISSIONS UNIT DATA SHEET(S)

Affected Source Sheet

Source Specific Emissions Data: Solid Materials Sizing, **Handling and Storage Facilities**

Required Information Regarding Dust Control Equipment Measures

- $1¹$ If water or chemical sprays are to be used on conveyors, transfer points, stockpiles, etc... for dust control, the location of all spray bars or spray systems should be shown on the plot plans and/or line drawings. The following information should be provided for each spray system:
	- a. Design water flow through spray bar
	- Type and amount of chemicals used and the mix ratio of chemical to b. water used at the sprays.
	- Methods employed to winterize sprays (e.g. keep sprays from freezing C. and becoming inoperable during cold weather)
- $2.$ A detailed written description should be submitted of dust control measures/programs that will be employed on haul roads and in areas of vehicle activity around material stockpiled. The haulways and areas to be treated should be shown by shading or similar description on the plant plan. The following points should be specifically addressed:
	- Equipment (e.g. water trucks, fixed spray bars, wheel and truck $a₁$ underbody washers, etc...) that will be used in this dust control program.
	- Frequency of application of water and chemical to roads and stockpile b. areas during dry periods.
	- Amount of chemical suppressants to be used, if applicable, in pounds or C. gallons per square yard of surface to be treated.
	- Type of haulroad or haulway surface(s) that will be maintained (e.g. \mathbf{d} . coarse gravel, reddog, etc...)
	- Approximate maximum length of haulroads (miles or feet). е.
	- f_{\perp} Maximum daily truck traffic on haulroads (number of trucks).
- $3.$ If full or partial enclosures are to be used to minimize dust entrainment, a drawing of each such enclosure should be submitted (for example at truck dump bins, breakers, conveyor transfer points).
- If particulate control devices such as baghouses or scrubbers are to be used, $\overline{4}$. complete an appropriate Air Pollution Control Device Sheet and furnish a drawing showing details of enclosures and ductwork associated with these control systems.

AFFECTED SOURCE SHEET

Source Specific Emissions Data: Solid Materials Sizing, Handling, and Storage Facilities

Plot Plan(s) and Line Drawing(s)

- Finish the plot plan(s) of the plant area which contains sufficient detail to show a. the scaled layout of the equipment involved in each materials handling system (e.g. conveyors, transfer points, crushers, screens, bins, stockpiles, truck dump bins, etc...). Show equipment or buildings described in other sections of this application on the plot plan as appropriate. The guidelines for Plot Plans should be followed to the extent possible.
- $b₁$ Furnish the line drawing(s) or schematic(s) showing each component or facet of each materials handling system (e.g. conveyors, transfer points, stockpiles, crushers, screens, bins etc...). Show process equipment described in other sections of this application as needed for clarity.
- On the line drawing(s) or schematic(s) furnished in accordance with item (b) C. assign an ID number to each conveyor, transfer point (including truck, barge and rail car loading/unloading etc...), storage structure, stockpile, crusher, and screening unit. If any equipment is shown on the line drawing(s) which was described in other sections of this application, use the ID numbers assigned to the equipment in those other sections and indicate equipment name or type (e.g. rotary dryer, vertical kiln etc...)
- d_{-} To the extent possible, note the numbers assigned for equipment and storage facilities as per item (c) on the Plot Plans(s).
- The assigned ID numbers for equipment and transfer points must be used to e. complete Tables 1, 2, and 3 following.

ID Number	OS1	B1 & B2	FH1,3,5,7,9		
Affected Source Name	OS ₁	B1 & B2	FH1,3,5,7,9		
Type Storage ¹	OS	B	B		
Material Stored	Coal	Coal	Coal		
Typical Moisture Content (%)	5	5	5		
Avg % of material passing 200 mesh sieve	5	5	5		
Maximum Total Yearly Throughput in storage (tons)	3,030,960	3,030,960	3,030,960		
Maximum Quantity of Material in Storage ² (tons)	40,000	50 each	50		
Maximum Stockpile Base Area (sq. ft.)	196,020	\mathbf{NA}	NA		
Maximum Stockpile height (ft)	90	NA.	NA		
Type dust controls during storage ³	FE/BH	PE	FE		
Method of material load- in to bin or stockpile ⁴	ST	TD	SS		
Type dust controls during load-in ⁵	FE/BH	PE	VF		
Method of material load- out to bin or stockpile ⁴	UC	UC	Feeder to Roller Mill		
Type dust controls during load-out ⁵	FE	FE	FE		

Table 1: Affected Storage Activity

ID Number	OS ₂	FH2,4,6,8,10	FCS	SSP	SB
Affected Source Name	OS ₂	FH2,4,6,8,10	FCS	SSP	SB
Type Storage ¹	OS	\bf{B}	B	Enclosed Pile	B
Material Stored	Limeston e	Limestone	Fly Ash	Aggregate/Slag	
Typical Moisture Content (%)	$\mathbf{1}$	1	1	$\mathbf{1}$	1
Avg % of material passing 200 mesh sieve	5	5	100	5	5
Maximum Total Yearly Throughput in storage (tons)	166,440	166,440	61,320	604,440	604,440
Maximum Quantity of Material in Storage ² (tons)	20,000	50 Each	200	200,000	100
Maximum Stockpile Base Area (sq. ft.)	87,120	NA	NA	Building	NA
Maximum Stockpile height (ft)	25	NA.	NA	NA	NA.
Type dust controls during storage ³	BH	PE	FE	FE	FE
Method of material load- in to bin or stockpile ⁴	TD	SS	SS	SS	SS.
Type dust controls during load-in ⁵	N	VF	FE	FE	FE
Method of material load- out to bin or stockpile ⁴	UC	Feeder to Roller Mill	FC	UC	FC
Type dust controls during load-out ⁵	FE	FE	PE	FE	PE

Table 1: Affected Storage Activity (Continued)

ID Type Number Conveyor or		Material Handled [(Note nominal size of	Material Conveying or Transfer Rate		Type Dust Control	Approximate Material Moisture		
	Transfer Point ⁶	material transferred] ⁷	Max. TPH	Max. TPY	Measures	Content (%)		
Coal Conveyors								
BC1	BC	$+1\frac{1}{2}$ "	346	3,030,960	PE	5		
BC2	BC	$+1\frac{1}{2}$ "	346	3,030,960	PE	5		
BC ₃	BC	$+1\frac{1}{2}$ "	346	3,030,960	PE	5		
BC4	BC	$+1\frac{1}{2}$ "	346	3,030,960	PE	5		
BC5	BC	$+1\frac{1}{2}$ "	346	3,030,960	PE	5		
BC ₆	BC	$+1\frac{1}{2}$ "	346	3,030,960	PE	5		
Ash/Aggregate Conveyors								
BC7	BC	$+1\frac{1}{2}$ "	69	604,440	PE	1		
BC ₈	BC	$+1\frac{1}{2}$ "	100	604,440	PE	$\mathbf{1}$		
BC ₉	BC	$+1\frac{1}{2}$ "	100	604,440	PE	$\mathbf{1}$		
	Filter Cake Conveyor							
BC10	BC	$+1\frac{1}{2}$	100	604,440	PE	$\mathbf{1}$		
	Limestone Conveyors							
BC11	BC	$+1\frac{1}{2}$ "	100	166,440	PE	$\mathbf{1}$		
BC12	BC	$+1\frac{1}{2}$ "	100	166,440	PE	$\mathbf{1}$		
Coal Transfer Points								
TPC1	07	$+1\frac{1}{2}$ "	346	3,030,960	PE	5		
TPC ₂	07	$+1\frac{1}{2}$ "	346	3,030,960	PE	5		
TPC3	OTH ₃	$+1\frac{1}{2}$ "	346	3,030,960	FE	5		
TPC4	OTH ₃	$+1\frac{1}{2}$ "	346	3,030,960	FE	5		
TPC5	01/OTH4	$+1\frac{1}{2}$ "	346	3,030,960	$\bf FE$	5		
TPC6	01/OTH4	$+1\frac{1}{2}$ "	346	3,030,960	$\bf FE$	5		
TPC7	01/OTH5	$+1\frac{1}{2}$ "	346	3,030,960	PE	5		
TPC8	01/OTH5	$+1\frac{1}{2}$ "	346	3,030,960	PE	5		
TPC9	OTH ₅	$+1\frac{1}{2}$ "	346	3,030,960	PE	5		
TPC10	OTH7	$+1\frac{1}{2}$ "	346	3,030,960	FE	5		

Table 2: Conveying and Transfer

(material.wpd)
WVDEP-OAQ:Revised 11/95

Table 3: Crushing and Screening

Describe method of determining emissions and dust control efficiencies (if by test on a similar unit provide report, if by emission factor reference emission factors):

Emissions from crushing operations were estimated using factors from General Permit G10-B or AP-42. Control efficiencies were obtained from an DAQ guidance document included with the 1993 Title V Certified Emission Statement.

- $\mathbf{1}$ Type Storage - Code as follows: (Note capacity of each bin, building or enclosure)
	- OS Open Stockpile
	- **B** -Bin or Storage Silo (full enclosure)
	- SB Storage Building (full enclosure)
	- E- 1 Enclosure (walls but no top)
	- **SWF-** Stockpiles with wind fences
	- OTH- Other Specify in footnote or attachment
- $2.$ Give maximum and average quantity of material in storage at any given time (e.g. silo capacity, stockpile size, etc...)
- 3. TYPE DUST CONTROLS DURING STORAGE If storage is by other than by bin or full enclosure Code as follows:
	- $N -$ **None**
	- **WS-** Water Sprays
	- $c-$ Spraying with chemical surfactant
	- OTH- Other Specify in footnote or attachment
- METHOD OF PLACING MATERIAL ONTO STOCKPILE OR INTO BINS OR LOADING 4. **OUT FROM STOCKPILES OR BINS - Code as follows:**
	- $c -$ **Clamshell**
	- TD-**Truck Dumping**
	- FE-**Front Endloader**
	- ST-**Stacking Tubes**
	- MS-**Mobile Conveyor - Stacker**
	- SS-**Stationary Conveyor - Stacker**
	- P. **Pneumatic Conveyor - Stacker**
	- FC-**Fixed Height Chute from bins**
	- TC-**Telescoping Chute from bins**
	- UC-Under-pole or under-bin reclaim conveyor
	- RC-Reclaim Conveyor (rake or bucket reclaim conveyor reclaiming from surface of stockpile)
	- OTH- Other Describe in a footnote or attachment
- 5. TYPE DUST CONTROLS - Code as follows:
	- N-**None**
	- **WS-** Water Sprays
	- **WSA- Water Sprays with Wetting Agents**
	- CS-**Chemical Dust Suppressant (sprays, etc...)**
	- FE-**Full Enclosures**
	- PE-**Partial Enclosures**
	- MD- Minimization of material drop height
	- EM-Enclosure and evacuation to mechanical collector
	- EB-Enclosure and evacuation to baghouse

ES-**Enclosure and evacuation to scrubber**

OTH- Other - describe in footnote or attachment

6. **TYPE CONVEYOR OR TRANSFER POINT - Code as follows:**

Conveyors

- BC-**Belt Conveyor**
- VC-**Vibrating Conveyor**
- SC-**Screw Conveyor**
- DL-Drag-link conveyor
- BE-**Bucket Elevator**
- PS-**Pneumatic System**
- OTH- Other describe in footnote or attachment

Transfer Points

- $01 -$ **Conveyor to Conveyor**
- $02 -$ **Conveyor to Bucket Elevator**
- $03 -$ **Conveyor to Hopper or Bin**
- 04-**Bucket Elevator to Hopper or Bin**
- Pneumatic conveyor to bin $05-$
- -06 **Truck Dumping onto ground**
- $07 -$ **Truck Dumping into hopper**
- Loading trucks through stationary chute 08-
- Loading trucks through telescoping chute 09-
- $10 -$ **Loading Trucks by endloader**
- $11 -$ Railcar unloading-side or bottom dumping
- $12-$ Railcar unloading-rotary unloader
- $13-$ Railcar loading /unloading by pneumatic system
- $14 -$ Railcar loading through stationary source
- Railcar loading through telescopic chute $15 -$
- $16-$ Railcar loading by front end-loader
- $17 -$ Railcar loading by railcar
- Barge loading/unloading by clamshell $18 -$
- $19-$ Barge unloading - bucket ladder unloader
- Barge unloading from a fixed-height conveyor or stationary chute $20-$
- $21 -$ Barge loading - variable height conveyor or telescoping chute
- $22 -$ Other - describe in footnote or attachment
- If more than one material is handled by the listed conveyor or transfer point list $7₁$ each material and furnish the requested data in the table for each material.
- Describe type of unit such as hammermill, ball mill, double-deck (DD) screen, 8. double roll (DR) crusher, etc...
- Describe nominal size reduction, example +2"/ -% 9.

REDACTED - CLAIM OF CONFIDENTIALITY 12-08-08

Attachment L EMISSIONS UNIT DATA SHEET GENERAL

To be used for affected sources other than asphalt plants, foundries, incinerators, indirect heat exchangers, and quarries.

Identification Number (as assigned on *Equipment List Form*): Gasifier Feed and PDO Gasifier Units

1. Name or type and model of proposed affected source:

The Gasifier Feed and PDQ Gasifier Unit includes the feed dust bunkers (FDB1 and 2) the Lock Hopper System (Lock Hopper No. 1 (LH1) through No. 12 (LH12)), the Gasifier Feed Bin (GFB1 and 2), the Start Up Vessel (SUV1 and 2), and the PDQ Gasifier Unit (PDQ1 and 2). These units are specifically designed for each process and the final design has not been completed. There are two identical sets of units for this facility. The process information for them follows.

2. On a separate sheet(s), furnish a sketch(es) of this affected source. If a modification is to be made to this source, clearly indicated the change(s). Provide a narrative description of all features of the affected source which may affect the production of air pollutants.

See the Process Flow Diagram

3. Name(s) and maximum amount of proposed process material(s) charged per hour:

4. Name(s) and maximum amount of proposed material(s) produced per hour:

5. Give chemical reactions, if applicable, that will be involved in the generation of air pollutants:

Controlled combustion of coal.

The identification number which appears here must correspond to the air pollution control device identification number appearing on the List Form.

8. Projected amount of pollutants that would be emitted from this affected source if no control devices were used:

1. There are two emission points identified as B1/1 and B1/2 which have the same emissions estimate. The above information is for one point. B1/1 is for gasifier train No. 1 and B1/2 is for gasifier train No. 2. This is a continuous emissions point.

2. Start ups are estimated at 30 starts per year at one hour per each start up of the gasifiers. This is a worst case estimate for the process starts and includes both gasifiers. There are two emission points B2/1 and B2/2. The above emission is for one point.

3. There are two emission points identified as B3/1 and B3/2 which have the same emissions estimate. The above information is for one point. B3/1 is for gasifier train No. 1 and B3/2 is for gasifier train No. 2. This is an emergency

NOTE: (1) An Air Pollution Control Device Sheet must be completed for any air pollution device(s) used to control emissions from this affected source.

(2) Complete the Emission Points Data Sheet.

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Attachment L EMISSIONS UNIT DATA SHEET GENERAL

To be used for affected sources other than asphalt plants, foundries, incinerators, indirect heat exchangers, and quarries.

Identification Number (as assigned on Equipment List Form): CO Shift

The identification number which appears here must correspond to the air pollution control device \star identification number appearing on the List Form.

NOTE: (1) An Air Pollution Control Device Sheet must be completed for any air pollution device(s) used to control emissions from this affected source.

(2) Complete the Emission Points Data Sheet.

To be used for affected sources other than asphalt plants, foundries, incinerators, indirect heat exchangers, and quarries.

Identification Number (as assigned on Equipment List Form): CO2/H2S Removal

The identification number which appears here must correspond to the air pollution control device \star identification number appearing on the List Form.

To be used for affected sources other than asphalt plants, foundries, incinerators, indirect heat exchangers, and quarries.

Identification Number (as assigned on Equipment List Form): Sour Water Stripper

 \star The identification number which appears here must correspond to the air pollution control device identification number appearing on the List Form.

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To be used for affected sources other than asphalt plants, foundries, incinerators, indirect heat exchangers, and quarries.

Identification Number (as assigned on Equipment List Form): Mercury Removal

 \star The identification number which appears here must correspond to the air pollution control device identification number appearing on the List Form.

To be used for affected sources other than asphalt plants, foundries, incinerators, indirect heat exchangers, and quarries.

Identification Number (as assigned on Equipment List Form): Methanol Synthesis Unit

The identification number which appears here must correspond to the air pollution control device \star identification number appearing on the List Form.

To be used for affected sources other than asphalt plants, foundries, incinerators, indirect heat exchangers, and quarries.

Identification Number (as assigned on Equipment List Form): **Sulfur Recovery**

The Sulfur Recovery Unit includes the Claus Furnace (CF), Waste Heat Boiler (WHB), Condenser I through III (SRCD1 through SRCD3), Reheater I and II (RH1 and RH2), Claus Reactor I and II (CLR1 and CLR2). Feed / Effluent Heat Exchanger (FEHESR), Tailgas Heater (TGH), Hydrogenation Reactor (HR), Tail Gas Scrubber (TGS), Recycle Gas Compressor (RGC), Cooler No. 1 and No. 2 (CL1 and CL2), Sulfur Storage Tank (STK), and Line/Safety Vents No. 1 through No. 3 (SRLSF1 through SRLSF3). These units are specifically designed for each process unit and the final design has not been completed. There is one (1) Sulfur Recovery Unit.

2. On a separate sheet(s), furnish a sketch(es) of this affected source. If a modification is to be made to this source, clearly indicated the change(s). Provide a narrative description of all features of the affected source which may affect the production of air pollutants.

See the process flow diagram.

3. Name(s) and maximum amount of proposed process material(s) charged per hour:

4. Name(s) and maximum amount of proposed material(s) produced per hour:

5. Give chemical reactions, if applicable, that will be involved in the generation of air pollutants:

See Attachment $G - Process Description$

The identification number which appears here must correspond to the air pollution control device identification number appearing on the List Form.

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To be used for affected sources other than asphalt plants, foundries, incinerators, indirect heat exchangers, and quarries.

Identification Number (as assigned on Equipment List Form): PSA System

 \star The identification number which appears here must correspond to the air pollution control device identification number appearing on the List Form.

To be used for affected sources other than asphalt plants, foundries, incinerators, indirect heat exchangers, and quarries.

Identification Number (as assigned on *Equipment List Form*): $CO₂ Purification (CO₂P)$

1. Name or type and model of proposed affected source:

The CO₂Purification Unit (CO₂P) refines CO2 for the process and sends CO₂ back to the Coal Preparation units for blanket gas and vents to emission point C1. This process will be either CO2 Stripping or Catalytic Purification. This unit is specifically designed for each process and the final design has not been completed. There is one (1) CO₂ Purification unit.

2. On a separate sheet(s), furnish a sketch(es) of this affected source. If a modification is to be made to this source, clearly indicated the change(s). Provide a narrative description of all features of the affected source which may affect the production of air pollutants.

There is no preliminary design for this unit.

3. Name(s) and maximum amount of proposed process material(s) charged per hour:

4. Name(s) and maximum amount of proposed material(s) produced per hour:

5. Give chemical reactions, if applicable, that will be involved in the generation of air pollutants:

CO2 Purification

The identification number which appears here must correspond to the air pollution control device identification number appearing on the List Form.

To be used for affected sources other than asphalt plants, foundries, incinerators, indirect heat exchangers, and quarries.

Identification Number (as assigned on Equipment List Form): Air Separation Units

 \star The identification number which appears here must correspond to the air pollution control device identification number appearing on the List Form.

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REDACTED - CLAIM OF CONFIDENTIALITY 12-08-08

Attachment L EMISSIONS UNIT DATA SHEET GENERAL

To be used for affected sources other than asphalt plants, foundries, incinerators, indirect heat exchangers, and quarries.

Identification Number (as assigned on Equipment List Form): Methanol to Gasoline (MTG)

1. Name or type and model of proposed affected source:

The Methanol to Gasoline Unit includes the MTG Reaction Unit: Separation, Methanol Recovery (SMR); Methanol Vaporization/HP Steam Generation (MVS); DME Reactor and MTG Reactors (DMR); MTG Reactors Regeneration Systems (MTGRS); GasLiquid /Liquid Separationm (GLS); Deethanizer (DE); Stabilizer (STB); Methanol Recovery(MR); Absorber (AB); HGT Reactors (HGT); HGT Product Stripper(PS). These units are specifically designed for each process unit and the final design has not been completed. The units in each grouping above are detailed in Attachment I on Pages I11 thought I18. The units include Emission Points E1 through E5. E1 is the Regeneration Heater (SURGH), E2 is the Reactivation Heater (SURH), E3 is the HGT Heater (RCH), E4 is Process Waste Regeneration Gas Silencer (RGSL), and E5 is the flaring of the tail gas when both no process use available.

On a separate sheet(s), furnish a sketch(es) of this affected source. If a modification is to be $2₁$ made to this source, clearly indicated the change(s). Provide a narrative description of all features of the affected source which may affect the production of air pollutants.

See MGT process flow diagrams.

The identification number which appears here must correspond to the air pollution control device identification number appearing on the List Form.

8. Projected amount of pollutants that would be emitted from this affected source if no control devices were used:

This unit does not yent except during upset conditions and regeneration when venting will occur through the safety vent to the flare and to atmosphere. See Following Sheets for Emission Points E1 through E3.

NOTE: (1) An Air Pollution Control Device Sheet must be completed for any air pollution device(s) used to control emissions from this affected source.

Attachment L Emission Unit Data Sheet (INDIRECT HEAT EXCHANGER)

Control Device ID No. (must match List Form): NA

Equipment Information

Fuel Requirements Manufacturer Not Selected

Emissions Stream

41. Have you included the *air pollution rates* on the Emissions Points Data Summary Sheet? Yes

Attachment L Emission Unit Data Sheet (INDIRECT HEAT EXCHANGER)

Control Device ID No. (must match List Form): NA

Fuel Requirements Manufacturer Not Selected

Emissions Stream

Attachment L Emission Unit Data Sheet
(INDIRECT HEAT EXCHANGER)

Control Device ID No. (must match List Form): NA

Revision 03/2007

Fuel Requirements Manufacturer Not Selected

Attachment L Emission Unit Data Sheet
(INDIRECT HEAT EXCHANGER)

Control Device ID No. (must match List Form): NA

Revision 03/2007

Fuel Requirements Manufacturer Not Selected

Emissions Stream

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Attachment L EMISSIONS UNIT DATA SHEET GENERAL

To be used for affected sources other than asphalt plants, foundries, incinerators, indirect heat exchangers, and quarries.

Identification Number (as assigned on Equipment List Form): Cooling Tower (CT)

The Induced Draft Cooling Tower supplies cooling for all process water in the system. The unit is specifically designed for each process unit and the final design has not been completed. There is one (1) Cooling Tower.

2. On a separate sheet(s), furnish a sketch(es) of this affected source. If a modification is to be made to this source, clearly indicated the change(s). Provide a narrative description of all features of the affected source which may affect the production of air pollutants.

No process flow diagram for cooling tower.

3. Name(s) and maximum amount of proposed process material(s) charged per hour:

Hot water at an anticipated 308,167 gallons per minute.

4. Name(s) and maximum amount of proposed material(s) produced per hour:

Cooled water.

5. Give chemical reactions, if applicable, that will be involved in the generation of air pollutants:

Not Applicable

The identification number which appears here must correspond to the air pollution control device identification number appearing on the List Form.

NOTE: (1) An Air Pollution Control Device Sheet must be completed for any air pollution device(s) used to control emissions from this affected source.

(2) Complete the Emission Points Data Sheet.

EMISSIONS UNIT DATA SHEET STORAGE TANKS

Provide the following information for each new or modified bulk liquid storage tank as shown on the Equipment List Form and other parts of this application. A tank is considered modified if the material to be stored in the tank is different from the existing stored liquid.

IF USING US EPA'S TANKS **EMISSION** ESTIMATION PROGRAM (AVAILABLE AT **WWW.epa.gov/tnn/tanks.html**), APPLICANT MAY ATTACH THE SUMMARY SHEETS IN LIEU OF COMPLETING SECTIONS III, IV, & V OF THIS FORM. HOWEVER, SECTIONS I, II, AND VI OF THIS FORM MUST BE COMPLETED. USEPA'S AP-42, SECTION 7.1, "ORGANIC LIQUID STORAGE TANKS," MAY ALSO BE USED TO ESTIMATE VOC AND HAP EMISSIONS (http://www.epa.gov/tnn/chief/).

I. GENERAL INFORMATION (required)

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 1 EPA = EPA Emission Factor, MB = Material Balance, SS = Similar Source, ST = Similar Source Test, Throughput Data, $O =$ Other (specify)

Remember to attach emissions calculations, including TANKS Summary Sheets if applicable.

EMISSIONS UNIT DATA SHEET STORAGE TANKS

Provide the following information for each new or modified bulk liquid storage tank as shown on the Equipment List Form and other parts of this application. A tank is considered modified if the material to be stored in the tank is different from the existing stored liquid.

IF USING US EPA's TANKS EMISSION ESTIMATION PROGRAM (AVAILABLE AT **WWW.epa.gov/tnn/tanks.html)**, APPLICANT MAY ATTACH THE SUMMARY SHEETS IN LIEU OF COMPLETING SECTIONS III, IV, & V OF THIS FORM. HOWEVER, SECTIONS I, II, AND VI OF THIS FORM MUST BE COMPLETED. USEPA'S AP-42, SECTION 7.1, "ORGANIC LIQUID STORAGE TANKS," MAY ALSO BE USED TO ESTIMATE VOC AND HAP EMISSIONS (http://www.epa.gov/tnn/chief/).

I. GENERAL INFORMATION (required)

¹ EPA = EPA Emission Factor, MB = Material Balance, SS = Similar Source, ST = Similar Source Test, Throughput Data, O = Other (specify)

Remember to attach emissions calculations, including TANKS Summary Sheets if applicable.

Revision 03/2007

Attachment L EMISSIONS UNIT DATA SHEET BULK LIQUID TRANSFER OPERATIONS

Furnish the following information for each new or modified bulk liquid transfer area or loading rack, as shown on the Equipment List Form and other parts of this application. This form is to be used for bulk liquid transfer operations such as to and from drums, marine vessels, rail tank cars, and tank trucks.

(1) Total of Railcar and Tank Truck Loading
(2) Vapor balance system with pressure, no anticipated emissions.

9. Proposed Monitoring, Recordkeeping, Reporting, and Testing

Please propose monitoring, recordkeeping, and reporting in order to demonstrate compliance with the proposed operating parameters. Please propose testing in order to demonstrate compliance with the proposed emissions limits.

OPERATION/AIR POLLUTION CONTROL DEVICE.

RECORDKEEPING. PLEASE DESCRIBE THE PROPOSED RECORDKEEPING THAT WILL ACCOMPANY THE MONITORING.

REPORTING. PLEASE DESCRIBE THE PROPOSED FREQUENCY OF REPORTING OF THE RECORDKEEPING.

TESTING. PLEASE DESCRIBE ANY PROPOSED EMISSIONS TESTING FOR THIS PROCESS EQUIPMENT/AIR POLLUTION CONTROL DEVICE.

10. Describe all operating ranges and maintenance procedures required by Manufacturer to maintain warranty

This is a loading rack system which has not been selected. Manufacturer's operating ranges and maintenance procedures will be followed as recommended.

L86 of L100

Attachment L EMISSIONS UNIT DATA SHEET BULK LIQUID TRANSFER OPERATIONS

Furnish the following information for each new or modified bulk liquid transfer area or loading rack, as shown on the *Equipment List Form* and other parts of this application. This form is to be used for bulk liquid transfer operations such as to and from drums, marine vessels, rail tank cars, and tank trucks.

WVDEP-OAQ Revision 04-2007

EMISSIONS UNIT DATA SHEET STORAGE TANKS

Provide the following information for each new or modified bulk liquid storage tank as shown on the Equipment List Form and other parts of this application. A tank is considered modified if the material to be stored in the tank is different from the existing stored liquid.

IF USING US EPA'S TANKS EMISSION ESTIMATION PROGRAM (AVAILABLE AT **WWW.epa.gov/tnn/tanks.html)**, APPLICANT MAY ATTACH THE SUMMARY SHEETS IN LIEU OF COMPLETING SECTIONS III, IV, & V OF THIS FORM. HOWEVER, SECTIONS I, II, AND VI OF THIS FORM MUST BE COMPLETED. USEPA'S AP-42, SECTION 7.1, "ORGANIC LIQUID STORAGE TANKS," MAY ALSO BE USED TO ESTIMATE VOC AND HAP EMISSIONS (http://www.epa.gov/tnn/chief/).

I. GENERAL INFORMATION (required)

 1 EPA = EPA Emission Factor, MB = Material Balance, SS = Similar Source, ST = Similar Source Test, Throughput Data, O = Other (specify)

Remember to attach emissions calculations, including TANKS Summary Sheets if applicable.

EMISSIONS UNIT DATA SHEET STORAGE TANKS

Provide the following information for each new or modified bulk liquid storage tank as shown on the Equipment List Form and other parts of this application. A tank is considered modified if the material to be stored in the tank is different from the existing stored liquid.

IF USING US EPA'S TANKS EMISSION ESTIMATION PROGRAM (AVAILABLE AT www.epa.gov/tnn/tanks.html), APPLICANT MAY ATTACH THE SUMMARY SHEETS IN LIEU OF COMPLETING SECTIONS III, IV, & V OF THIS FORM: HOWEVER, SECTIONS I, II, AND VI OF THIS FORM MUST BE COMPLETED. US EPA'S AP-42, SECTION 7.1, "ORGANIC LIQUID STORAGE TANKS," MAY ALSO BE USED TO ESTIMATE VOC AND HAP EMISSIONS (http://www.epa.gov/tnn/chief/)

I. GENERAL INFORMATION (required)

Revision 12/01/2000

¹ EPA = EPA Emission Factor, MB = Material Balance, SS = Similar Source, ST = Similar Source Test, Throughput Data, O = Other (specify)

□ Remember to attach emissions calculations, including TANKS Summary Sheets if applicable.

FUGITIVE EMISSIONS FROM PAVED HAULROADS

INDUSTRIAL PAVED HAULROADS (including all equipment traffic involved in process, haul trucks, endloaders, etc.)

Surface material silt content (%)
Surface dust loading (lb/mile) $L =$

s in Section N

Source: AP-42 Fifth Edition - 11.2.6 Industrial Paved Roads

E = 0.077 x | x (4 ÷ n) x (s ÷ 10) x (L ÷ 1000) x (W ÷ 3)^{0.7} =

Ib/Vehicle Mile Traveled (VMT)

Where:

For lb/hr: $[1b + VMT] \times [VMT + trip] \times [Trips + Hour] =$ lb/hr

For TPY: $[lb + VMT] \times [VMT + trip] \times [Trips + Hour] \times [Top + 2000 lb] =$ Tons/year

ATTACHMENT M

AIR POLLUTION CONTROL DEVICE(S)

Attachment M Air Pollution Control Device Sheet (FLARE SYSTEM)

Control Device ID No. (must match Emission Units Table): FL

Characteristics of the Waste Gas Stream to be Burned

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Attachment M Air Pollution Control Device Sheet (CONDENSER SYSTEM)

Control Device ID No. (must match Emission Units Table): LR1 and LR2

Equipment Information and Filter Characteristics

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Control Device ID No. (must match Emission Units Table): BH1 through BH5 (For Roller Mills with Heating) Baghouse design/selection will be made to meet the controlled emissions requirements. The information contained within this form is yet to be determined.

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Control Device ID No. (must match Emission Units Table): **BH6** and BH13 (For Feed Dust Bunkers) Baghouse design/selection will be made to meet the controlled emissions requirements. The information contained within this form is yet to be determined.

22. Type of Pollutant(s) to be collected (if particulate give specific type): 23. Is there any SO₃ in the emission stream? \Box No \Box Yes $SO₃$ content: ppmv 24. Emission rate of pollutant (specify) into and out of collector at maximum design operating conditions: **OUT** IN **Pollutant** lb/hr grains/acf lb/hr grains/acf **Particle Size Distribution at Inlet** 25. Complete the table: **Fraction Efficiency of Collector** to Collector **Particulate Size Range (microns) Weight % for Size Range** Weight % for Size Range $0 - 2$ $2 - 4$ $4 - 6$ $6 - 8$ $8 - 10$ $10 - 12$ $12 - 16$ $16 - 20$ $20 - 30$ $30 - 40$ $40 - 50$ $50 - 60$ $60 - 70$ $70 - 80$ $80 - 90$ $90 - 100$ >100

Control Device ID No. (must match Emission Units Table): BH7-12 and BH14-19 (For Lock Hoppers) Baghouse design/selection will be made to meet the controlled emissions requirements. The information contained within this form is yet to be determined.

Control Device ID No. (must match Emission Units Table): VF1-10 (For Feed Bunkers) Baghouse design/selection will be made to meet the controlled emissions requirements. The information contained within this form is yet to be determined.

Control Device ID No. (must match Emission Units Table): VF11-12 and VF13-14 (For Feed Bins and Startup Vessels) Baghouse design/selection will be made to meet the controlled emissions requirements. The information contained within this form is yet to be determined.

Attachment M Air Pollution Control Device Sheet (BAGHOUSE)

Control Device ID No. (must match Emission Units Table): BHCS1-5 and BHLS1-2 (Stockpiles) Baghouse design/selection will be made to meet the controlled emissions requirements. The information contained within this form is yet to be determined.

Revision 03/15/2007

 $\label{eq:2.1} \begin{split} \mathcal{L}_{\text{max}}(\mathbf{r}) & = \frac{1}{2} \sum_{i=1}^{N} \mathcal{L}_{\text{max}}(\mathbf{r}) \\ & = \frac{1}{2} \sum_{i=1}^{N} \mathcal{L}_{\text{max}}(\mathbf{r}) \mathcal{L}_{\text{max}}(\mathbf{r}) \\ & = \frac{1}{2} \sum_{i=1}^{N} \mathcal{L}_{\text{max}}(\mathbf{r}) \mathcal{L}_{\text{max}}(\mathbf{r}) \end{split}$

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ATTACHMENT N

SUPPORTING EMISSIONS CALCULATIONS

(1) SO2 measurement was calculated from H2S emissions as an SO2 equivalent.
(2) Maximum hourly emissions from source (leaks, etc) cannot be quantified.

Potesta & Associates, Inc. Project No. 0101-08-0324

> 16.69 5.92

By: PEW Checked By: CCS Date: November 17, 2008 - Revised May 2009 Date: December 3, 2008 - Revised May 2009

 $\frac{44.23}{19.51}$

 $\frac{189.37}{83.94}$

 $\frac{4.12}{1.47}$

Total TSP =
Total PM10 =

Coal Summary of Emissions

Potesta & Associates, Inc. Project No. 0101-08-0234

Batch or Continuous Drops

AP-42 Section 11.2.3 (9/88) emission factor equation: $e = k * 0.0032 * [(U/5)^{1.3} / (M/2)^{1.4}]$ lb/ton

Defining transfer point empirical expression variables, where:

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NOTES:

1. Emissions are calculated for worst case scenario. If a zero has been entered, the transfer point is not part of the worst case scenario.

2. PM10=PM/2.1.

Crusher

1. Secondary crushing factor 0.06 lb/ton from DAQ General Permit G10-C.

2. PM10=PM/2.1.

3. Control efficiency from DAQ General Permit G10-C Refernce Document.

Potesta & Associates, Inc. Project No. 0101-08-0234

By: PEW

Date: November 17, 2008 - Revised May 2009

Checked By: CCS Date: December 3, 2008 - Revised May 2009

Vehicular Activity
Paved Haulroads

E = [k * (sL/2)^0.65 * (W/3)^1.5 - C] * (1 - (P/4*N) = lb / Vehicle Mile Traveled (VMT) 1. AP42, 13.2.1.

Potesta & Associates, Inc. Project No. 0101-08-0234

Stockpile

Stockpile is enclosed in a building and the building is vented through baghouses.

Round to $=$

Source/Emission Point Emissions Uncontrolled(2) Controlled (Ib/hr) (Ib/hr) (tpy) (tpy) Baghouse No. CS1 1.20 5.20 0.06 0.26 Baghouse No. CS2 1.20 5.20 0.06 0.26 1.20 5.20 0.06 Baghouse No. CS3 0.26 Baghouse No. CS4 1.20 5.20 0.06 0.26 PM 4.80 20.80 0.24 1.04 **PM10** 4.80 20.80 0.24 1.04

 $\overline{2}$

1. PM (TSP) = PM10

2. Back calculated and assumes indicated control efficiency.

Potesta & Associates, Inc. Project No. 0101-08-0324

By: PEW

Date: November 17, 2008 - Revised May 2009

Checked By: CCS Date: December 3, 2008 - Revised May 2009

Limestone Summary of Emissions

Potesta & Associates, Inc. Project No. 0101-08-0324

Batch or Continuous Drops

AP-42 Section 11.2.3 (9/88) emission factor equation: $e = k * 0.0032 * [(U/5)^{1.3} / (M/2)^{1.4}]$ ib/ton

 $e =$

Defining transfer point empirical expression variables, where:

Calculating transfer point emission factor using above equation:

0.0097 lb/ton

NOTES:

1. Emissions are calculated for worst case scenario. If a zero has been entered, the transfer point is not part of the worst case scenario.

2. Actual feed rate to the plant system is 19 tons per hour. Estimated filling rate of bin is 100 tons per hour.

3. PM10=PM/2.1.

Crusher

1. Tertiary crushing factos (PM 0.0054 lb/ton and PM10 0.0024 lb/ton) from AP-42, Section 11.19.2, Table 11.19.2-2.

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 $\frac{(typ)}{0.10}$
0.10

By: PEW

Date: November 17, 2008 - Revised May 2009

Checked By: CCS Date: December 3, 2008 - Revised May 2009

 (lb/hr)

 $\frac{0.14}{0.14}$

 $\frac{(%)}{85}$
 $Total$

WT/WC

Vehicular Activity Paved Haulroads

 $\overline{\text{PR}}$

 (lb/hr)

 $\frac{0.94}{0.94}$

(Ib/VMT)

 $\frac{1.71}{\text{Total}}$

 $(\frac{(\text{tpy})}{0.65})$

Emission Factors⁽¹⁾

 $\overline{5}$

6,935

 0.11

E = [k * (sL/2)^0.65 * (W/3)^1.5 - C] * (1 - (P/4*N) = lb / Vehicle Mile Traveled (VMT) 1. AP42, 13.2.1.

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Stockpile

Stockpile is enclosed in a building and the building is vented through baghouses.

1. PM (TSP) = $PM10$

2. Back calculated and assumes indicated control efficiency.

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Ash Summary of Emissions

Ash Facilities Particulate Matter PTE

Potesta & Associates, Inc. Project No. 0101-08-0324

Batch or Continuous Drops: Ash/Aggregate

AP-42 Section 11.2.3 (9/88) emission factor equation: $e = k * 0.0032 * [(U/5)^{1.3} / (M/2)^{1.4}]$ ib/ton

Defining transfer point empirical expression variables, where:

Througput

6,250

61.320

 0.21

168

KG/H

TPH $TPD =$

 0.50

Calculatin equation:

 $e =$ lb/ton 0.0097

Batch or Continuous Drops: Filter Cake (Estimated at 10% of Ash/Aggregate)

 $U =$

 $M =$

AP-42 Section 11.2.3 (9/88) emission factor equation: $e = k * 0.0032 * [(U/5)^{1.3} / (M/2)^{1.4}]$ lb/ton

Defining transfer point empirical expression variables, where:

 $k =$ 0.74 dimensionless

mph (mean wind speed in WV) $\overline{7}$

 1.0 % Moisture Content

PM10 and TSP assumed equal in this case.

 0.60

 1.04

 $PM10 =$

Ash Facilities Particulate Matter PTE

Potesta & Associates, Inc. Project No. 0101-08-0324

By: PEW

Date: November 17, 2008 - Revised May 2009

Checked By: CCS Date: December 3, 2008 - Revised May 2009

 $\frac{10,000}{0.25}$

 $rac{85}{36}$

WT/WC

Vehicular Activity
Paved Haulroads

 PR

 $\frac{1.71}{\text{Total}}$

9

27,740

 0.11

E = [k * (sL/2)^0.65 * (W/3)^1.5 - C] * (1 - (P/4*N) = lb / Vehicle Mile Traveled (VMT) 1. AP42, 13.2.1.

Transgas Sulfur and Misc Trucking Particulate Matter PTE

Potesta & Associates, Inc. Project No. 0101-08-0324

By: PEW Checked By: CCS Date: November 17, 2008 - Revised May 2009 Date: December 3, 2008 - Revised May 2009

Sulfur and Misc Trucking Summary of Emissions

Ash Facilities Particulate Matter PTE

Potesta & Associates, Inc. Project No. 0101-08-0324

By: PEW

Date: November 17, 2008 - Revised May 2009

Vehicular Activity

Paved Haulroads

Road
2,400
0.45

 $226,271$
 24
9,428

 $\begin{array}{c} 26 \\ 24 \\ 2 \end{array}$

Length In (ft) =
Length (mi) =
Total Hauled (tpy) =
Load Weight (tons) =
Trucks Per Year =

Total Hauled (tph) =
Load Weight (tons) =
Trucks Per Hour =

Cooling Tower Particulate Matter PTE Potesta & Associates, Inc. Project No. 0101-08-0324

Cooling Tower Emissions

Liquid drift is assumed to be PM10 and will be added to PM and PM10 totals. Emissions estimated by AP-42 Section 13.4, Wet Cooling Towers.

> Water $Flow =$ 18,490,000 gallon per hour (Uhde 70,000 m3/hr) 308,167 gallons per minute Circulating Water TDS = 5,000 ppm (estimated) 0.001 % Drift (estimated) Design Drift Rate = Operating Hours = 8,760 hrs/year Liquid Drift $=$ 7.71 lbs/hr 33.77 tpy

> > Rounding to $= 2$

Methanol Tank and Fugitives VOC PTE

Potesta & Associates, Inc. Project No. 0101-08-0324

By PEW

Date: November 17, 2008 - Revised May 2009

Checked By: CCS Date: December 3, 2008 - Revised May 2009

Methanol System (Vapor Sources)

 $ib/kg = 2.2046$

1. Table 5-1 and 5-2 (Reduction by LDAR) of Protocol for Equipment Leak Emissions Estimate (EPA-453/R-95-017) dated November 1995.
Valves
Connectors 38 % Reduction Table 5-2
Connectors 93 % Reduction Table 5-2

Methanol Tank

1. Based on

8,760 hours per year.

Gasoline Fugitives, Storage, and Loading Racks VOC PTE

Potesta & Associates, Inc.
Project No. 0101-08-0324

Date: December 3, 2008 - Revised May 2009

Ib/y

14.106

 $\frac{102}{12}$

 $\mathbf 6$

 12

12
60

Checked By: CCS

By: PEW

Date: November 17, 2008 - Revised May 2009

Gasoline Emissions (Fugitive, Tanks, and Loading Racks)

.
Samant Laok Carinala

Gasoline System Fugitives

Total Uncontrolled HAPS (Protated on TANKS Output) = 0.593 = 1 2.618
1. Table 5-1 and 5-2 (Reduction by LDAR) of Protocol for Equipment Leak Emissions Estimate (EPA-453/R-95-017) dated November 1995.
1. Valves

Gasoline Tanks

One Tank Calculated
Total Emissions
Ib/hr⁽¹⁾ tpy TANKS 4.0 Output (for one tank) Three Tanks Losses **Ib/h** \overline{w} Deck Fitting Deck Sean Total Components

Gasoline (RVP 13)

Unidentified

Components

Berzene Rim Seal Loss
1,506.60 Withdrawl Loss $\frac{\text{Loss}}{3,022.61}$ $rac{\text{Loss}}{0}$ Emissions
4,702.69 0.53684 2.351 1.61 7.053 $\begin{array}{r} 0.52383 \\ 0.00261 \\ \hline 0.00079 \\ \hline 0.00383 \\ \hline 0.00044 \\ \hline 0.00205 \end{array}$ $\begin{array}{r}\n 129.34 \\
 \hline\n 3.12 \\
 \hline\n 6.94 \\
 \hline\n 12.14 \\
 \hline\n 2.43 \\
 \hline\n 12.14\n \end{array}$ 4,588.73
22.85
6.94
33.58
3.82
17.93 $\begin{array}{r} 2.294 \\ 0.011 \\ \hline 0.003 \\ 0.017 \\ \hline 0.002 \\ 0.009 \\ \hline \end{array}$ $\begin{array}{r} \text{NA} \\ \hline 0.0078 \\ \hline 0.0024 \\ \hline 0.0115 \\ \hline 0.0013 \\ \hline 0.0062 \\ \hline \end{array}$ $\frac{1,483,37}{6.56}$ 2,976.02
13.16 \mathbf{N}^{μ} $\begin{array}{r} \hline 0.033 \\ \hline 0.009 \\ \hline 0.051 \\ \hline 0.006 \\ \hline 0.027 \end{array}$ $\begin{array}{r} \underline{.3.1} \\ \underline{0} \\ 14.31 \\ \underline{0.93} \\ 3.86 \end{array}$ Seizene
Isooctane
Toluene
Ethylbenzen
Xylene (-m) $\frac{0}{7.13}$
0.46
1.93 Isopropyl benzene

1,2,4-

Trimethylbenzene

Cyclohexane

Hexane (-n) 0.07 0.87 $0,15$ 1.09 0.00012 0.001 0.0004 0.003 $\mathfrak o$

ń

 $\frac{0.31}{1.82}$

1. Based on 8,760 hours per year.

 $\frac{0.15}{0.91}$
6.01

Gasoline Loading Racks (LR1 and LR2)

VOC losses from loading gasoline to truck or railcar with the control of the vapor recovery system. AP-42, Section 5.2,

 $\frac{4.34}{0.42}$
1.73

 $\frac{4.8}{3.15}$
19.79

tounding to

 $\frac{0.00055}{0.00036}$

 $\frac{0.002}{0.002}$

 $\frac{0.0017}{0.0011}$
0.0068

 $\frac{0.006}{0.006}$

 $\rm Gasoline\,Fugitives, Storage,$ and $\rm Loading\,Racks$ VOC PTE

Potesta & Associates, Inc.
Project No. 0101-08-0324

Checked By: CCS
Date: December 3, 2008 - Revised May 2009

By: PEW
Date: November 17, 2008 - Revised May 2009 Gasoline Emissions (Fugitive, Tanks, and Loading Racks)

Summary of Gasoline Fugitives and Tanks Emissions

Sulfur Tank Emissions H₂S PTE

Potesta & Associates, Inc. Project No. 0101-08-0324

Checked By: CCS Date: December 3, 2008 - Revised May 2009

Sulfur Tank Filling

By: PEW

5 % H2S is lost to vapor Assume = 10 ppm H2S in Sulfur Throughput = 2669.53 kg/hr 5886.314 lb/hr 51,564,110.64 lbs/yr H2S Emisions = 0.003 lb/hr 51,564,200.00 say (lbs/yr) 8,760 hrs/yr 141,271.78 lbs/day 26.280 lbs/yr 0.0130 tpy

Sulfur Vehicle Filling (Assume same loss as filling tank)

Date: November 17, 2008 - Revised May 2009

Mass Balance Conversion to English Units for EUDS

Potesta & Associates, Inc. Project No. 0101-08-0324

REDACTED - CLAIM OF CONFIDENTIALITY 12-08-08

By: PEW Checked By: CCS Date: December 3, 2008 - Revised May 2009 Date: November 17, 2008 - Revised May 2009

This is a mass balance conversion of the Emission Unit Data Sheets in Section L

Mass Balance Conversion to English Units for EUDS

Potesta & Associates, Inc. Project No. 0101-08-0324

REDACTED - CLAIM OF CONFIDENTIAL ITV 12-08-08

 $\rm N21$ of $\rm N22$

Mass Balance Conversion to English Units for EUDS

Potesta & Associates, Inc. Project No. 0101-08-0324

Quantity 1

 $\begin{array}{c}\n 32 \\
 1 \\
 1\n \end{array}$

TANKS 4.0.9d Emissions Report - Detail Format Tank Indentification and Physical Characteristics

Automatic Gauge Float Well/Unbolted Cover, Ungasketed
Roof Leg or Hanger Well/Aquistable
Sample Pipe or Well/Aquistable
Sample Pipe or Well (24-in. Diam.)/Slit Fabric Seal 10% Open
Vacuum Breaker (10-in. Diam.)/Weighted Me

Meterological Data used in Emissions Calculations: Charleston, West Virginia (Avg Atmospheric Pressure = 14.25 psia)

file://C:\Program Files\Tanks409d\summarydisplay.htm

TANKS 4.0.9d Emissions Report - Detail Format Liquid Contents of Storage Tank

TK-1 Through TK-3 - Internal Floating Roof Tank Charleston, West Virginia

TANKS 4.0.9d Emissions Report - Detail Format Detail Calculations (AP-42)

TK-1 Through TK-3 - Internal Floating Roof Tank Charleston, West Virginia

TANKS 4.0.9d Emissions Report - Detail Format Individual Tank Emission Totals

Emissions Report for: Annual

TK-1 Through TK-3 - Internal Floating Roof Tank Charleston, West Virginia

 $32 + 32 + 1$

TANKS 4.0.9d Emissions Report - Detail Format Tank Indentification and Physical Characteristics

Acor Leg or Hanger Well/Adjustable
Sample Pipe or Well (24-in. Diam.)/Silt Fabric Seal 10% Open
Vacuum Breaker (10-in. Diam.)/Weighted Mech. Actuation, Gask.

Meterological Data used in Emissions Calculations: Charleston, West Virginia (Avg Atmospheric Pressure = 14.25 psia)

file://C:\Program Files\Tanks409d\summarydisplay.htm

TANKS 4.0.9d Emissions Report - Detail Format Liquid Contents of Storage Tank

IFR Methanol Tank - Internal Floating Roof Tank Charleston, West Virginia

TANKS 4.0 Report

Page 3 of 5

 \cdots

TANKS 4.0.9d Emissions Report - Detail Format Detail Calculations (AP-42)

IFR Methanol Tank - Internal Floating Roof Tank
Charleston, West Virginia

file://C:\Program Files\Tanks409d\summarydisplay.htm

TANKS 4.0.9d Emissions Report - Detail Format Individual Tank Emission Totals

Emissions Report for: Annual

IFR Methanol Tank - Internal Floating Roof Tank Charleston, West Virginia

Transgas Development Systems

Task Order 1 under

Services Agreement for Construction Permitting Support

Response on DEP Questions

Prepared by

Uhde

 $\ldots/2$

Table of Content

Introduction 1

1.1 General

Additional to the Information contained in the permit application, the West Virginia DEP requested more detailed data for the valuation of hazardous air pollutions emissions and explanatory discussion on the information basis for the calculation of the emissions of the CTL plant (questions 2 and 5 of the letter from DEP to TGDS, dated January 8th 2009).

Further, to visualize the emissions points, a block flow diagram showing the emissions points separately as well as all lines (including safety valves) going to flare have been requested (question 3 of mentioned letter).

Uhde has been commissioned by TGDS with task order No. 1 under the Services Agreement for Construction Services Support to supply response on above mentioned questions.

1.2 **Basis for Emission Calculation**

All emission calculations have to the maximum extent possible been based on commercial operating experience for the technologies applied in the TGDS CTL plant, especially the Puertollano IGCC applying the PRENFLO gasification process, which has more than 10 years operational experience up to date as well as the New Zealand Synfuel Plant applying the MTG process, which was operated for more than 10 years from the mid 1980's up to the mid 1990's.

All base values for the emissions and the calculation methods applied have been developed specifically for the TGDS CTL plant utilizing proprietary in-house modeling and calculation tools based on project specific design basis data, such as coal and fuel gas specifications, and applying the experience and know-how from operating data and proven start-up and operating procedures from the mentioned commercial plants.

 $... / 3$

Uhde

 $... / 4$

$\mathbf 2$

Discussion on Hazardous Air Pollutants

General

- The discussion on Hazardous Air Pollutants (HAPS) hereafter reflects the current status of engineering work done for the TGDS CTL facility
- The list HAPS from the EPA website (http://www.epa.gov/ttn/atw/188polls.html) was compared with the material balance of the individual process Units.
- Concentration of the HAPS are taken from the material balance. If not available from the material balance, values are estimated according experience from existing plants (e.g. Puertollano).
- In addition to the identification and, where possible, quantification of HAPS in the CTL facility, the mechanism of formation as well as possible emission of HAPS and emission control is shortly described.
- References for emission points given relate to the Block Flow Diagram under chapter 4 hereof.
- Flare emission for emergency cases not considered in maximum emission calculations. Startup and continuous emissions basis explained in chapter 3.
- For all HAPS (except HCl) a flare destruction efficiency from 98% is assumed, e.g. under chapter 2.1.4 during start-up 98% of the COS will be converted to SOx in the flame, etc.
- For leak HAPS emission refer to section 3.16

2.1 Carbonyl sulfide (COS)

$2.1.1$ Place of formation / use

Gasification Reactor

In the Gasification Reactor a part of the sulfur, coming with the coal into the reactor, reacts to COS.

$2.1.2$ **Mechanism of formation**

$S + CO \rightarrow COS$

$2.1.3$ **Balance**

$2.1.3.1$ **Production Rate**

$2.1.4$ Emissions to atmosphere and emission control

There will be no continuous emission:

- Process closed to atmosphere
- Relief to flare during start-up (B2/1 and B2/2) Pressure relief to flare (emergency relief; c1, c2); conversion of COS to SOx in flare.

Maximum COS emission during start-up (relief periods and flow rates see chapter 3.5):

- Flow rate during start-up: 224.1 kg/h (one gasifier)
- Destruction rate: 98%
- Hourly emission: $224.1 \text{ kg/h} \times 0.02 = 4.482 \text{ kg/h}$ (9.9 lb/hr)
- Yearly emission: 60 starts per yr x 1 hr/start x 9.9 lb/hr / 2000 lb/ton = 0.3 tons/yr.
- During normal operation a part of COS is hydrolysed on COS Shift section to H2S; remaining COS is converted in sulfur recovery (Claus) plant to pure sulfur and CO2

 $... / 5$

2.2° **Hydrogen Sulfide (H2S)**

$2.2.1$ Place of formation / use

Gasification Reactor

In the Gasification Reactor most of the sulfur, coming with the coal into the reactor, reacts to H2S.

$2.2.2$ **Mechanism of formation**

$S + H2 - > H2S$

$2.2.3$ **Balance**

$2.2.3.1$ **Production Rate**

$2.2.4$ Emissions to atmosphere and emission control

There will be no continuous emission

- Process closed to atmosphere
- Relief to flare during start-up (B2/1 and B2/2) Pressure relief to flare (emergency relief; c1, c2); $\ddot{}$ conversion of H2S in SOx in flare
	- Maximum H2S emission during start-up (relief periods and flow rates see chapter 3.5): Flow rate during start-up: 1163.5 kg/h (one gasifier)
	- Destruction rate: 98%
	- Hourly emission: 1163.5 kg/h x 0.02 = 23.27 kg/h (51.3 lb/hr)
	- Yearly emission: 60 starts per yr x 1 hr/start x $\overline{5}1.3$ lb/hr / 2000 lb/ton = 0.7 tons/yr.
- During normal operation H2S is converted in sulfur recovery (Claus) plant to pure sulfur and water

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2.3 **Nickel Carbonyl**

$2.3.1$ Place of formation / use

Gasification Reactor

Stainless steel pipes

Under certain pressure and temperature conditions Nickel from stainless steel pipes (or SS internals from e.g. vessel) reacts with the CO of the syngas.

$2.3.2$ **Mechanism of formation**

Ni + 4 CO -> Ni(CO)4

$2.3.3$ **Balance**

$2.3.3.1$ **Production Rate**

Hold (order of magnitude: ppm only)

$2.3.4$ Emissions to atmosphere and emission control

There will be no continuous emission

- Process closed to atmosphere
	- Relief to flare during start-up (B2/1, B2/2) and pressure relief (emergency relief; c1) to flare will cause nickel emissions (destruction from nickel carbonyl to nickel and CO2 in the flame of the flare).
	- Nickel emission not quantifiable, but conservative estimated significantly lower than 0.1 ton/yr. Maximum Nickel emission (based on measurement in commercial gasification plant) during start-up (relief periods and flow rates see chapter 3.5):

Flowrate of the syngas (one gasifier): 560979 kg/h

Concentration of Nickel in syngas: max. 0.8 ppmwt (measured value in commercial gasification plant)

Hourly emission (one gasifier): 560979 kg/h x 0.8 ppm = 0.561 kg/h (1.237 lb/hr)

- Yearly emission: 60 starts per yr x 1 hr/start x 1.237 lb/hr / 2000 lb/ton = 0.037 tons/yr
- Nickel carbonyl is converted in sulfur recovery (Claus) plant to pure nickel and CO2. Nickel will be adsorbed by the sulfur catalyst

 $... / 7$

2.4 **Hydrogen Cyanide (HCN)**

$2.4.1$ Place of formation / use

Gasification Reactor

$2.4.2$ **Mechanism of formation**

 $N + C + H \rightarrow HCN$

$2.4.3$ **Balance**

$2.4.3.1$ **Production Rate**

$2.4.4$ Emissions to atmosphere and emission control

There will be no continuous emission

Process closed to atmosphere

- Relief to flare during start-up (B2/1, B2/2) and emergency pressure relief;(c1) will convert HCN \Box to NO_x
	- Maximum HCN emission during start-up (relief periods and flow rates see chapter 3.5): Flow rate during start-up: 51.6 kg/h (one gasifier)
	- Destruction rate: 98%

Hourly emission: 51.6 kg/h x $0.02 = 1.03$ kg/h (2.27 lb/hr)

- Yearly emission: 60 starts per yr x 1 hr/start x 2.27 lb/hr / 2000 lb/ton = 0.07 tons/yr
- During normal operation the main part of HCN is converted in CO-Shift to H2O, N2 and CO2
- HCN not converted in the CO-Shift is converted in sulfur recovery (Claus) plant to pure H2O, N₂ and C_{O2}

 $... / 8$

2.5 **Hydrochloric Acid (HCI)**

$2.5.1$ Place of formation / use

Gasification Reactor

$2.5.2$ **Mechanism of formation**

 $Cl2 + H2 - > 2 HCl$

$2.5.3$ **Balance**

$2.5.3.1$ **Production Rate**

Formed in the gasification reactor the concentration in syngas before quench: 0.0068 mol% (68 ppm) Flowrate 60.5 kg/h (1.67 kmol/h). After quench of the syngas maximum concentration 1 ppm

$2.5.4$ Emissions to atmosphere and emission control

There will be no continuous emission

- Process closed to atmosphere
	- \overline{a} Relief to flare during start-up (B2/1, B2/2 after quench) and emergency pressure relief (b3/1; $b3/2$).
	- Maximum HCI emission during start-up (relief periods and flow rates see chapter 3.5): Flowrate of the syngas (one gasifier): 28516.8 kmol/hr
	- Concentration of HCI after quench: 1 ppm
	- Hourly emission (one gasifier): 28516.8 kmol/hr x 1ppm x 36.45 kg/kmol = 1.04 kg/h (2.29 $lb/hr)$
	- Yearly emission: 60 starts per yr x 1 hr/start x 2.29 lb/hr / 2000 lb/ton = 0.07 tons/yr (no HCI destruction in flare)
	- During normal operation the quench of the PDQ gasifier and the scrubber wash out the gaseous HCI from the syngas, then HCI will be neutralized with caustic soda.

 $... / 9$

2.6 **Mercury**

$2.6.1$ Place of formation / use

Gasification Reactor (Component of the feed coal)

$2.6.2$ **Mechanism of formation**

Released from coal during gasification.

$2.6.3$ **Balance**

$2.6.3.1$ **Production Rate**

Depending on mercury in the feed coal (none specified in design coal).

$2.6.4$ Emissions to atmosphere and emission control

There will be no continuous emission

- Process closed to atmosphere
- Relief to flare during start-up (B2/1, B2/2 and C2) and pressure relief (emergency relief; c1) to flare will cause mercury emissions.
- Maximum Mercury emission during start-up (relief periods see chapter 3.5 and 3.7): Average Mercury concentration in typical hard coal: 0.1 ppm (maximum: 1 ppm) Feed stream of hard coal (one gasifier): 154934 kg/h Concentration of Mercury: 0.1 ppm (maximum: 1 ppm) Hourly emission: 154934 kg/h x 0.1 ppm = 0.155 kg/h (0.342 lb/hr) [Maximum: 1.55 kg/h (3.42

 lb/hr]

Yearly emission: (60 starts per yr x 1 hr/start + 4 starts per yr * 0.5 hr/start) x 0.342 lb/hr / 2000 lb/ton = 0.011 tons/yr (Maximum: 0.11 tons/yr)

During normal operation mercury is adsorbed from the syngas in a fixed bed adsorber behind the acid gas removal.

 $... / 10$

Trace Components of Coal 2.7

$2.7.1$ Place of formation / use

The ash from coal can typically contain several trace components listed as HAPS:

- Antimony $\,$
- \equiv Arsenic
- Beryllium $\,$
- Chromium \equiv
- Cobalt \equiv
- Lead \equiv
- Manganese
- Mercury
- Nickel
- Selenium \equiv

$2.7.2$ **Mechanism of formation**

During gasification of the coal, these components will be enclosed in the slag.

$2.7.3$ **Balance**

$2.7.3.1$ **Production Rate**

Depending on trace components in the feed coal (none specified in design coal).

$2.7.4$ Emissions to atmosphere and emission control

There will be no emission

- HAPS are enclosed from slag.
- Solid material will be kept wet, to avoid particle emissions \mathbb{Z}^+
- During pressure relief (emergency relief; b3/1, b3/2) to flare the slag will be held in the Knock- \Box Out Drum.

 $... / 11$

2.8 **Methanol**

$2.8.1$ Place of formation / use

Methanol Reactor Rectisol Gas cleaning

$2.8.2$ **Mechanism of formation**

CO + 2 H2 -> CH3OH

Methanol is used as solvent in Rectisol plant

2.8.3 **Balance**

Acc. vendor Information methanol concentration in Syngas downstream acid gas removal is 0.0036 mol% (0.05 wt%). Flow rate stream 5 (ex acid gas removal) @ 100% load: 130 kg/h.

$2.8.3.1$ **Production Rate**

Stream 6: \overline{a}

7671.00 kmol/h total with 93.38 mol% Methanol 7163 kmol/h (229 216 kg/h) pure Methanol

Circulation rate Rectisol: hold kg/h \overline{a}

$2.8.4$ Emissions to atmosphere and emission control

There will be no continuous emissions

- Process closed to atmosphere
- Relief to flare during start-up (C2) and pressure relief to flare (emergency relief c3, d1) $\ddot{}$
- Maximum Methanol emission during start-up (relief periods and flow rates see chapter 3.7): \overline{a} Flow rate during start-up: 130 kg/h
	- Destruction rate: 98%

Hourly emission: 130 kg/h x $0.02 = 2.6$ kg/h (5.73 lb/hr)

Yearly emission: 4 starts per yr x 0.5 hr/start x 5.73 lb/hr / 2000 lb/ton = 0.006 tons/yr).

- Pressure less vessels and tanks are equipped with appropriate emission control (e.g. inert gas blanketing)
- Methanol formed in methanol reactor will be converted to gasoline

 $... / 12$

2.9 **MTG Gasoline**

$2.9.1$ Place of formation

DME and MTG Reactor

Mechanism of formation $2.9.2$

$2.9.3$ **Balance**

$2.9.3.1$ **Production Rate**

Certain HAPS are formed during the MTG reaction process (Ex MTG Reactor) and are contained in the Raw MTG Gasoline. Some HAPS are degraded in the Heavy Gasoline Treatment (HGT) and show a lower concentration in the Final Gasoline product (Balance Stream 8).

1) Delisted

$2.9.4$ Emissions to atmosphere and emission control

There will be no emissions

- Process closed to atmosphere \overline{a}
- Emergency pressure relief to flare (emergency relief e1) \mathbf{r}
- Pressure less vessels and tanks are equipped with appropriate emission control (e.g vapor re- \overline{a} covery)
- Gasoline is the product of the plant and has to be handled with the usual care similar to crude oil based gasoline

Explanation on Emission Points $\mathbf{3}$

General:

- For more details of emission points A1, A2, B1, B2 and B3 please refer to Attachment 2.

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- Hydrogen supplied to the MTG plant is no direct source of emission (a higher or lower hydrogen supply from the PSA to the MTG respectively doesn't lead to change in the overall plant emission)
- PSA Tailgas is recirculated inside the CTL facility and therefore is no direct source for emissions
- MTG Tailgas during normal plant operation is recirculated inside the CTL facility and therefore is no direct source for emissions; only when both gasifiers are down at the same time (i.e. the coal to methanol plant is shutdown) and the MTG plant is continued to operate using methanol feed from the storage, MTG Tailgas is flared (Emission Point E5)
- Emission calculations have been based on available data and information according to current status of engineering work done for the TGDS CTL project
- Data not calculable at the time being has been estimated based on previous experience (backed up by commercial experience, where possible)
- Start-up emissions have been considered as far as possible
- Emissions by emergency relief have not been considered (refer to emergency relief points as identified in section 4)
- For VOC emission in flare during start-up of gasifier (Emission Point B2) only the methane concentration of the syngas is considered, as syngas contain no other hydrocarbons, AP-42 values are based on a mixture of propane and propylene and do not represent flaring of syngas.
- For VOC emission in flare during start-up of acid gas removal (Emission Point C2), additionally to the methane contained in syngas, methanol entrained in syngas from the Absorber has been considered (as per section 2.8.4).

3.1 **A1 Coal Preparation Vent Mill**

Continuous emission originates from hot gas generator as well as from coal; gas circulation (and venting rates) in the coal preparation is calculated based on Design Coal data from TGDS; Coal Preparation is a commercially proven design. During normal operation fired with hydrogen, during start-up fired with natural gas.

$-CO$

Hydrogen is used as heating medium for the drying step; CO comes only from coal devolatilization; specific CO from coal devolatilization has been estimated based on tests that Uhde has performed for other coal.

- NOx

Specific NOx acc. coal preparation vendor (100 ppmv)

- SOx

Only during cold start-up (Heating medium natural gas); Calculated acc. max. sulfur content of natural gas specified by TGDS

$-$ PM

Particle emissions from baghouse filter: industrial standard

- VOC

Value has been calculated based on specific VOC from tests that UHDE has performed for other coal.

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3.2 **B3 Dry Dust Feeding**

Release of overpressure of Feed Bin via pressure control valve - normally no flow to atmosphere (no emission). Gas release only during malfunction of pressure control of the Feed Bin (Gas composition like emission point A2).

3.3 **A2 Dry Dust Feeding**

Emission only during start-up; Flow rate is determined based on design for feeding cycle, which is derived from the design as commercially demonstrated in the PRENFLO plant in Puertollano, Spain (30 starts per year per gasifier; i.e. overall 60 gasifier start per year).

- CO

CO residual content in CO2 carrier gas is reduced to less than 1 ppm by CO2 purification step; This can be achieved by state-of-the-art catalytic purification or CO2 stripping:

- NOx

Not applicable (The CO2 contains no NOx)

- SOx

Residual sulfur content specified by AGR vendor (performance demonstrated in commercial plants); the number given is a conservative approach, as the CO2 purification step would further reduce sulfur content.

 $-_{PM}$

Particle emissions from baghouse filter: industrial standard

- VOC

Not applicable (The CO2 contains no VOCs)

3.4 **B1 Dry Dust Feeding**

In the coal feeding; CO2 (used as carrier gas) is vented during the depressurization of the coal lock hoppers; vent rates are determined by the design of the feeding cycle, which is derived from design as commercially proven in the PRENFLO plant in Puertollano; Calculation is based on 8760 hours operating time per year at full capacity, which is a very conservative approach (real operating hours will be less).

 $-CO$

CO residual content in CO2 carrier gas (some of which is vented during feeding process) is reduced to less than 1 ppm by CO2 purification step; This can be achieved by state-of-the-art catalytic purification or CO2 stripping;

NO_x

Not applicable (The CO2 contains no NOx)

 $-SOx$

Residual sulfur content specified by AGR vendor (performance demonstrated in commercial plant); the number given is a conservative approach, as the CO2 purification step would further reduce sulfur content.

- PM

Particle emissions from baghouse filter: industrial standard

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 $-_{VOC}$ Not applicable (The CO2 contains no VOCs)

3.5 **B2 Raw Syngas**

Emissions are from flaring of raw syngas during start-up of the gasifier; Flaring rate and time is calculated based on commercially proven start-up procedure in PRENFLO plant in Puertollano; Number of start-ups per year has been estimated from commercial experience in Puertollano and other gasification plants, modified by adjustments made in current design such as integration of redundancies, inclusion of lessons learnt etc (30 start per gasifier per year; per start 1hr; max. 620 tons/hr syngas per gasifier).

- CO

Specific CO concentration in flare offgas: assumption acc. flare vendor information (1000 ppmv)

- NOx

Specific NOx concentration in flare offgas: assumption acc. flare vendor information (250ppmv)

- SOx

Calculated acc. conservative assumption of max. sulphur (H2S) content of syngas based on coal sulfur content; to reduce SOx emission, low sulfur start-up coal has to used during start-ups rather than design coal (e.g. PRB coal). This is readily available from the market. A typical PRB coal composition has been used for calculation of the sulfur concentration;

- PM

Not applicable (smokeless flare acc. table 13.5-1 of AP42)

- VOC

Not applicable (only small methane content in syngas causes some HC emissions)

3.6 C1 CO2 Offgas

Through C1 emissions originating from the CO2 removed in the AGR as well as from the Regeneration Off-gas of the MTG plant are release, as further explained in the following.

CO₂ from AGR

Continuous emission of CO2 produced during the CO Shift of syngas for adjusting the required H2 : CO ratio for downstream MeOH synthesis, which is selectively removed in the AGR (total flow rate: 382.4 tons/hr);

 $-CO$

CO residual content is reduced to less than 1 ppmv by CO2 purification step; This can be achieved by state-of-the-art catalytic purification or CO2 stripping; Flow rate offgas: 7883.5 kmol/h

CO concentration offgas: 1 ppmv CO flow rate: 7883.5 kmol/h x 1 ppmv / 10^{6} = 0.00788 kmol/h Molecular weight CO: 28 kg/kmol CO flow rate: 0.00788 kmol/h x 28 kg/kmol = 0.22 kg/h Conversion factor: 1 lb = 0.454 kg Hourly CO flow rate: 0.22 kg/h / 0.454 kg/lb = 0.49 lb/h Yearly CO flow rate: 0.49 lb/h x 8000 h / 2000 lb/ton = 1.96 tons/yr

 $-$ NO_x Not applicable (The CO2 offgas contains no NOx)

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$-SOx$

Residual sulfur content specified by AGR vendor (performance demonstrated in commercial plant); the number given is a conservative approach, as the CO2 purification step would further reduce sulfur content.

Flow rate offgas: 7883.5 kmol/h SOx concentration offgas: 10 ppmv SOx flow rate: 7883.5 kmol/h x 10 ppmv / 10^{6} = 0.0788 kmol/h Molecular weight SOx: 66 kg/kmol SOx flow rate: 0.0788 kmol/h x 66 kg/kmol = 5.20 kg/h Conversion factor: 1 lb = 0.454 kg Hourly SOx flow rate: 5.20 kg/h / 0.454 kg/lb = 11.56 lb/h Yearly SOx flow rate: 11.56 lb/h x 8000 h / 2000 lb/ton = 46.25 tons/yr

$-_{PM}$

Not applicable (The CO2 offgas contains no PM)

- VOC

Not applicable (The CO2 offgas contains no VOCs)

Regeneration Off-Gas

Emission is caused by venting of regeneration Off-gas from the MTG plant (regeneration of catalyst). Calculation of flow rates and emission periods based on actual design and commercial plant experience

$- CO$

Calculation of emissions based on maximum specific CO concentration of regeneration offgas emitted to atmosphere; regeneration offgas will be routed to CO2 purification section to remove any CO contained down to less than 1 ppm prior to emitting to atmosphere

CO concentration of regeneration offgas emitted: 1 ppmv

Yearly Emission:

Yearly Regen Offgas Flowrate: 10088680 m3n/yr = 356.3 MMscf/yr CO flow: 10088680 m³n/yr x 1 ppmv / 10^6 = 10.1 m³n/hr Molecular weight CO: 28 kg/mol Molar volume: 22.414 m³/kmol CO flow: 10.1 m³n/yr / 22.414 m³/kmol x 28 kg/kmol = 12.6 kg/yr Conversion factor: 1 lb = 0.454 kg Yearly CO flow rate: 25206 kg/yr / 0.454 kg/lb = 27.76 lb/yr 55519.8 lb/yr / 2000 lb/ton = 0.014 tons/yr

Max. Hourly Emission: Max Hourly Regen Offgas Flowrate: 7000 m3n/hr = 247 203 scf/hr CO flow: 7000 m³n/hr x 2000 ppmv / 10^6 = 0.007 m³n/hr Molecular weight CO: 28 kg/mol Molar volume: 22.414 m³/kmol CO flow: 0.007 m³n/hr / 22.414 m³/kmol x 28 kg/kmol = 0.0087 kg/hr Conversion factor: 1 lb = 0.454 kg Maximum Hourly CO flow rate: 17.5 kg/hr / 0.454 kg/lb = 0.0193 lb/hr

$-$ NO x

Not applicable (Regeneration temperatures too low for NO_x formation)

$-SOx$

Not applicable (No sulfur in regeneration gas or catalyst)

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Not applicable (Regeneration gas does not contain any particle matter)

- VOC

Not applicable (Regeneration gas used is nitrogen / air and does not contain VOC)

Total

 $-CO$ **Yearly Emission** AGR CO2 + Regeneration Off-Gas = Total 1.96 tons/yr + 0.014 tons/yr = 1.974 tons/yr

Max. Hourly Emission AGR CO2 + Regeneration Off-Gas = Total 0.49 lb/hr + 0.0193 lb/hr = 0.5093 lb/hr

- NOx Not applicable

- SOx **Yearly Emission** AGR CO2 + Regeneration Off-Gas = Total 46.25 tons/yr + 0.0 tons/yr = 46.25 tons/yr

Max. Hourly Emission AGR CO2 + Regeneration Off-Gas = Total 11.56 lb/hr + 0.0 lb/hr = 11.56 lb/hr

 $-_{PM}$ Not applicable

- VOC Not applicable

C2 Acid Gas Removal 3.7

Emissions are from flaring of clean syngas during start-up. Flaring rate and times has been based on a typical start-up sequence as demonstrated in commercial plants. Number of start-ups has been estimated based on availability figures drawn from commercial experience, the specific TGDS plant design (e.g. parallel trains etc.) and modified by adjustments made such as integration of redundancies, inclusion of lessons learnt etc (4 starts per year; 0.5 hr per start, with 50% load; i.e. max. 140 tons/hr syngas)).

- CO

Specific CO concentration in flare offgas: assumption acc. flare vendor information (1000 ppmv) Syngas rate to flare: 140 tons/hr Flue gas rate: 863207 m³n/h CO concentration flue gas: 1000 ppmv CO flow: 863207 m³n/h x 1000 ppmv / 10^6 = 863.207 m³n/h Molecular weight CO: 28 kg/mol Molar volume: 22.414 m³/kmol CO flow: 863.207 m³n/h / 22.414 m³/kmol x 28 kg/kmol = 1079 kg/h Conversion factor: 1 $lb = 0.454$ kg Hourly CO flow rate: 1079 kg/h / 0.454 kg/lb = 2375 lb/h Yearly CO flow rate: 2375 lb/h x 4 starts per year x 0.5 h / 2000 lb/ton = 2.4 tons/yr

- NOx

Specific NOx concentration in flare offgas: assumption acc. flare vendor information (250 ppmv)

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Syngas rate to flare: 140 tons/hr Flue gas rate: 863207 m³n/h NOx concentration flue gas: 250 ppmv NOx flow: 863207 m³n/h x 250 ppmv / 10^6 = 215.80 m³n/h Molecular weight NO2: 46 kg/mol Molar volume: 22.414 m³/kmol NOx flow: 215.80 m³n/h / 22.414 m³/kmol x 46 kg/kmol = 442.9 kg/h Conversion factor: 1 lb = 0.454 kg Hourly NOx flow rate: 481.4 kg/h $/$ 0.454 kg/lb = 957.6 lb/h Yearly NOX flow rate: 957.6 lb/h x 4 starts per year x 0.5 h / 2000 lb/ton = 0.96 tons/yr

 $-SOx$

Calculated based on conservative assumed average sulfur content of syngas flared of 100 ppmy during start-up. Conservative estimate, as normally the physical solvent process applied in the AGR achieves design specification for sulfur (less than 1 ppm) very quickly. Syngas rate to flare: 140 tons/hr Syngas rate to flare: 11561 kmol/h SOx concentration syngas gas: 100 ppmv Molecular weight SOx: 66 kg/mol SOx flow: 11561 kmol/h x 100 ppm x 66 kg/kmol / 10^6 = 76.3 kg/h Conversion factor: 1 lb = 0.454 kg Hourly SOx flow rate: 76.3 kg/h / 0.454 kg/lb = 168.1 lb/h Yearly SOx flow rate: 168.1 lb/h x 4 starts per year x 0.5 h / 2000 lb/ton = 0.17 tons/yr

- PM

Not applicable (smokeless flare acc. table 13.5-1 of AP42)

 $-_{VOC}$

Total flow rate to flare: 11561.1 kmol/h (50% load) Methane (HC) concentration: 0.006 mol% Methane (HC) flow: 11561.1 kmol/h x 0.006 mol% / 100 mol% = 0.695 kmol/h Heating value methane 890000 kJ/kmol Methane heat flow: 0.695 kmol/h x 890000 kJ/koml / 3600 s/h / 1000 kJ/MJ = 0.172 MW Conversions factor: 1 MMBTU/hr = 0.29308 MW Methane heat flow: 0.172 MW / 0.29308 MW/MMBTU/hr = 0.585 MMBTU/hr Emission factor Total Hydrocarbons acc. table 13.5-1 of AP 42: 0.14 lb/10^6BTU Hourly HC emission: 0.585 MMBTU/hr x 0.14 lb/10^6BTU = 0.082 lb/hr methane Hourly VOC emission: 0.082 lb/hr + 5.73 lb/hr MeOH (refer to section $2.8.4$) = 5.81 Yearly HC emission: 0.082 lb/hr x 4 starts/yr x 0.5 hr/starts / 2000 lb/tons = 0.000082 tons/yr Yearly VOC emission: 0.000082 tons/yr + 0.006 tons/yr MeOH (refer to section 2.8.4) = 0.006 tons/yr

3.8 **E1 MTG Reaction**

Emissions from flue gas of fired heater used during start-up and regeneration of MTG catalyst; Calculation of flow rates and emission periods based on actual design and commercial plant experience (Average 25887.5 MMBTU/yr; Max. 30 MMBTU/hr). During normal operation heater is fired with syngas, during front end shut down (i.e. operation of MTG plant from storage, no syngas available) no regeneration of catalyst will be performed, i.e. heater will not be operated. Syngas (used as fuelgas) has a lower heating value (LHV) of 296 BTU/scf. Therefore average fuelgas flowrate is 87.46 MMscf/yr (max. 101351 scf/hr). Flue Gas Flow rate can be derived from fuelgas flowrate (for syngas fuelgas) by multiplying with a factor of 3.02.

 $-CO$

CO specific concentration based on industrial standard for fired heaters (120 ppmv)

Yearly Emission: Flue gas rate: 7475169 m $\frac{3}{1}$ yr = 264 MMscf/hr

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CO concentration flue gas: 120 ppmv CO flow: 7475169 m³n/yr x 120 ppmv / 10 \textdegree 6 = 897.02 m³n/yr Molecular weight CO: 28 kg/mol Molar volume: 22.414 m³/kmol CO flow: 897.02 m³n/yr / 22.414 m³/kmol x 28 kg/kmol = 1120.6 kg/yr Conversion factor: 1 lb = 0.454 kg Yearly CO flow rate: 1120.6 kg/yr / 0.454 kg/lb = 2468.2 lb/yr Yearly CO flow rate: 2468.2 lb/yr / 2000 lb/ton = 1.24 tons/yr

Max. Hourly Emissions: Max. flue gas rate: 8862 m^3 n/hr = 306080 scf/hr CO concentration flue gas: 120 ppmv CO flow: 8862 m³n/hr x 120 ppmv / 10^6 = 1.063 m³n/hr Molecular weight CO: 28 kg/mol Molar volume: 22.414 m³/kmol CO flow: 1.063 m³n/hr / 22.414 m³/kmol x 28 kg/kmol =1.33 kg/hr Conversion factor: $1 lb = 0.454 kg$ Maximum Hourly CO flow rate: 1.33 kg/hr / 0.454 kg/lb = 1.93 lb/hr

$-$ NO x

NOx specific concentration based on industrial standard for fired heaters (100 ppmv)

Yearly Emission:

Flue gas rate: 7475169 m³n/yr = 264 MMscf/hr NOx concentration flue gas: 100 ppmv NOx flow: 7475169 m³n/yr x 100 ppmv / 10^6 = 747.5 m³n/yr Molecular weight NO2: 46 kg/mol Molar volume: 22.414 m³/kmol NOx flow: 747.5 m³n/yr / 22.414 m³/kmol x 46 kg/kmol = 1534.1 kg/yr Conversion factor: $1 lb = 0.454 kg$ Yearly NOx flow rate: 1534.1.1 kg/yr / 0.454 kg/lb = 3379 lb/yr Yearly NOx flow rate: 3679 lb/yr / 2000 lb/ton = 1.69 tons/yr

Max. Hourly Emissions: Max. flue gas rate: 8862 m³n/hr = 306080 scf/hr NOx concentration flue gas: 100 ppmv NOx flow: 8862 m³n/hr x 100 ppmv / 10^6 = 0.886 m³n/hr Molecular weight NOx: 46 kg/mol Molar volume: 22.414 m³/kmol NOx flow: 0.886 m³n/hr / 22.414 m³/kmol x 46 kg/kmol = 1.82 kg/hr Conversion factor: 1 lb = 0.454 kg Maximum Hourly NOx flow rate: 1.82 kg/hr / 0.454 kg/lb = 4.01 lb/hr

- SOx

Not applicable (Heated with sulfur-free syngas)

- PM

Based on natural gas estimated PMs (since syngas is used as fuel gas (main components CO and H2) the real values will be significantly lower, as soot formation from syngas is lower than from natural gas):

Yearly Emission Heat Requirement 25887.5 MMBTU/yr

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Emission factor acc. Table 1.4-2 of AP 42 (PM total): 7.6 lb/10^6 scf Average heating value of natural gas: 1020 BTU/scf Yearly PM emission: 25887.5 MMBTU/yr x 7.6 lb/10^6 scf / (1020 MMBTU/ 10^6 scf) = 192.9 lb/yr 192.9 lb/yr / 2000 lb/ton = 0.10 tons/yr

Max. Hourly Emission: Max. Heat Requirement 30 MMBTU/hr Emission factor acc. Table 1.4-2 of AP 42 (PM total): 7.6 lb/10^6 scf Average heating value of natural gas: 1020 BTU/scf Max. Hourly PM emission: 30 MMBTU/hr x 7.6 lb/10^6 scf / (1020 MMBTU/ 10^6 scf) = 0.223 lb/hr

 $-$ VOC

Based on natural gas estimated VOC (since syngas is used as fuel gas (main components CO and H2) the real values will be significantly lower since syngas has no source VOC):

Yearly Emissions:

Heat Requirement 25887.5 MMBTU/yr Emission factor acc. Table 1.4-2 of AP 42 (VOC): 5.5 lb/10^6 scf Average heating value of natural gas: 1020 BTU/scf Yearly VOC emission: 25887.5 MMBTU/yr x 5.5 lb/10^6 scf / 1020 MMBTU/yr /10^6 scf = 139.6 lb/yr 139.6 lb/yr / 2000 lb/ton = 0.07 tons/yr

Max. Hourly Emissions

Max. Heat Requirement 30 MMBTU/hr Emission factor acc. Table 1.4-2 of AP 42 (VOC): 5.5 lb/10^6 scf Average heating value of natural gas: 1020 BTU/scf Max. Hourly VOC emission: 30 MMBTU/yr x 5.5 lb/10^6 scf / (1020 MMBTU /10^6 scf) = 0.162 lb/hr

3.9 **E2 MTG Reaction**

Emissions from flue gas of fired heater used during start-up and regeneration /reactivation of MTG catalyst; Calculation of flow rates and emission periods based on actual design and commercial plant experience (67466.4 MMBTU/yr; Max. 120 MMBTU/hr). During normal operation heater is fired with syngas, during front end shut down (i.e. operation of MTG plant from storage, no syngas available) no regeneration of catalyst will be performed, i.e. heater will not be operated. Syngas (used as fuelgas) has a lower heating value (LHV) of 296 BTU/scf. Therefore average fuelgas flowrate is 227.9 MMscf/yr (max. 405405 scf/hr). Flue Gas Flow rate can be derived from fuelgas flowrate (for syngas fuelgas) by multiplying with a factor of 3.02.

 $-CO$

CO specific concentration based on industrial standard for fired heaters (120 ppmv)

Yearly Emission: Flue gas rate: 19479418 m³n/yr = 687.9 MMscf/yr CO concentration flue gas: 120 ppmv CO flow: 19479418 m³n/yr x 120 ppmv / 10^6 = 2337.5 m³n/yr Molecular weight CO: 28 kg/mol Molar volume: 22.414 m³/kmol CO flow: 2337.5 m³n/yr / 22.414 m³/kmol x 28 kg/kmol = 2920 kg/yr Conversion factor: 1 lb = 0.454 kg

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Yearly CO flow rate: 2920 kg/yr / 0.454 kg/lb = 6432 lb/yr Yearly CO flow rate: 6432 lb/yr / 2000 lb/ton = 3.22 tons/yr

Max. Hourly Emission: Max. flue gas rate: 34647 m^3 n/hr = 1223547 scf/hr CO concentration flue gas: 120 ppmv CO flow: 34647 m³n/hr x 120 ppmv / 10^6 = 4.16 m³n/hr Molecular weight CO: 28 kg/mol Molar volume: 22.414 m³/kmol CO flow: 4.16 m³n/hr / 22.414 m³/kmol x 28 kg/kmol = 5.19 kg/hr Conversion factor: 1 $lb = 0.454$ kg Maximum Hourly CO flow rate: 5.19 kg/hr / 0.454 kg/lb = 11.44 lb/hr

-NO_x

NOx specific concentration based on industrial standard for fired heaters (100 ppmv)

Yearly Emission: Flue gas rate: 19479418 m³n/yr = 687.9 MMscf/yr NOx concentration flue gas: 100 ppmv NOx flow: 19479418 m³n/yr x 100 ppmv / 10^6 = 1948 m³n/yr Molecular weight NO2: 46 kg/mol Molar volume: 22.414 m³/kmol NOx flow: 1948 m³n/yr / 22.414 m³/kmol x 46 kg/kmol = 3997.9 kg/yr Conversion factor: 1 $lb = 0.454$ kg Yearly NOx flow rate: 3997.9 kg/yr / 0.454 kg/lb = 8806 lb/yr Yearly NOx flow rate: 8806 lb/yr / 2000 lb/ton = 4.4 tons/yr

Max. Hourly Emissions: Max. flue gas rate: 34647 m³n/hr= 1223547 scf/hr NOx concentration flue gas: 100 ppmv NOx flow: 34647 m³n/hr x 100 ppmv / 10^6 = 3.46 m³n/hr Molecular weight NO2: 46 kg/mol Molar volume: 22.414 m³/kmol NOx flow: 3.46 m³n/hr / 22.414 m³/kmol x 46 kg/kmol = 7.1 kg/hr Conversion factor: 1 $lb = 0.454$ kg Maximum Hourly NOx flow rate: 7.1 kg/hr / 0.454 kg/lb = 15.64 lb/hr

- SOx

Not applicable (Heated with sulfur-free syngas)

- PM

Based on natural gas estimated PMs (since syngas is used as fuel gas (main components CO and H2) the real values will be significantly lower, as soot formation from syngas is lower than from natural gas):

Yearly Emission: Heat Requirement 67466.4 MMBTU/yr Emission factor acc. Table 1.4-2 of AP 42 (PM total): 7.6 lb/10^6 scf Average heating value of natural gas: 1020 BTU/scf Yearly PM emission: 67466.4 MMBTU/yr x 7.6 lb/10^6 scf / 1020 MMBTU/yr /10^6 scf = 502.7 lb/yr 502.7 lb/yr / 2000 lb/ton = 0.25 tons/yr

Max. Hourly Emission:

Max. Heat Requirement 120 MMBTU/hr Emission factor acc. Table 1.4-2 of AP 42 (PM total): 7.6 lb/10^6 scf Average heating value of natural gas: 1020 BTU/scf

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Max. Hourly PM emission: 120 MMBTU/hr x 7.6 lb/10^6 scf / (1020 MMBTU/ 10^6 scf) = 0.89 lb/hr

 $-VOC$

Based on natural gas estimated VOC (since syngas is used as fuel gas (main components CO and H2) the real values will be significantly lower since syngas has no source VOC)::

Yearly Emission Heat Requirement 67466.4 MMBTU/yr Emission factor acc. Table 1.4-2 of AP 42 (VOC): 5.5 lb/10^6 scf Average heating value of natural gas: 1020 BTU/scf Yearly VOC emission: 67466.4 MMBTU/yr x 5.5 lb/10^6 scf / 1020 MMBTU/yr /10^6 scf = 363.8 lb/yr 363.8 lb/yr / 2000 lb/ton = 0.18 tons/yr

Max. Hourly Emission:

Max. Heat Requirement 120 MMBTU/hr Emission factor acc. Table 1.4-2 of AP 42 (VOC): 5.5 lb/10^6 scf Average heating value of natural gas: 1020 BTU/scf Max. Hourly VOC emission: 120 MMBTU/yr x 5.5 lb/10^6 scf / (1020 MMBTU /10^6 scf) = 0.65 lb/hr

3.10 **E3 MTG HGT**

Emissions from flue gas of fired heater used as pre-heater for the heavy gasoline treatment step in the MTG plant. Calculation of flow rates based on actual design and commercial plant experience (25280 MMBTU/yr; Max. 4 MMBTU/hr). During normal operation heater is fired with syngas, during front end shut down (i.e. operation of MTG plant from storage, no syngas available), operation with MTG fuel gas (this will be 4 times a year, each 10 hours, refer also to Emission Point E5). Syngas (used as fuelgas) has a lower heating value (LHV) of 296 BTU/scf. Therefore average fuelgas flowrate is 85.41 MMscf/yr (max. 13514 scf/hr). Flue Gas Flow rate can be derived from fuelgas flowrate (for syngas fuelgas) by multiplying with a factor of 3.02.

 $-CO$

CO specific concentration based on industrial standard for fired heaters (120 ppmv)

Yearly Emission: Flue gas rate: 7296000 m³n/yr = 258 MMscf/yr CO concentration flue gas: 120 ppmv CO flow: 7296000 m³n/yr x 120 ppmv / 10^6 = 875.5 m³n/yr Molecular weight CO: 28 kg/mol Molar volume: 22.414 m³/kmol CO flow: 875.5 m³n/yr / 22.414 m³/kmol x 28 kg/kmol = 1094 kg/yr Conversion factor: 1 lb = 0.454 kg Yearly CO flow rate: 1094 kg/yr / 0.454 kg/lb = 2409 lb/yr Yearly CO flow rate: 2409 lb/yr / 2000 lb/ton = 1.20 tons/yr

Max. Hourly Emission: Max. flue gas rate: 1155 m³n/hr = 40812 scf/hr CO concentration flue gas: 120 ppmv CO flow: 1155 m³n/hr x 120 ppmv / 10^6 = 0.14 m³n/hr Molecular weight CO: 28 kg/mol Molar volume: 22.414 m³/kmol CO flow: 0.14 m³n/hr / 22.414 m³/kmol x 28 kg/kmol = 0.17 kg/hr Conversion factor: 1 lb = 0.454 kg

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Maximum Hourly CO flow rate: 0.17 kg/hr / 0.454 kg/lb = 0.381 lb/hr

- NOx

NOx specific concentration based on industrial standard for fired heaters (100 ppmv)

Yearly Emission:

Flue gas rate: 7296000 m³n/yr = 258 MMscf/yr NOx concentration flue gas: 100 ppmv NOx flow: 7296000 m³n/yr x 100 ppmv / 10^6 = 729.6 m³n/yr Molecular weight NO2: 46 kg/mol Molar volume: 22.414 m³/kmol NOx flow: 729.6 mªn/yr / 22.414 mª/kmol x 46 kg/kmol = 1497.4 kg/yr Conversion factor: $1 lb = 0.454$ kg Yearly NOx flow rate: 1497.4 kg/yr / 0.454 kg/lb = 3298.2 lb/yr Yearly NOx flow rate: 3298.2 lb/yr / 2000 lb/ton = 1.65 tons/yr

Max. Hourly Emissions: Max. flue gas rate: 1155 m³n/hr = 40812 scf/hr NOx concentration flue gas: 100 ppmv NOx flow: 1155 m³n/hr x 100 ppmv / 10^6 = 0.12 m³n/hr Molecular weight NO2: 46 kg/mol Molar volume: 22.414 m³/kmol NOx flow: 0.12 m³n/hr / 22.414 m³/kmol x 46 kg/kmol = 0.25 kg/hr Conversion factor: $1 lb = 0.454 kg$ Maximum Hourly NOx flow rate: 0.26 kg/hr / 0.454 kg/lb = 0.55 lb/hr

- SOx

Not applicable (Heated with sulfur free syngas)

- PM

Based on natural gas estimated PMs (since syngas is used as fuel gas (main components CO and H2) the real values will be significantly lower, as soot formation from syngas is lower than from natural gas):

Yearly Emission:

Heat Requirement 25280 MMBTU/yr Emission factor acc. Table 1.4-2 of AP 42 (PM total): 7.6 lb/10^6 scf Average heating value of natural gas: 1020 BTU/scf Yearly PM emission: 25280 MMBTU/yr x 7.6 lb/10^6 scf / 1020 MMBTU/yr /10^6 scf = 188.4 lb/yr 188.4 lb/yr / 2000 lb/ton = 0.09 tons/yr

Max. Hourly Emission:

Max. Heat Requirement 4 MMBTU/hr Emission factor acc. Table 1.4-2 of AP 42 (PM total): 7.6 lb/10^6 scf Average heating value of natural gas: 1020 BTU/scf Max. Hourly PM emission: 4 MMBTU/hr x 7.6 lb/10^6 scf / (1020 MMBTU/ 10^6 scf) = 0.03 lb/hr

$-$ VOC

Based on natural gas estimated VOC (since syngas is used as fuel gas (main components CO and H2) the real values will be significantly lower since syngas has no source VOC):

Yearly Emission:

Heat Requirement 25280 MMBTU/yr Emission factor acc. Table 1.4-2 of AP 42 (VOC): 5.5 lb/10^6 scf

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Average heating value of natural gas: 1020 BTU/scf Yearly VOC emission: 25280 MMBTU/yr x 5.5 lb/10^6 scf / 1020 MMBTU/yr /10^6 scf = 163.3 lb/yr 163.3 lb/yr / 2000 lb/ton = 0.07 tons/yr

Max. Hourly Emission:

Max. Heat Requirement 4 MMBTU/hr Emission factor acc. Table 1.4-2 of AP 42 (VOC): 5.5 lb/10^6 scf Average heating value of natural gas: 1020 BTU/scf Max. Hourly VOC emission: 4 MMBTU/yr x 5.5 lb/10^6 scf / (1020 MMBTU /10^6 scf) = 0.022 lb/hr

E4 MTG Reaction 3.11

MTG Regeneration Off-Gas is routed to the CO2 purification (Stream 31). All emissions of the Regeneration Off-Gas is added to Emission Point C1.

3.12 **E5 MTG Separation**

Emission is caused by flaring of MTG tail gas, when MTG plant is in operation (using MeOH feed from storage) and entire front end plant is down - in this case tailgas cannot be recycled to front end plant as normally. Flaring rate is based on actual design and commercial plant experience. Flaring periods are derived from availability data of commercial plants for different plant sections, as well as current design and modification such as lessons learnt etc. (4 times per year, each 10 hr, total flow rate 3.89 tons/h); The approach to calculate emissions based on complete flaring MTG tailgas is conservative, because actually part of the tail gas could also be utilized as fuel gas in fired heaters of the MTG process. This will further reduce the emissions.

 $-CO$

Specific CO concentration in flare offgas: assumption acc. flare vendor information (1000 ppmv) Flowrate to flare: 7780 lb/hr Flue gas rate: 18853 m^3 n/h = 665781 scf/hr CO concentration flue gas: 1000 ppmv CO flow: 18853 m³n/h x 1000 ppmv / 10^6 = 18.85 m³n/h Molecular weight CO: 28 kg/mol Molar volume: 22.414 m³/kmol CO flow: 18.85 m³n/h / 22.414 m³/kmol x 28 kg/kmol = 23.55 kg/h Conversion factor: 1 lb = 0.454 kg Hourly CO flow rate: 23.55 kg/h / 0.454 kg/lb = 51.88 lb/h Yearly CO flow rate: 51.88 lb/h x 4 times per year x 10 h / 2000 lb/ton = 1.04 tons/yr

- NOx

Specific NOx concentration in flare offgas: assumption acc. flare vendor information (250 ppm) Flowrate to flare: 7780 lb/hr Flue gas rate: 18853 $m^3n/h = 665781$ scf/hr NOx concentration flue gas: 250 ppmv NOx flow: 18853 m³n/h x 250 ppmv / 10^6 = 4.71 m³n/h Molecular weight NO2: 46 kg/mol Molar volume: 22.414 m³/kmol NOx flow: 4.71 m³n/h / 22.414 m³/kmol x 46 kg/kmol = 9.67 kg/h Conversion factor: $1 lb = 0.454 kg$ Hourly NOx flow rate: 9.67 kg/h / 0.454 kg/lb = 21.3 lb/h Yearly NOX flow rate: 21.3 lb/h x 4 times per year x 10 h / 2000 lb/ton = 0.43 tons/yr

$-SOx$

Not applicable (no sulfur in tailgas)

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- PM

Not applicable (smokeless flare acc. table 13.5-1 of AP42)

 $-_{VOC}$

Flow rate to flare: 7780 lb/hr

Heating value: 18325 BTU/lb

Heat input to flare: 7780 lb/hr x 18325 BTU/lb = 142 600 000 BTU/hr = 142.6 MMBTU/hr Emission factor Total Hydrocarbons acc. table 13.5-1 of AP 42: 0.14 lb/MMBTU Total Hydrocarbons emission: 142.6 MMBTU/hr x 0.14 lb/MMBTU = 19.96 lb/hr Yearly HC emission: 19.96 lb/hr x 4 times per yr x 10 hr/times per yr = 798.4 lb/yr 798.4 lb/yr / 2000 lb/tons = 0.4 tons/yr

3.13 F Start-up Steam Boiler

Emission results from firing of natural gas in a start-up steam boiler (81.84 MMBTU/hr) during cold start-up of the plant; Flow rates and periods for fuel gas are derived from calculation based on assumed steam requirements and start-up time for CTL plant start-up as per other design studies for CTL plants (4 starts per year, each 96 hr); Operation with natural gas, which has an heating value of 1020 BTU/scf. Fuelgas flow rate is 83890 scf/hr. Flue gas rate can be derived from fuel gas flowrate (for natural gas) by multiplying with a factor of approx. 12.9.

- CO

CO specific concentration based on industrial standard for fired heaters (120 ppmv) Flue gas rate: 30631 m^3 n/h = 1081724 scf/hr CO concentration flue gas: 120 ppmv CO flow: 30631 m³n/h x 120 ppmv / 10^6 = 3.68 m³n/h Molecular weight CO: 28 kg/mol Molar volume: 22.414 m³/kmol CO flow: 3.68 m³n/h / 22.414 m³/kmol x 28 kg/kmol = 4.59 kg/h Conversion factor: 1 $lb = 0.454$ kg Hourly CO flow rate: 4.59 kg/h / 0.454 kg/lb = 10.11 lb/h Yearly CO flow rate: 10.11 lb/h x 4 starts per year x 96 h per start / 2000 lb/ton = 1.94 tons/yr

- NOx

NOx specific concentration based on industrial standard for fired heaters (100 ppmv) Flue gas rate: 30631 m³n/h = 1081724 scf/hr NOx concentration flue gas: 100 ppmv NOx flow: 30831 m³n/h x 100 ppmv / 10^6 = 3.08 m³n/h Molecular weight NO2: 46 kg/mol Molar volume: 22.414 m³/kmol NOx flow: 3.08 m³n/h / 22.414 m³/kmol x 46 kg/kmol = 6.32 kg/h Conversion factor: $1 lb = 0.454$ kg Hourly NOx flow rate: 6.32 kg/h / 0.454 kg/lb = 13.92 lb/h Yearly NOX flow rate: 13.92 lb/h x 4 starts per year x 96 h per start / 2000 lb/ton = 2.67 tons/yr

- SOx

Calculated acc. max. sulfur content of natural gas specified by TGDS Fuel gas rate: 2380 m^3 n/h = 83890 scf/hr Molar volume: 22.414 m³/kmol Fuel gas rate: 2380 m³n/h / 22.414 = 106 kmol/h Sulfur concentration syngas gas: 20 ppmv Molecular weight sulfur: 66 kg/mol SOx flow: 106 kmol/h x 20 ppm x 66 kg/kmol / 10^6 = 0.14 kg/h Conversion factor: $1 lb = 0.454 kg$ Hourly SOx flow rate: 0.14 kg/h / 0.454 kg/lb = 0.31 lb/h Yearly SOx flow rate: 0.31 lb/h x 4 starts per year x 0.5 h / 2000 lb/ton = 0.06 tons/yr

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- PM

Based on natural gas estimated PMs:

Heat Requirement 81.84 MMBTU/hr

Average heating value of natural gas: 1020 BTU/scf

Emission factor acc. Table 1.4-2 of AP 42 (PM total): 7.6 lb/10^6 scf

Hourly PM emission: 81.84 MMBTU/yr x 7.6 lb/10^6 scf / 1020 MMBTU/yr /10^6 scf = 0.61 lb/hr Yearly PM emission: 0.61 lb/yr / 2000 lb/ton x 4 start/yr x 96 hr/start = 0.12 tons/yr

- VOC

Based on natural gas estimated VOC (since syngas is used as fuel gas (main components CO and H2) the real values will be lower):

Heat Requirement 81.84 MMBTU/hr

Emission factor acc. Table 1.4-2 of AP 42 (VOC): 5.5 lb/10^6 scf

Average heating value of natural gas: 1020 BTU/scf

Hourly VOC emission: 81.84 MMBTU/hr x 5.5 lb/10^6 scf / 1020 MMBTU/yr /10^6 scf = 0.44 lb/yr Yearly VOC emission: 0.44 lb/yr / 2000 lb/ton x 4 start/yr x 96 hr/start = 0.09 tons/yr

3.14 **G** Flare

Emission result from pilot flame operation of the flare (operating with natural gas); Flow rates and number of pilot burners are based on typical flare data; Flare Flue gas flow can be derived from natural gas feed to pilot burners multiplied with factor of approx. 12.7

 $-CO$

Specific CO concentration in flare offgas: assumption based on conversion rate of 99.5 (typical value for natural gas burning in flare)

Natural gas flow rate to flare: 25 m^3 n/h = 880 scf/hr

Natural gas conversion rate: 99.5 %

Molecular weight CO: 28 kg/mol

Molar volume: 22.414 m³/kmol

CO flow: (1-0.995) x 25 m³n/h / 22.414 m³/kmol x 28 kg/kmol = 0.16 kg/h

Conversion factor: 1 lb = 0.454 kg

Hourly CO flow rate: 0.15 kg/h / 0.454 kg/lb = 0.35 lb/h

Yearly CO flow rate: 0.35 lb/h x 8760 h / 2000 lb/ton = 1.54 tons/yr

- NOx

Specific NOx concentration in flare offgas: assumption acc. flare vendor information (250 ppmv) Natural gas flow rate to flare: 25 m^3 n/h = 880 scf/hr Flue gas rate: 317.5 m^3 n/h = 11176 scf/hr NOx concentration flue gas: 250 ppmv NOx flow: 317.5 x 250 ppmv / 10^{16} = 0.08 m³n/h Molecular weight NO2: 46 kg/mol Molar volume: 22.414 m³/kmol NOx flow: 0.08 m³n/h / 22.414 m³/kmol x 46 kg/kmol = 0.164 kg/h Conversion factor: $1 lb = 0.454 kg$ Hourly NOx flow rate: 0.164 kg/h / 0.454 kg/lb = 0.36 lb/h Yearly NOX flow rate: 0.36 lb/h x 8760 h / 2000 lb/ton = 1.58 tons/yr

 $-SOx$

Calculated acc. max. sulfur content of natural gas specified by TGDS Natural gas flow rate to flare: 25 m^3 n/h = 880 scf/hr Molar volume: 22.414 m³/kmol Natural gas flow rate to flare: 25 m³n/h / 22.414 m³/kmol = 1.12 kmol/h SOx concentration natural gas: 20 ppmv Molecular weight SOx: 66 kg/mol SOx flow: 1.12 kmol/h x 20 ppm x 66 kg/kmol / 10^6 = 0.0015 kg/h

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Conversion factor: 1 lb = 0.454 kg Hourly SOx flow rate: 0.0015 kg/h / 0.454 kg/lb = 0.0033 lb/h Yearly SO flow rate: 0.0033 lb/h x 8760 h / 2000 lb/ton = 0.015 tons/yr

- PM

Flow rate to flare for pilot burners: $25 \text{ m}^3 \text{n/h} = 880 \text{ scf/hr}$ Conversions factor: $1 \text{ scf} = 0.02686 \text{ m}^3 \text{n}$ Flow rate: 25 m³n/h / 0.02686 m³n/scf = 931 scf/hr Emission factor total PM acc. table 1.4-2 of AP 42: 7.6 lb/10^6scf Hourly PM emission: 931 scf/hr x 7.6 lb/10^6scf / 10^6 = 0.0071 lb/hr Yearly PM emission: 0.0071 lb/hr x 8760 hr/yr = 62.2 lb/yr 62.2 lb/yr / 2000 lb/tons = 0.031 tons/yr

 $-_{VOC}$

Flow rate to flare for pilot burners: $25 \text{ m}^3 \text{n/h} = 880 \text{ scf/hr}$ Conversions factor: $1 \text{ scf} = 0.02686 \text{ m}^3 \text{n}$ Flow rate: 25 m³n/h / 0.02686 m³n/scf = 931 scf/hr Emission factor VOC acc. table 1.4-2 of AP 42: 5.5 lb/10^6scf Hourly VOC emission: 931 scf/hr x 5.5 lb/10^6scf / 10^6 = 0.0051 lb/hr Yearly VOC emission: 0.0051 lb/hr x 8760 hr/yr = 44.9 lb/yr 44.5 lb/yr / 2000 lb/tons = 0.022 tons/yr

3.15 **Safety Valves**

All safety relieve valves or rupture discs open only in case of emergencies or malfunctions. Typical relief cases are for example:

- § Fire / Explosion
- \$ Power failure
- § Utility (e.g. cooling water, stream) failure
- § Operator failure
- S Etc.

List of main safety valves

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3.16 Leak emissions estimate

The calculation of the fugitive emissions are based on the report EPA-453/R-95-017 (1995
Protocol for Equipment Leak Emission Estimates). Potential emission calculation is based on table 2-1, emission control is based on table 5-1 and 5-2.

For Detail calculation refer to attachment 3.

Leak emission overview CO

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Leak emission overview SOx (H2S as SOx equivalent)

Leak emission overview H2S (HAP)

Leak emission overview VOC

Leak emission overview Methanol (HAP)

$\overline{\mathbf{4}}$ **Block Flow Diagram**

Please find requested Block Flow Diagram as Attachment 1 hereto.

The points a1, b1, b2, b3, c1, c2, c3, c4, c5, c6, c6, d1, d2, e1 and e2 indicate emergency relieve points (safety valve or rupture disk) and therefore are not considered for overall emission calculation.

4.1 **Stream Numbers and designations**

Following hereafter is a short description of each stream shown in the Block Flow Diagram

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Emission Summary

In the "Remarks" columns the letters C, D and S have following meaning:

-
- C: Continuous during normal operation
D: Discontinuous during normal operation
- S: Occurs during start-up

CO:

Note 1: Max. Emissions from leaks cannot be quantified

NO_x

SO_x

<u>Fotal TPY | 92.104</u>
Note 1: Max. Emissions from leaks cannot be quantified

PM

 $... / 33$

VOC

 $\ldots/34$

Total TPY | 22.005
Note 1: Max. Emissions from leaks cannot be quantified

 $\ldots/35$

HAPS:

Note 1: Figure excluding leak emissions, as max. emissions from leaks cannot be quantified

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ATTACHMENT 2

TO TASK ORDER 1

RESPONSE ON DEP QUESTIONS

TRANSGAS DEVELOPMENT SYSTEMS, LLC

CTL PROJECT

Description of Emission Sources and Calculation in the Coal Preparation and Gasification

PREPARED BY

UHDE

TGDS CTL Project, West Virginia

TABLE OF CONTENTS

CTL Plant, Short Description of Emission Sources

1

Coal Preparation (Unit 111)

During the normal operation several process trains pulverize and dry the feed for the gasification. The feed is heated up to $80 - 110\degree$ in an inert atmosphere and the most of the moisture vaporizes. Fuel gas and steam are used for heat generation.

The water vapor containing gas is filtered in bag filters and discharged into the atmosphere – Emission Points A1. This gas contains following impurities:

- S Particulate matter the concentration is below 10 mg/m3 (wet basis), 5 mg/m3 is applied for TGDS CTL, which is industrial standard.
- § Sulfur oxides the concentration of SOx depends on concentration of sulfur compounds in the fuel gas and in other gases supplied to the gas burner. For TGDS plant set-up pure hydrogen will be used as fuel gas; SOx emission only occurs during start-ups, when entire plant is down, i.e. when no hydrogen is available
- $\text{\$}$ Carbon monoxide and NOx from the gas burner typical concentrations for nature gas combustion are 120 ppmv CO and 100 ppmv NOx $@3\%$ O₂. To reduce this emissions a gas with high hydrogen content can be used as fuel gas and the fuel gas based heating can be partly replaced by steam heating.

If the fuel gas and other gases fed to burner contain such nitrogen compounds as NH3 or HCN, these components are an additional source of NO_x.

For TGDS CTL pure hydrogen will be applied as fuel gas with the exception of start-up, when entire plant is down, i.e. when no hydrogen is available; Therefore CO from fuel gas only occurs during those events.

S Gases and vapors releases from the feed during milling and heating, mainly CO and VOC. The concentration depends on feedstock. For unknown feedstocks Uhde carry out milling and drying tests to determine these emissions.

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Emission Points A1 for the whole plant:

 S NO_x

Normal Operation:

- o 100 ppmvd NO_x @3% O₂ (industrial standard)
- hot gas generation, dry basis @3% O₂ 7 967 m³_n/h for each \circ gasifier train on the basis of 100% heat required being generated by combustion;

TGDS design: 78.1% of required heat is generated by combustion and 21.9% by steam condensation

 \circ resulting NO_x emission, calculated as NO₂:

7 967 m3n/hr x 100 ppmv NOx x 2 gasifiers x 25 tons/y x 78.1% (heat required by fuel gas) / 22,4 kmol / m3n x 46 kg/kmole (molecular weight of NO2) \times 8760 h / year = 22.645 metric ton / year;

 \Rightarrow 24.68 tons/year (1 short ton = 0.9072 metric ton)

There are four (4) A1 points operating at a time with a total of five points (A1/1, A1/2, A1/3, A1/4, & A1/5) so the maximum emission rate from an individual A1 emission point is 24.68 tons/year \div 4 = 6.17 tons/year

Start-up:

o Natural Gas used during start-up of coal preparation, when no hydrogen available

 \circ 100 ppmv NO_x in Offgas

hot gas generation, 32 218 m3n/hr with Start-up coal \circ

resulting NO_x emission: \circ

32 218 m3n/hr x 100 ppmv NOx x 2 gasifiers x 4 (times per year) x 10 hours (start-up time) / 22,4 kmol / m3n x 46 kg/kmole (molecular weight of NOx = 0.53 metric ton / year;

 \Rightarrow 0.58 tons/year (1 short ton = 0.9072 metric ton)

Overall: 25.26 tons/year

Hourly emission:

 \Rightarrow 24.68 ton/year * 2000 lb/ton / 8760 h/y = 5.63 lb/h

There are four (4) A1 points operating at a time with a total of five points (A1/1, A1/2, A1/3, A1/4, & A1/5) so the maximum emission rate from an individual A1 emission point is 5.63 lb/h \div 4 = 1.41 lb/h

Hourly emission during start-up:

 \Rightarrow 0.58 ton/year * 2000 lb/ton / 10h/y / 4 times per y / 4 coal preparation trains = 7.25 lb/h

Only one of total five emissions point is in operation during start-up

Particulate mater S

Normal Operation:

- \circ 5 mg/m3 PM concentration (industrial standard)
- 204 000 m3/h offgas from baghouse filter \circ
- o resulting emission:

5 mg/m3 x 204 000 m3/h x 8760 hr /year = 8.9 metric tons / year

 \Rightarrow 9.8 tons/year (1 short ton = 0.9072 metric ton)

There are four (4) A1 points operating at a time with a total of five points (A1/1, A1/2, A1/3, A1/4, & A1/5) so the maximum emission rate from an individual A1 emission point is 9.8 tons/year \div 4 = 2.45 tons/year

Start-up:

 \circ 5 mg/m3 PM concentration (industrial standard)

463 000 m3/h offgas from baghouse filter with start-up coal \bullet

 \circ resulting emission:

5 mg/m3 x 463 000 m3/h x 2 gasifier x 4 x 10hr /year = 0.185 metric tons / year

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 \Rightarrow 0.204 tons/year (1 short ton = 0.9072 metric ton)

Overall: 10.004 tons/year

Hourly Emission:

 \Rightarrow 9.8 ton/year * 2000 lb/ton / 8760 h/y = 2.24 lb/h

There are four (4) A1 points operating at a time with a total of five points (A1/1, A1/2, A1/3, A1/4, & A1/5) so the maximum emission rate from an individual A1 emission point is 2.24 $\frac{1}{2}$ h ÷ 4 = 0.56 lb/h Hourly emission during start-up:

 \Rightarrow 0.204 tons/year * 2000 lb/ton / 10h/y / 4 times per y / 4 coal preparation trains = 2.55 lb/h

Only one of total five emissions point is operation during start-up

VOC from the feedstock (normal operation and start-up) S

- 3.5 mg VOC/kg (dry ash free) coal based on investigation of other \circ coal
- o coal throughput, daf 260 tons/h
- resulting emission: \circ

3.5 mg VOC/kg (daf) x 260 tons/h x 8760 hours/yr = 8 tons/y

 \Rightarrow 8 ton/year * 2000 lb/ton / 8760 h/y = 1.83 lb/h

There are four (4) A1 points operating at a time with a total of five points (A1/1, A1/2, A1/3, A1/4, & A1/5) so the maximum emission rate from an individual A1 emission point is 8.0 tons/year \div 4 = 2.0 tons/year and 1.83 lb/h \div 4 = 0.4575 lb/h

S CO from the feedstock

Normal Operation: Coal only from devolatilization

- o 7.1 mg/kg coal dry based on investigation of a previous coal
- coal throughput 291 metric tons/h coal dry \circ
- resulting emission: \circ

7.1 mg/kg coal dry x 290 000 kg/hr coal dry $*$ 8760 hours / year = 18095 kg/yr CO => 20 tons / yr (1 short ton = 0.9072 metric ton) There are four (4) A1 points operating at a time with a total of five points (A1/1, A1/2, A1/3, A1/4, & A1/5) so the maximum emission rate from an individual A1 emission point is 20.0 tons/year \div 4 = 5.0 tons/year

Start-up:

- o Natural Gas used during start-up of coal preparation, when no hydrogen available
- o 120 ppmv CO in Offgas
- hot gas generation, 32 218 m3n/hr with Start-up coal \circ

resulting CO emission: \circ

32 218 m3n/hr x 120 ppmv CO x 2 gasifiers x 4 (times per year) x 10 hours (start-up time) / 22,4 kmol / m3n x 28 kg/kmole (molecular weight of CO) = 0.39 metric ton / year;

 \Rightarrow 0.43 tons/year (1 short ton = 0.9072 metric ton)

Overall: 20.43 tons/year

Hourly Emission:

 \Rightarrow 20 ton/year * 2000 lb/ton / 8760 h/y = 4.57 lb/h

There are four (4) A1 points operating at a time with a total of five points (A1/1, A1/2, A1/3, A1/4, & A1/5) so the maximum emission rate from an individual A1 emission point is 4.57 lb/h \div 4 = 1.1425 lb/h Hourly Emission during start-up:

 \Rightarrow 0.43 $*$ 2000 lb/ton / 10h/y / 4 times per y / 4 coal preparation trains $= 5.375$ lb/h

Only one of total five emissions point is operation during start-up

Only during start-up:

- o Natural Gas used during start-up of coal preparation, when no hydrogen available
- o 20 ppmv H2S in Natural Gas
- \circ hot gas generation, 47 968 m3n/hr with Start-up coal
- resulting SO_x emission: \circ

47 968 m3n/hr x 20 ppmv H2S x 2 gasifiers x 4 (times per year) x 10 hours (start-up time) / 22,4 kmol / m3n x 64 kg/kmole (molecular weight of SOx) = 0.22 metric ton / year;

 \Rightarrow 0.24 tons/year (1 short ton = 0.9072 metric ton)

 \Rightarrow 0.24 ton/year * 2000 lb/ton / 40 h/y / 4 coal preparation trains = 3 Ib/h (Note: only one coal train started at a time, i.e. one gasifier at 50% load)

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Coal Dust Feeding (Unit 112)

The feedstock from the coal preparation enters the feed dust bunker via the cyclone filter, from where it flows by gravity to the lock hoppers.

Once a lock hopper is filled with feed dust, the lock hopper is pressurized from atmospheric pressure to about 1 to 2 bar above the feed bin working pressure with an inert gas (N2 or CO2). The selection of inert gas depends on the field of application of the product gas (Syngas) of the gasification process.

The pressurized feed dust from one or more lock hoppers is continuously supplied via a dense phase conveying system to the feed feed bin and the empty lock hopper is depressurized. The gas from lock hoppers is preliminary filtered in dust filter, expanded, filtered again in a filter placed above the feed dust bunker and discharged into the atmosphere (Emission Points B1).

The released lock hopper gas contains particulate mater. If the inert gas contains any impurities, e.g. CO, these impurities are discharded into the atmosphere.

The pressure in Feed Bin is about 5 bar higher than the gasifier pressure and it is controlled and kept on the right level by injection of a inert gas into the feed bin or recycle of a relative small amount of gas. In case of failure of the gas recycle the relieved gas is discharged into the atmosphere (Emission Points B3). This gas contains the inert gas impurities and additional a small amount of particulate mater.

Before start-up of a Gasifier the dense phase conveying system is taken into operation and a specified steady feeding stream in pipes downstream lock hoppers has to be adjusted. However, the feed is not routed into the Gasifier but into the start-up vessel, where the dust is settled down and the gas is dedusted and discharged into the atmosphere (Emission Points A2).

If a low sulfur start up feedstock is used, the feed bin has to be emptied first. The feed bin content - normal sulfur feedstock - has to be recycled pneumatically via Start-up Vessel to the Feed Dust Bunker. The transport gas is dedusted and released into the atmosphere (Emission Points A2).

Emission Points B1:

- \$ Particulate mater
	- o 5 mg/m3 PM (industrial standard from baghouse)
	- o discharge from lock hoppers 45 300 m3/h @ 0.9 bar, 90 ℃
	- o feedstock change 30 times/year/train
		- 32.000 $m³$ /feed change

o resulting emission:

45 300 m3n/h x 5 mg/m3 x 8760 h/yr + 30 starts/yr/train x 32 000 m3n/h * 5 mg/m3 * 2 trains = 1.99 metric tons / yr = > 2.2 tons/y (1 short ton = 0.9072 metric ton)

 \Rightarrow 2.2 ton/year * 2000 lb/ton / 8760 h/y = 0.50 lb/h

There are two (2) B1 points $(B1/1 \& B1/2)$ operating at a time so the maximum emission rate from an individual B1 emission point is 2.2 tons/year $\div 2 = 1.1$ tons/year and 0.50 lb/h $\div 2 = 0.25$ lb/h

S CO

 \circ 1 ppmv CO in the CO₂ stream

45 300 m^3 _n/h o discharge from lock hoppers

o feedstock change 30 times/year/train

32.000 m^3 _n/feed change

resulting emission: \circ

(1 ppmv CO $*$ 45 300 m3n/h x 8760 h/yr + 32 000 m3n/feed change $*$ 30 starts/year/train * 2 trains * 1ppmy CO) / 22.4 kmol /m3n * 28 kg/kmole (molecular weight of CO) = 0.5 metric tons => 0.55 tons/y (1 short ton = 0.9072 metric ton)

 \Rightarrow 0.55 ton/year * 2000 lb/ton / 8760 h/y = 0.13 lb/h

There are two (2) B1 points (B1/1 & B1/2) operating at a time so the maximum emission rate from an individual B1 emission point is 0.55 tons/year $\div 2 = 0.275$ tons/year and 0.13 lb/h $\div 2 = 0.065$ lb/h

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- \circ 10 ppmv SOx in the CO₂ stream
- o discharge from lock hoppers 45 300 m^3 _n/h
- o feedstock change 30 times/year/train 32.000 m^3 /feed change
- o resulting emission:

(10 ppmv x 45300 m3n/h x 8760 h/yr + 32 000 m3n/feed change * 30 starts/year/train * 2 trains * 10 ppmv SOx)/ 22.4 kmol/m3n * 64 kg/kmole (molecular weight of SOx) = 11.4 metric tons => 12.56 tons/y $(1 \text{ short ton} = 0.9072 \text{ metric ton})$

 \Rightarrow 12.56 ton/year * 2000 lb/ton / 8760 h/y = 2.87 lb/h

There are two (2) B1 points (B1/1 & B1/2) operating at a time so the maximum emission rate from an individual B1 emission point is 12.56 tons/year $\div 2 = 6.28$ tons/year and 2.87 lb/h $\div 2 = 1.435$ lb/h

Emission Points B3: (failure of feed bin gas recycle)

Emergency release only (doesn't typically occur) => not considered

Emission Points A2 (start-up feeding):

- § fuel change and start-up each gasifier 30 times/y
- start-up time is approximately 3 hours \mathcal{S}
- gas discharge 30 000 m_{n}^3 /start-up S
- \mathcal{S} **CO**
	- o content 1 ppmv
	- o CO emission:

30 000 m3n/start-up x 30 starts/yr/train x 2 trains x 1 ppmv CO / 22.4 kmol/m3n x 28 kg/kmole (mol. Wt. CO) << 0.1 tons/y (1 short ton = 0.9072 metric ton)

 \Rightarrow 0.1 ton/y * 2000 lb/ton * 1/60 Starts/yr => << 3.33 lb/Start

 \Rightarrow 3.33 lb/Start / 3 hr/Start = 1.11lb/hr (maximum emission)

- Particulate mater
	- content 5 mg/m3 (industrial standard baghouse) \circ
	- o emission

30 000 m3n/start-up x 30 starts/yr/train x 2 trains x 5 mg/m3 PM << 0.1 tons/y (1 short ton = 0.9072 metric ton)

=> 0.1 ton/y * 2000 lb/ton * 1/60 Starts/y => << 3.33 lb/Start

 \Rightarrow 3.33 lb/Start / 3 hr/Start = 1.11 lb/hr (maximum emission)

- SOx S
	- content 10 ppmv \circ

o SOx emission:

30 000 m3n/start-up x 30 starts/yr/train x 2 trains x 10 ppmv SOx / 22.4 kmol/m3n x 64 kg/kmole (mol. Wt. SOx) = 0.06 tons/y (1 short ton = 0.9072 metric ton)

 \Rightarrow 0.06 ton/y * 2000 lb/ton * 1/60 Starts/y = 2 lb/Start

 \Rightarrow 2 lb/Start / 3 hr/Strart = 0.67 lb/hr (maximum emission)

3

Gasification (Unit 113)

In the PRENFLO™ gasifier the feed particles are partially oxidized with oxygen to be supplied from an air separation unit. Liquid slag is flowing down the cooled wall of the gasifier and falls through the quench zone into the slag pool. From there it is discharged via the slag removal.

The generated raw gas and fly ash are also conducted downwards and leave the gasifier though the opening at the bottom into the quench zone. Here the raw gas is quenched by a free down flow water film and additionally cooled and saturated by water spray nozzles. The remaining fly ash is removed from the gas by scrubbing. The dedusted gas is routed to the CO-Shift.

There is no emission source during the normal operation.

During the start-up the main burners are started one after the other at a reduced pressure, e.g. at 15 bar. The generated gas cannot be forwarded to the CO-Shift, as long as the required specification - composition and pressure - is not achieved. The gas generated by all burners would cause a quick pressure increase in the closed gas space. However, the pressure increase rate is limited. To reduce the pressure increase rate a part of the generated gas is sent to the flare (Emission Source B2), where the combustible gases are almost completely burned. However, a small part of CO remains unburned (typical 0.5%), H2S and COS are converted to SOx and NOx are formed. Also during the gasifier shutdown the pressure in the Gasification has to be reduced, therefore the gas is sent to the Flare. At decreasing pressure, the efficiency of sulfur removal decreases, therefore, a part of the sulfur is emitted via Flare as SOx.

Emission Source B2, gas from Gasification, discharge through Flare:

- Gas production during start-up of one Gasifier: 100 000 m_a^3 /start-up S
- Number of starts: 30 per year per gasifier \mathcal{S}
- S Start-up with PRB-coal or other low sulfur coal (used as start-up coal for SOx emission reduction)
	- o Sulfur emission
		- \$ H2S- content in raw gas based on 0.5%S in coal: 0.17% Ś SOx emission:

100 000 m3n/start-up x 30 starts x 2 gasifiers x 0.17% / 22.4 kmol/m3n x 64 kg/kmole (mol. Weight $SO2$) = 29 metric tons/ yr \Rightarrow 32 tons/y (1 short ton = 0.9072 metric ton)

 \Rightarrow 32 ton/y * 1/(30 hr/y) / 2 gasifiers * 2000 lb/ton = 1066 lb/hr There are two (2) B2 points (B2/1 & B2/2) operating at a time so the maximum emission rate from an individual B2 emission point is 32.0 tons/year $\div 2 = 16.0$ tons/year and 1066 lb/hr

- CO emission Ω
	- \$ CO- content in gas 60%
	- CO combustion rate in flare S. 99.5%
	- CO emission: \mathcal{S}

100 000 m3n/start-up x 30 starts x 2 gasifiers x 60% (CO concentration in syngas) x 0.5% (CO remaining after combustion) / 22.4 kmol/m3n x 28 (molecular wt. CO) = 22.5 metric tons/yr = > 24.8 tons tons/y (1 short ton = 0.9072 metric ton)

 \Rightarrow 24.8 ton/y * 1/(30 hr/y) / 2 gasifiers * 2000 lb/ton = 827 lb/h There are two (2) B2 points (B2/1 & B2/2) operating at a time so the maximum emission rate from an individual B2 emission point is 24.8 tons/year $\div 2 = 12.4$ tons/year and 827 lb/h

NO_x emission Ω

TGDS CTL Project, West Virginia

- S NO_x in flare offgas 250 ppmv (industrial standard)
- S Flare offgas during start-up: 296 000 m3n/start-up
- S NO_x emission

296 000 m3n/start-up x 30 starts x 2 gasifiers x 250 ppmv / 22.4 kmol/m3n x 46 kg/kmole(molecular wt. $NO2$) = 9.1 metric tons/yr => 10 tons/y (1 short ton = 0.9072 metric ton) \Rightarrow 10 ton/y * 1/(30 hr/y) / 2 gasifiers * 2000 lb/ton = 333 lb/h There are two (2) B2 points (B2/1 & B2/2) operating at a time so the maximum emission rate from an individual B2 emission point is 10.0 tons/year $\div 2 = 5.0$ tons/year and 333 lb/h

VOC emission \circ

- $\mathbb S$ Total flow rate to flare: 28516.8 kmol/h (one gasifier)
- S Methane (HC) concentration: 0.0025 mol%
- § Methane (HC) flow: 28516.8 kmol/h x 0.0025 mol% / 100 $mol\% = 0.713$ kmol/h
- S Heating value methane 890000 kJ/kmol
- S Methane heat flow: 0.713 kmol/h x 890000 kJ/koml / 3600 s/h / 1000 kJ/MJ = 0.176 MW
- $\text{\$}$ Conversions factor: 1 MMBTU/hr = 0.29308 MW
- $\text{\$}$ Methane heat flow: 0.172 MW / 0.29308 MW/MMBTU/hr = 0.601 MMBTU/hr
- S Emission factor Total Hydrocarbons acc. table 13.5-1 of AP 42: 0.14 lb/10^6BTU
- S Hourly HC emission: 0.601 MMBTU/hr x 0.14 lb/10^6BTU = 0.084 lb/hr
- S Yearly HC emission: 0.084 lb/hr x 30 starts/yr x 2 gasifier x 1 hr/starts / 2000 lb/tons = 0.0025 tons/yr
- $\frac{1}{2}$ There are two (2) B2 points (B2/1 & B2/2) operating at a time so the maximum emission rate from an individual B2

emission point is 0.0025 tons/year $\div 2 = 0.00125$ tons/year and 0.084 lb/h

ATTACHMENT 3

TO TASK ORDER 1

RESPONSE ON DEP QUESTIONS

TRANSGAS DEVELOPMENT SYSTEMS, LLC

CTL PROJECT

FUGITIVE EMISSIONS

PREPARED BY

UHDE

Fugitive emissions CO

Gasification

All available literature describes fugitive emission of VOCs for Refinery or organic chemical industry. The calculations of CO fugitive emissions is based on the current status of engineering work.

All calculation are based on the report EPA-453/R-95-017 (1995 Protocol for Equipment Leak Emission Estimates).

The fugitive emissions are calculated based on the EPA's average emission factor approach:

Emission = Factor x W_f x # components

The weight fraction (W_i) of CO in the Syngas is approx. 30%.

Used Control efficiencies for the calculation of the actual fugitive emissions:

Connectors 69% Pressure Relief Valves 90% **Others** 99%

Table 1 CO Leak Emission Estimate for gasification incl. scrubbing

11 Acc. Tab 5-2 (Reduction by LDAR)

22 Based on 8760 hr/yr

CO-Shift

All available literature describes fugitive emission of VOCs for Refinery or organic chemical industry. The calculations of CO fugitive emissions is based on the current status of engineering work.

All calculation are based on the report EPA-453/R-95-017 (1995 Protocol for Equipment Leak Emission Estimates).

The fugitive emissions are calculated based on the EPA's average emission factor approach:

Emission = Factor x Wf x # components

The weight fraction (Wf) of CO in the Syngas (incl. Reactors) is estimated with 36%. The weight fraction (Wf) of CO in the Syngas (from Reactors) is estimated with 29%.

Used Control efficiencies for the calculation of the actual fugitive emissions: Connectors 69% Pressure Relief Valves 90% **Others** 99%

Table 1: CO Leak Emission Estimate for CO-Shift up-stream & incl. reactors

Table 2: CO Leak Emission Estimate for CO-Shift downstream reactors

11 Acc. Tab 5-2 (Reduction by LDAR) 22 Based on 8760 hr/yr

Acid Gas Removal

All available literature describes fugitive emission of VOCs for Refinery or organic chemical industry. The calculations of CO fugitive emissions is based on the current status of engineering work.

All calculation are based on the report EPA-453/R-95-017 (1995 Protocol for Equipment Leak Emission Estimates).

The fugitive emissions are calculated based on the EPA's average emission factor approach:

Emission = Factor x W_f x # components

The weight fraction (W_t) of CO in the Syngas is estimated with 29%.

Used Control efficiencies for the calculation of the actual fugitive emissions:

Connectors 69% Pressure Relief Valves 90% Others 99%

Table 1 CO Leak Emission Estimate for Acid Gas Removal

11 Acc. Tab 5-2 (Reduction by LDAR) 22 Based on 8760 hr/yr

Methanol Synthesis

All available literature describes fugitive emission of VOCs for Refinery or organic chemical industry. The calculations of CO fugitive emissions is based on the current status of engineering work.

All calculation are based on the report EPA-453/R-95-017 (1995 Protocol for Equipment Leak Emission Estimates).

The fugitive emissions are calculated based on the EPA's average emission factor approach:

Emission = Factor x W_f x # components

The weight fraction (W_f) of CO in the Syngas is estimated with 86%.

Used Control efficiencies for the calculation of the actual fugitive emissions: Connectors 69% Pressure Relief Valves 90% Others 99%

Table 1 CO Leak Emission Estimate for Methanol Synthesis

11 Acc. Tab 5-2 (Reduction by LDAR)

22 Based on 8760 hr/yr

PSA

All available literature describes fugitive emission of VOCs for Refinery or organic chemical industry. The calculations of CO fugitive emissions is based on the current status of engineering work.

All calculation are based on the report EPA-453/R-95-017 (1995 Protocol for Equipment Leak Emission Estimates).

The fugitive emissions are calculated based on the EPA's average emission factor approach:

Emission = Factor x W_f x # components

The weight fraction (W_t) of CO in the Syngas is estimated with 86%.

Used Control efficiencies for the calculation of the actual fugitive emissions: Connectors 69% Pressure Relief Valves 90% **Others** 99%

Table 1 CO Leak Emission Estimate for PSA

11 Acc. Tab 5-2 (Reduction by LDAR) 22 Based on 8760 hr/yr

Fugitive Emissions H2S

Gasification

All available literature describes fugitive emission of VOCs for Refinery or organic chemical industry. The calculations of H2S fugitive emissions is based on the current status of engineering work.

All calculation are based on the report EPA-453/R-95-017 (1995 Protocol for Equipment Leak Emission Estimates).

The fugitive emissions are calculated based on the EPA's average emission factor approach:

Emission = Factor x Wf x # components

The weight fraction (Wf) of H2S in the Syngas is approx. 0.21%.

Used Control efficiencies for the calculation of the actual fugitive emissions: Connectors 69% Pressure Relief Valves 90% **Others** 99%

Table 1 H2S Leak Emission Estimate for gasification incl. scrubbing

11 Acc. Tab 5-2 (Reduction by LDAR)

22 Based on 8760 hr/yr

CO-Shift

All available literature describes fugitive emission of VOCs for Refinery or organic chemical industry. The calculations of H2S fugitive emissions is based on the current status of engineering work.

All calculation are based on the report EPA-453/R-95-017 (1995 Protocol for Equipment Leak Emission Estimates).

The fugitive emissions are calculated based on the EPA's average emission factor approach:

Emission = Factor x Wf x # components

The weight fraction (Wf) of H2S in the Syngas is approx. 0.32%.

Used Control efficiencies for the calculation of the actual fugitive emissions: Connectors 69% Pressure Relief Valves 90% **Others** 99%

Table 1 H2S Leak Emission Estimate for CO-Shift

11 Acc. Tab 5-2 (Reduction by LDAR)

22 Based on 8760 hr/yr

Sour Gas

Units: Slurry Stripping, Sour Water Stripping, Sulfur Recovery

All available literature describes fugitive emission of VOCs for Refinery or organic chemical industry. The calculations of H2S fugitive emissions is based on the current status of engineering work.

All calculation are based on the report EPA-453/R-95-017 (1995 Protocol for Equipment Leak Emission Estimates).

The fugitive emissions are calculated based on the EPA's average emission factor approach:

Emission = Factor x Wf x # components

The weight fraction (Wf) of H2S in the acid gas is approx. 2%.

Used Control efficiencies for the calculation of the actual fugitive emissions: Connectors 69% Pressure Relief Valves 90% **Others** 99%

Table 1 H2S Leak Emission Estimate for all unit with sour gas

11 Acc. Tab 5-2 (Reduction by LDAR)

22 Based on 8760 hr/yr

Acid Gas Removal

All available literature describes fugitive emission of VOCs for Refinery or organic chemical industry. The calculations of H2S fugitive emissions is based on the current status of engineering work.

All calculation are based on the report EPA-453/R-95-017 (1995 Protocol for Equipment Leak Emission Estimates).

The fugitive emissions are calculated based on the EPA's average emission factor approach:

Emission = Factor x W_f x # components

The weight fraction (W_i) of CO in the Syngas is estimated with 0.32%.

Used Control efficiencies for the calculation of the actual fugitive emissions:

Connectors 69% Pressure Relief Valves 90% **Others** 99%

Table 1 H2S Leak Emission Estimate for Acid Gas Removal

11 Acc. Tab 5-2 (Reduction by LDAR)

22 Based on 8760 hr/yr

Acid Gas in Sulfur Recovery Unit (one recovery train)

All available literature describes fugitive emission of VOCs for Refinery or organic chemical industry. The calculations of H2S fugitive emissions is based on the current status of engineering work.

All calculation are based on the report EPA-453/R-95-017 (1995 Protocol for Equipment Leak **Emission Estimates).**

The fugitive emissions are calculated based on the EPA's average emission factor approach:

Emission = Factor x Wf x # components

The weight fraction (Wf) of H2S in the acid gas is approx. 40%.

Used Control efficiencies for the calculation of the actual fugitive emissions: Connectors 69% Pressure Relief Valves 90% **Others** 99%

Table 1 H2S Leak Emission Estimate for sulfur recovery

11 Acc. Tab 5-2 (Reduction by LDAR) 22 Based on 8760 hr/yr

Fugitive emissions VOC

Acid gas removal

All available literature describes fugitive emission of VOCs for Refinery or organic chemical industry. The calculations of VOC fugitive emissions is based on the current status of engineering work.

All calculation are based on the report EPA-453/R-95-017 (1995 Protocol for Equipment Leak Emission Estimates).

The fugitive emissions are calculated based on the EPA's average emission factor approach:

Emission = Factor x W_f x # components

The weight fraction (W_t) of Methanol in liquid steams is estimated with 100%. The weight fraction (W_i) of Methanol in gaseous steams is estimated with 100%.

Used Control efficiencies for the calculation of the actual fugitive emissions:

Table 1 VOC (Methanol) Leak Emission Estimate for acid gas removal

11 Acc. Tab 5-2 (Reduction by LDAR)

22 Based on 8760 hr/yr

33 Number in parenthesis represents total number; second number (not in parenthesis) remaining number of connectors not welded; weld control efficiency 100% acc. Table 5-1 LL Light Liquid

MeOH Synthesis

All available literature describes fugitive emission of VOCs for Refinery or organic chemical industry. The calculations of VOC fugitive emissions is based on the current status of engineering work.

All calculation are based on the report EPA-453/R-95-017 (1995 Protocol for Equipment Leak Emission Estimates).

The fugitive emissions are calculated based on the EPA's average emission factor approach:

Emission = Factor x W_f x # components

The weight fraction (W_t) of Methanol in liquid steams is estimated with 96%. The weight fraction (W_i) of Methanol in gaseous steams is estimated with 15%.

Used Control efficiencies for the calculation of the actual fugitive emissions: Connectors 69% Pressure Relief Valves 90% **Others** 99%

Table 1 VOC (Methanol) Leak Emission Estimate for Methanol Synthesis

11 Acc. Tab 5-2 (Reduction by LDAR)

22 Based on 8760 hr/yr

33 Number in parenthesis represents total number; second number (not in parenthesis) remaining number of connectors not welded; weld control efficiency 100% acc. Table 5-1 LL Light Liquid

MTG

All available literature describes fugitive emission of VOCs for Refinery or organic chemical industry. The calculations of VOC fugitive emissions is based on the current status of engineering work.

All calculation are based on the report EPA-453/R-95-017 (1995 Protocol for Equipment Leak Emission Estimates).

The fugitive emissions are calculated based on the EPA's average emission factor approach:

Emission = Factor x W_f x # components

The weight fraction (W_t) of Methanol in liquid steams is estimated with 96%. The weight fraction (W_t) of Methanol in gaseous steams is estimated with 96%.

Used Control efficiencies for the calculation of the actual fugitive emissions: Connectors 69% Pressure Relief Valves 90% **Others** 99%

Table 1 VOC (Methanol) Leak Emission Estimate for MTG

1 Acc. Tab 5-2 (Reduction by LDAR)

2 Based on 8760 hr/yr

3 Number in parenthesis represents total number; second number (not in parenthesis) remaining number of connectors not welded; weld control efficiency 100% acc. Table 5-1 LL Light Liquid

Fugitive emissions VOC (Methanol free)

All available literature describes fugitive emission of VOCs for Refinery or organic chemical industry. So all calculations of VOC fugitive emissions will be in the accuracy of a rough estimate.

All calculation are based on the report EPA-453/R-95-017 (1995 Protocol for Equipment Leak Emission Estimates).

The fugitive emissions are calculated based on the EPA's average emission factor approach:

Emission = Factor x W_f x # components

The weight fraction (W_i) of VOC in liquid steams is estimated with 100%. The weight fraction (W_i) of VOC in gaseous steams is estimated with 50%.

Used Control efficiencies for the calculation of the actual fugitive emissions:

Connectors 69%

Pressure Relief Valves 90% 99%

Others

Table 2 VOC Leak Emission Estimate for MTG

11 Acc. Tab 5-2 (Reduction by LDAR)

22 Based on 8760 hr/yr

LL Light Liquid

Table 3 Total VOC Leak Emission Estimate for MTG

ATTACHMENT O

MONITORING, RECORDKEEPING, REPORTING,
AND TESTING PLANS

 \mathcal{L}

ATTACHMENT O

MONITORING, RECORDKEEPING, REPORTING, AND TESTING PLANS

Transgas Development Systems, LLC will work with DAQ to identify and address Monitoring, Recordkeeping, Reporting, and Testing Plans. See Section L for proposed Monitoring, Recordkeeping, Reporting, and Testing Plans.

ATTACHMENT P

PUBLIC NOTICE

EXAMPLE LEGAL ADVERTISEMENT

Publication of a proper Class I legal advertisement is a requirement of the application process. In the event the applicant's legal advertisement fails to follow the requirements of 45CSR 13 (45-13-8) or the requirements of Chapter 59, Article 3, of the West Virginia Code, the application will be considered incomplete and no further review of the application will occur.

The applicant, utilizing the format for the Class I legal advertisement appearing below, shall cause such legal advertisement to appear a minimum of one (1) day in the newspaper most commonly read in the area where the facility exists or will be constructed. The notice must be published no earlier than five (5) working days of receipt by this office of your application. The original affidavit of publication must be received by this office no later than the last day of the public comment period.

The advertisement shall contain, at a minimum, the name of the applicant, the type and location of the source, the type and amount of air pollutants that will be discharged, the nature of the permit being sought, the proposed start-up date for the source and a contact telephone number for more information.

The location of the source should be as specific as possible starting with: 1.) the street address of the source: 2.) the nearest street or road: 3.) the nearest town or unincorporated area, and 4.) the county.

Types and amounts of pollutants discharged must include all regulated pollutants (PM, PM_{10} , VOC, SO₂, Xylene, etc.) and their potential to emit or the permit level being sought in units of tons per year (including fugitive emissions).

In the event the 30th day is a Saturday, Sunday, or legal holiday, the comment period will be extended until 5:00 p.m. on the following regularly scheduled business day.

AIR QUALITY PERMIT NOTICE **Notice of Application**

Notice is given that TransGas Development Systems, LLC has applied to the West Virginia Department of Environmental Protection, Division of Air Quality, for a Regulation 13 Construction Permit for a coal to gasoline plant to be located off of West Virginia State Route 52 near Wharncliffe, in Mingo County, West Virginia.

The applicant estimates the potential to discharge the following Regulated Air Pollutants will be: NOx, of 50 tons per year (tpy), SOx of 84 tpy, CO of 90 tpy, VOC of 40.52 tpy, HAPS of 7.47, PM of 95.76 tpy of which 29.81 tpy is fugitive, and PM10 of 67.19 tpy of which 10.13 tpy is fugitive.

Startup of operation is planned to begin on or about the 1st day of May, 2013. Written comments will be received by the West Virginia Department of Environmental Protection, Division of Air Quality, 601 57th Street, SE, Charleston, WV 25304, for at least 30 calendar days from the date of publication of this notice.

Any questions regarding this permit application should be directed to the DAQ at (304) 926-0499, extension 1227, during normal business hours.

Dated this the (Insert Date) day of December, 2008.

TransGas Development Systems, LLC By: **Adam Victor** President 630 First Avenue, Suite 30G New York, New York 10013-3799

ATTACHMENT Q

BUSINESS CONFIDENTIAL CLAIMS

Precautionary Notice - Claims of Confidentiality

The person submitting this information may assert that some or all of the information submitted is entitled to confidential treatment as provided by West Virginia Legislative Rule 45CSR31, entitled "Confidential Information." Information covered by such a claim will be disclosed by the Division of Air Quality (DAQ) only to the extent, and by means of the procedures, set forth in 45CSR31. Please contact the West Virginia Secretary of State's Office at 304/558-6000 to obtain a copy of 45CSR31 in order to ensure that all required procedures are followed.

Information concerning the "types and amounts of air pollutants discharged," as that term is defined in WVCSR §45-31-2.4, shall not be claimed as confidential.

Any claim of confidentiality shall be made in accordance with the requirements of 45CSR31 and must accompany the information at the time it is submitted to the DAQ. If no claim of confidentiality is made at the time of submission or is not made in accordance with the requirements of 45CSR31, the DAQ may make the information available to the public without further notice.

Included below are procedures to be followed in submitting information claimed as confidential. This information is intended to assist a person with claiming confidential information and is not meant to relieve a person of his/her obligation to review the provisions of 45CSR31 and to comply with such rule. The procedures are as follows:

- Indicate clearly the items of information claimed confidential by marking each page 1. with the term "Claimed Confidential," with the date of such claim of confidentiality. With the exception of documents of a size greater than 81/2" x 14", information claimed confidential must be submitted on colored paper.
- Include a cover document which justifies the claim of confidentiality in accordance $2.$ with the specific criteria under WVCSR §45-31-4.1. A sample cover document is attached for your information and use. The cover document will be available for public disclosure and must include the following information:
	- (a) The identity of the person making the submission of information claimed confidential:
	- The reason for the submission of information: (b)
	- (c) The name, an address in the State of West Virginia and telephone number of the designee who shall be contacted in accordance with 45CSR31;
	- (d) Identification of each segment of information within each page that is submitted as confidential and the justification for each segment claimed confidential, including the criteria under WVCSR §45-31-4.1;
- (e) The period of time for which confidential treatment is desired (e.g., until a certain date, until the occurrence of a specified event or permanently); and,
- (f) Signature of a responsible official or an authorized representative of such person.
- $3.$ At the same time as the information claimed confidential is submitted to the DAO on colored paper, a complete set of the information, including the cover document previously required under paragraph 2, must be submitted on white paper with the information claimed to be confidential blacked or whited out and the words "Redacted Copy - Claim of Confidentiality" marked clearly on each such page, so that the information is suitable for public disclosure. In the case of drawings and blueprints, mark each page with the words "Redacted Copy — Claim of Confidentiality," include the title or legend of the drawing, and black or white out the information claimed confidential. The redacted page may be $8\frac{1}{2}$ " x 11" in size.
- 4. In the case of a permit application or supplemental information to an application, DAQ requires an applicant to submit three (3) copies of the application. Of those three (3) copies, one (1) must be a complete set of the application containing the information claimed confidential on colored paper and two (2) must be redacted copies. The DAQ reserves the right, however, to request additional copies of the information containing the confidential material.

Attachment
Attachment Q Business Confidential Claim

This form contains each of the required elements for the cover document required under 45CSR31. The person submitting this form may wish to attach an additional page(s) to provide adequate justification under the ARationale@ section of the form.

Reason for Submittal of Confidential Information: Allow review of information pertaining to DAQ issuing a Regulation 13 Permit Application.

NOTE: Must be signed and dated in BLUE INK.

Provide justification that the criteria set forth in § 45CSR31-4.1.a - e have been met.

4.1.a. The claim of confidentiality has not expired by its terms, nor been waived or withdrawn;

The confidentially agreements do not have an expiration date due to the nature of the information contained in the application.

4.1.b. The person asserting the claim of confidentiality has satisfactorily shown that it has taken reasonable measures to protect the confidentiality of the information, and that it intends to continue to take such measures:

The information contained within the application is fully protected under non-disclosure and confidentiality agreements by all parities involved in the original development of the processes, the design of the facility, and the permit application process.

4.1.c. The information claimed confidential is not, and has not been, reasonably obtainable without the person's consent by other persons (other than governmental bodies) by use of legitimate means (other than discovery based on a showing of special need in a judicial or quasi-judicial proceeding);

The information available herein is not available and is not to be made available to outside parties.

4.1.d. No statute specifically requires disclosure of the information; and

TransGas Development Systems, LLC believes there are no statutes that require disclosure of the information.

4.1.e. Either--

4.1.e.1. The person has satisfactorily shown that disclosure of the information is likely to cause substantial harm to the business's competitive position; or

This is a unique facility with many parties involved in preparing and providing information on the systems. Release of this information could cause substantial harm to TransGas Development Systems, LLC competitive position in the coal to gasoline market.

4.1.e.2. The information is voluntarily submitted information, and its disclosure would likely to impair the State's ability to obtain necessary information in the future.

The State should not disclose this information to anyone.

APPENDIX

SUPPLEMENTAL INFORMATION

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TRANSGAS DEVELOPMENT SYSTEMS, LLC

CTL PROJECT

Process Description

CTL Plant

PREPARED BY

UHDE

Rev. 00 September 2008

- 1) All information and/ or data provided by Uhde shall not be deemed as a commercial or technical proposal and in particular do not contain any definite "not-to-exceed prices".
- (2) All information and/ or data have to be reviewed on the basis of more sufficient and detailed initial data to be provided by Transgas Development Systems, LLC and in consequence the information and/ or data provided by Uhde may differ considerably from such reviewed data.
- (3) Uhde, however, makes no representations or warranties, express or implied, as to the quality, accuracy and completeness of the information and/ or data provided hereunder. The receiving party accepts all risk of use of and reliance on the provided information and/ or data.
- (4) Uhde shall in no event be liable to the receiving party for loss of profit, loss of assets, loss of business opportunities, loss of use, loss of production, or any financial or economic loss, or for any indirect, incidental, special or consequential loss or damages, irrespective whether such liability should be based on contract, warranty, tort (including negligence), strict liability or otherwise.

Uhde

TGDS CTL Project, West Virginia

TABLE OF CONTENTS

CTL Plant, Process Description

Uhde

TGDS CTL Project, West Virginia

Coal Preparation and Gasification 1

1.1 **Introduction**

The PRENFLO™ Direct Quench (PDQ) process is an innovative coal gasification process and an alternative design of the proven PRENFLO™ technology, which has been successfully installed in its PSG (PRENFLO™ Steam Generation) version in the world's largest IGCC in Puertollano, Spain.

The PDQ process has been developed specifically for the hydrogen and chemical applications.

The "slagging" gasifier set-up features a cylindrical, cooled (membrane-wall) reactor cage provided with protrusions (muffles) for coal burners arranged horizontally in an opposed-firing configuration. By installing the burners with a small angle to the radial, a swirling motion is imposed upon the gas flow pattern inside the reactor cage, which will impose a centrifugal movement of the liquid slag formed during gasification towards the cylindrical cage wall. The syngas produced during gasification is led down to the slagbath surface, diverted upwards and quenched with water (direct quench).

The liquid slag flows downward along the vertical cylindrical wall through a (slag) tap in the conical bottom of the cage into a water bath (slagbath) where it solidifies and scatters into small granules. The arrangement thereby ensures a simple and effective segregation between the syngas (product) on the one hand and the bulk of the slag/ash on the other. Fly ash will partially be washed out by quench water, partially be entrained with the syngas.

The membrane-wall reactor cage is installed inside a vessel providing the pressurised containment. The membrane-wall tubes are cooled by raising steam.

The gasification consists of several parallel trains. Only one train is described exemplary for all.

The gasification comprises a number of subsystems, which will be described individually:

1.2 **Coal Preparation (Unit 111)**

The Coal Milling and Drying Unit consists of several identical process trains. During normal operation one train provides the feed for one gasification train.

The system configuration has been selected to satisfy the following two major requirements:

- safe operation of dried, pulverized coal, i.e. an inert atmosphere \bullet
- minimize energy consumption, i.e. use a recycle of conveying and drying gas. \bullet

Coal from the coal bunker is transported by weighing belts to the mill. Each mill is equipped with its own bunker.

The bunkers are vented with the bunker vent fans through the bunker filters

The coal feeder controls the mill capacity. The total feed stream is gravity-dropped straight into the mill

In the coal (roller) mills the raw coal is milled and dried under a slight under-pressure and under inert conditions (low oxygen concentration). Potential hazards of self-

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The inert gas loaded with fine particles is sent to the bag house filter, a fabric type filter, for separation.

The pulverized coal is transported from the bag house filter to the pulverized coal storage vessel (Unit 121/2) using transportation screws and rotary feeders.

1.3 **Coal Dust Feeding (Unit 112)**

The coal from the coal preparation enters the coal dust bunker via the cyclone filter on top of the coal dust bunker, from where it is fed to the Plant. The coal flows by gravity to the lock hoppers placed under the coal dust bunker. Each lock hopper is connected to one outlet cone of the coal dust bunker.

Once a lock hopper is filled with coal dust, the lock hopper is disconnected and pressurized from atmospheric pressure to about 1 to 2 bar above the coal feed bin working pressure. This is done by connecting the lock hopper to an inert gas system (N2 or CO2). The selection of inert gas depends on the field of application of the product gas (Syngas) of the gasification process.

After opening the valve at the outlet of the lock hopper the pressurized coal dust from one or more lock hopper is continuously supplied via a dense phase conveying system to the coal feed bin. The transport medium of the conveying system is also an inert gas system (N2 or CO2).

working pressure of the coal feed bin is about 5 bar higher than the gasifier pressure. Each Vessel of the coal preparation and coal dust feeding is connected to the cyclone filter placed above the coal dust bunker.

Each lock hopper is additionally equipped with a coarse filter.

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Gasification (Unit 113) 1.4

The coal dust enters the via the burners into the PRENFLOTM gasifier. The coal particles are partially oxidized with oxygen to be supplied from an air separation unit (ASU). Liquid slag is flowing down the cooled wall of the gasifier and falls through the quench zone into the slag pool. From there it is discharged via the slag removal. The generated raw gas and fly ash are also conducted downwards and leave the gasifier though the opening at the bottom into the first quench zone. Here the raw gas is quenched by a free down flow water film and additionally cooled and saturated by water spray nozzles further down in the quench zone.

Scrubbed fly ash from the raw gas is collected in the slag collecting vessel and in the surrounding water bath. The water overflow of the collecting vessel flows together with fly ash particles into the surrounding water bath. The surplus water exits the water bath and is pumped by the Quench Water Circulating Pump to the quench nozzles. Additionally a continuous water stream is withdrawn from the water bath to the Slurry Recovery System to avoid an accumulation of solids.

1.5 Slag Removal (Unit 114)

The granulated slag is discharged from the slag collecting vessel to the slag lock

A slag extractor is placed under each of the lock hoppers for receiving the slag during emptying of the lock hopper. The coarse material is removed from the slag extractor. The remaining water, which contains the fines-a combination of slag fines and unburned coal-is pumped to the slurry recovery.

1.6 **Slag Fines Removal (Unit 115)**

The key elements of the slurry recovery are the slurry flash vessels with integrated heat exchangers. The slurry flash vessel I is fed by the discharged slurry coming from the gasifier water bath.

The slurry then enters the Slurry Flash Vessel II at the bottom . The filtrate preheaters I and II, which are integrated in the slurry flash vessels are fed in counter current flow with process water coming from the process water tank, which is preheated with flash gas from the depressurized slurry.

The remaining slurry and the slurry coming from the slag extractor enter the slurry flash vessel 3.

The flash gas of the three vessels are combined and sent to the sulfur recovery unit. The flashed slurry is transferred by the slurry pump to the clarifier in the slurry filtration system.

1.7 **Scrubbing System (Unit 116)**

The raw gas, withdrawn from the gasifier is sent to the scrubbing system consisting of the venturi scrubber and the downstream scrubber. In the scrubbing system the raw gas the remaining fly ash particles as well undesired trace components are removed. The scrubbed gas is handed over to further gas treating.

The scrubber system has a continuous water circulation. The water accumulating in the bottoms of the scrubber is circulated via the scrubber circulating pump and the LP-boiler back to the scrubber and the venturi scrubber.

1.8 **Water Pretreatment (Unit 117)**

A flocculent is added to the flashed slurry from the slurry recovery before it is fed to the clarifier. In the clarifier the flashed slurry is separated into a solid rich stream and clear overflow water stream. The overflow water is collected as process water in the process water tank

To avoid an accumulation of chlorides in the water system a continuous blow down of process water is send to the waste water treatment.

The solid rich underflow of the clarifier is transported via the clarifier bottoms pump to the sludge storage tank. This tank supplies the belt filter press unit continuously with sludge via the sludge pump. The processed filter cake is sent to the coal yard. The filtrate of the belt filter press is sent to the filtrate tank. From the tank the filtrate pumped via the filtrate pump back to the clarifier and also for scrubbing purpose to the belt filter.

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Gas Treatment $\mathbf{2}$

2.1 **CO-Shift System (Unit 233)**

$2.1.1$ **General**

The raw syngas coming from the wet scrubbing (Unit 116) consists mainly of hydrogen $(H₂)$ and carbon monoxide (CO). The amount of hydrogen in the raw syngas is too low to satisfy the required H2/CO ratio for the downstream Methanol Synthesis Unit.

Therefore additional hydrogen has to be produced by a CO-shift process. This is achieved by processing a major part of the raw syngas in the CO-Shift Unit. As the feedgas contains sulphur components, the catalyst to be used has to be sulphur tolerant and is referred to as sour shift catalyst.

The carbon monoxide (CO) is converted catalytically with water (H_2O) into hydrogen $(H₂)$ and carbon dioxide (CO₂). In addition the applied catalyst reduces hydrogen cyanide (HCN) and carbonyl sulphide (COS) in presence of water to ammonia (NH3) and carbon monoxide (CO) and to hydrogen sulphide (H2S) and carbon dioxide (CO2) near to equilibrium levels.

Due to the direct quench configuration in the Gasification Unit 116 the water content of the raw syngas is high enough to fulfil the water vapour needs of the CO shift reaction. The steam to dry gas ratio is sufficient to provide the required steam for the reaction as well as to moderate the temperature increase in the shift reactor.

$2.1.2$ **Process Flow**

The raw syngas coming from the wet scrubbing (Unit 116) at a temperature of approx. 218 °C is fed via Rawgas Superheater which heat up the rawgas to reaction temperatur to the Sour Shift Reactor. In order to control the H2/CO ratio at Battery Limits of the CO Shift Unit a bypass is installed across the one CO Shift Reactor 2330R001. The inlet temperature to the shift reactor is controlled by a by-pass across the Rawgas Superheater. Only one shift reactor is foreseen. This insures that a major part of raw syngas is routed across the CO shift catalyst and simultaneously treated for COS / HCN hydrolysis. This reduces the load of these components to the downstream H2S/CO2 Removal Unit 2350.

The considerable heat of reaction of this process is being used for the production of saturated HP-steam in the downstream Heat Recovery (e.g. Shift Gas Cooler, saturated LP-steam in Syngas Cooler 1 and for warming up of boiler feed water in Syngas Cooler 2).

The steam contained in the raw syngas is only partially consumed by the conversion reaction. The remaining part is condensed and separated from the gas during cooling of the syngas. The formed process condensate is separated in Condensate Separator I and directly recycled back to the gasifcation unit via Condensate Pump I. Final cooling of the shifted gas is carried out by air cooling in the Syngas Air Cooler and by water cooling.

The process condensate from this further cooling is separated in Condensate Separators II and routed to the Sour Water Stripping Unit 328.

The cooled gas gas is pressurised further by Syngas Compressor before routing to the H2S/CO2-Removal Unit 235.

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$CO₂$ / H₂S Removal (Unit 235) 2.2

$2.2.1$ General

The Rectisol Wash is a physical wash system with methanol as preferable solvent. It removes the acid gases CO2, H2S, COS and HCN from a feed gas from a gasification unit. The Rectisol Wash Unit consists of a methanol wash section, recycle gas compression, CO2 product recovery section, hot regeneration section and the methanol/water separation. In addition a small recycle feed gas from the sulphur recovery unit is treated. The process description refers to the corresponding simplified PFD

$2.2.2$ **Process Flow**

Feed Gas Precooling (not shown)

The feed gas is supplied to B.L. of the Rectisol plant from the upstream CO Shift unit. After mixing of the internal recycle gas and injection of methanol, the feed gas is cooled down against cold product gases. The mixture of the cooled down feed gas is routed to the Absorber Column (methanol wash column).

The condensed loaded methanol/water is separated in a upstream knock out drum and fed to the MeOH $/$ H₂O Separation.

Absorber Column (methanol wash column)

In the lower section of the wash column HCN is removed by the cold methanol. H₂S and COS is washed out in the middle section, the $CO₂$ is removed in the upper section. The heat of solution in the $CO₂$ section is covered partially by warming up of the methanol, partially by a cooling stage (cooling medium evaporating refrigerant). As the solubility of $CO₂$ in methanol is less than the solubility of H₂S and COS, the

methanol flow in the $CO₂$ removal section is greater than in the H2S removal section. The methanol surplus from the $CO₂$ removal section is taken off from the upper chimney tray of the column, this methanol is only loaded with $CO₂$. The stream, which is drawn off from the sulfur removal section (lower chimney tray), is additionally loaded with COS and H_2S . For the HCN removal section only a very small split stream of the total solvent is required.

The small solvent stream also loaded with HCN is drawn off the bottom of the column. One part of this methanol is pumped as methanol injection into the feed gas. The other part is fed as HCN/methanol/water mixture into the HCN Separation Column after being warming up.

Solvent Flash I (Recovery of CO and H₂)

After subcooling the two main loaded methanol streams from the side draws of the Absorber Column are expanded to an intermediate pressure in order to recover dissolved hydrogen and to limit the CO content in the CO2 product and in the tailgas. Option 1: The flashed gases are recompressed and recycled into the feed gas to the Rectisol wash unit to recover CO $+H_2$.

Option 1: If there is demand on heating gas for the coal preparation (milling) this gas can also used as fuel gas.

Solvent Flash II (CO2 Production)

CO2 is recovered by warming up and expansion of loaded methanol streams into the CO2 production column. H2S is removed from the CO2 product in this column by means of a part of the sulphur free methanol from, routed to the top. The overhead of the CO2 column is sent to the $CO₂$ Wash Column

The flashed $CO₂$ warmed up in is passed through the $CO₂$ Wash Column for methanol removal. Wash medium is demineralised water. The wash water from the column bottom loaded with methanol is pumped and warmed-up into the MeOH/ H_2O Rectifier (methanol/water separation) for methanol recovery.

Hydrogenated tailgas from the sulfur recovery can also be fed to this column

$CO₂$ Stripper

The flashed solvent from the Flash Solvent II column contains to much $CO₂$ for the sourgas. In this column CO2 is stripped off the methanol by means of LP-Nitrogen in order to revover the CO₂ and to increase the H2S concentration in the H2S fraction. The solvent washed $CO₂$ top fraction is routed to the Solvent Flash II column and the bottom product which contans the H_2S is pumped to the Hot Regeneration Column.

Hot Regeneration Column

The H₂S enriched solvent is regenerated in this column. All dissolved acid gases are stripped off by means of methanol vapour, generated in a reboiler, heated by steam. The lean methanol from the bottom of the column is pumped back to the Absorber (wash) Column. Before reuse the lean solvent is cooled down. Methanol vapour is condensed from the H2S fraction leaving the top. The column overhead product is H₂S enriched sourgas which is routed to the sulfur recovery unit together with the overhead fraction of the HCN Separation Column.

HCN Separation Column

The HCN/methanol/water mixture from bottom of Absorber Column is fed to the HCN separation column, where HCN is removed by stripping. The column is heated by means of steam in a reboiler. After passing the condenser in the top of the column for methanol condensation the remaining uncondensed vapour (enriched in HCN) is mixed to the H2S fraction from the hot regeneration column.

Methanol/Water Separation

The bottom product from the HCN Separation Column and the wash water from the bottom of the CO₂ Wash Column together with condensate from the feed knock out drum is separated in the methanol/water separation column by distillation to methanol (top) and waste water (bottom). The column is heated by steam, as reflux lean

methanol from the hot regeneration column is used. The methanol vapour from the top supports the stripping in the Hot Regeneration Column, the waste water is routed to B.L. after being cooled down. This water stream contains impurities contained in the feed gas.

CO₂ Wash Column

 $CO₂$, warmed up in, is passed through the $CO₂$ Wash Column for methanol removal. The methanol free $CO₂$ is compressed cooled down and used as sluicing gas in the coal gasification. Wash medium is demineralised water. The wash water from the column bottom loaded with methanol is pumped and warmed-up into the Methanol/Water Separation for methanol recovery.

2.3 **Sour Water Stripping (Unit 238)**

$2.3.1$ General

The Sour Water Stripping Unit 2380 treats process condensate from the CO Shift Unit 233 and from the tailgas treating section of Sulphur Recovery Unit 241 for removal of H2S, CO2, NH3 and HCN. The removal of this components is accomplished by stripping with water vapour in a stripper column equipped with packings.

$2.3.2$ **Process Flow**

In SWS Feed Vessel sour water from the CO Shift Unit 233 is flashed and collected in order to provide a stable continuous feed to the Sour Water Stripper. The vessel is equpped with a balance line between the vessle and the stripper top section.

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The sour water is pumped from SWS Feed Vessel to the top of Sour Water Stripper via the Feed/Effluent Exchanger. In the Feed/Effluent Exchanger the sour water is heated by exchanging heat with the stripped water leaving Sour Water Stripper at the bottom.

The Sour Water Stripper can be equpped with structured packings or with Pall Rings. The packing is divided into three beds of which the top bed is used as a washing section for the sour water reflux coming from the SWS Reflux vessel. In the other packed bed sections of Sour Water Stripper, the sour water is contacted in counter current flow with steam from SWS Reboiler. The feed distributor of the Sour Water Stripper is located between the first and the second packed bed section.

LP steam is used for operation of SWS Reboiler. The major part of the vapours from the top of Sour Water Stripper is condensed in SWS Overhead Condenser. The outlet temperature of this air cooler is approx. 80°C to prevent salt formation. The remaining sour gas is separated from the formed condensate in SWS Reflux Vessel and sent to the Sulphur Recovery Unit 241.

The condensate is returned to Sour Water Stripper as reflux by SWS Reflux Pump.

The stripped water is reused and sent back to the Scrubbing Unit 116 via the Feed/Effluent Exchanger.

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Mercury Removal (Unit 239) 2.4

$2.4.1$ General

The Mercury Removal Unit 239 is located downstream of the CO2/H2S-Removal Unit 235 which is based on the Rectisol process. Although the Rectisol unit removes the major part of mercury from the syngas due to the very low operating temperatures during absorption an additional adsorbent vessel is foreseen downstream of the CO2/H2S removal in order to assure a complete separation of an mercury that might be left over.

$2.4.2$ **Process Flow**

The Mercury Removal Unit consists of only one piece of equipment - the Mercury Adsorber. The vessel is equipped with a bed of impregnated activated carbon which is specially designed for mercury adsorption. The treated, completely dry syngas enters the vessel at the top and leaves the vessel as purified syngas at the bottom.

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Sulphur Recovery (Unit 241) 2.5

$2.5.1$ **General**

The Sulphur Recovery Unit 241 processes H2S, CO2 and NH3 containing sour gas from stripping of process condensates in Sour Water Stripping Unit 238 and H2S, COS, CO2 and HCN containing acid gas from the CO2/H2S Removal Unit 235. NH3 and HCN are decomposed to less dangerous components, sulphur containing components are converted to elemental sulphur and CO2 passes this unit as inert component.

The produced liquid sulphur is collected, degassed and solidified. The residual tailgas of the sulphur recovery process is hydrogenated and recycled back to the CO2/H2S Removal Unit 2350.

The H2S contained in the combined feeds is partially combusted with oxygen coming from Unit 478, producing the intermediate product sulphur dioxide according to the reaction:

> \Leftrightarrow SO₂ + H₂O $H_2S + 3/2 O_2$ \leftrightarrow 3/x S_x + 2 H₂O $2 H_2S + SO_2$

Both reactions can be summarised to:

 $H_2S + 1/2 O_2$ \leftrightarrow $1/x S_x + H_2O$

Water (H2O) is a side product of all of these reactions. They are executed in one thermal and two catalytic steps in series. In all steps the gaseous sulphur species are in thermal equilibrium with liquid sulphur:

 $S_x = 12 S_2 \leftrightarrow 4 S_6 \leftrightarrow 3 S_8 \leftrightarrow S_{Liouid}$

The vapours from Unit 115 and the vapours from Unit 238 contain ammonia. This ammonia is decomposed to nitrogen and hydrogen in this thermal step The produced liquid sulphur is routed to the collecting and degassing system. Because the catalytic conversion is not complete, the tail gas at the outlet of the last pass of the sulphur condenser contains un-reacted H_2S and SO_2 as well as gaseous sulphur species acc. to their vapour pressures. Therefore, this tail gas is treated for hydrogenation of this sulphur components, pressurised and sent back to the $CO₂$ /H₂S Removal Unit.

Process Flow

Sulphur Recovery

The NH3 containing sour gases from Unit 115 and Unit 238 are combined and introduced to the unit separately from the acid gas produced by Unit 235.

Pressure control valves in the feed lines control the pressure in the tops of the columns, from which the feeds originate.

The ammonia containing sour gas and a portion of the acid, ratio controlled by the flow of feeds, are fed together with the required oxygen to the central part of the Claus Furnace Burner, while the other portion of acid gas is routed to the outer part. This burner is integrated to the combustion chamber of Claus Furnace.

Due to the low H2S content of the combined sour gas and acid gas feeds, pure oxygen is needed to reach the required temperature for ammonia destruction in the Claus furnace.

The required oxygen is supplied directly from Unit 478. A feed forward control is setting oxygen to acid gas ratio. This feed back control ensures the correct stoichiometric conditions for the partial combustion reaction and with this, an optimum sulphur recovery.

The heat of reaction of the partial combustion of H2S is used primarily for production of saturated MP Steam in the Claus Furnace Waste Heat Boiler. The produced steam is released pressure controlled, whereas the required boiler feed water flows level controlled to the steam drum.

Further cooling of the process gas is performed in the first pass of Sulphur Condenser where elemental sulphur vapours are condensed and separated from the gas by an integrated sulphur separator. The transferred heat is used for production of LP-Steam in the outer shell of this heat exchanger.

The process gas is reheated in Reheater I and routed to the Claus Reactor I. The required inlet temperature is controlled by steam flow to the First Reheater. In the first converter the formation of sulphur is continued resulting in an increase of gas temperature. The formed sulphur vapour is condensed again in the second pass of Sulphur Condenser. After separation of the liquid sulphur the process gas is reheated again in Reheater II, routed across Claus Reactor II for further reaction and cooled in third pass of Sulphur Condenser for sulphur condensation and separation. The LP steam produced in Sulphur Condenser is released pressure controlled to the header. A part of the LP steam is consumed internally in the Sulphur Recovery Unit for tracing of sulphur lines and other various heating purposes.

The liquid sulphur separated from the process gas in Sulphur Condenser flows by gravity to the first compartment of the Sulphur Pit via sulphur locks.

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Tailgas Treating and Compression

As the reaction of H2S and COS to elemental sulphur is not complete the tailgas from Sulphur Condenser III has to be treated for hydrogenation before it can be recycled to the CO2/H2S Removal Unit 235.

The tailgas is mixed with pure hydrogen to provide the required amount of hydrogen for the catalytic reaction. The hydrogen flow is controlled by analysis of the H2 content in the reactor effluent gas.

The tailgas is heated in the Feed/Effluent Exchanger and subsequently in the MPsteam heated Tailgas Heater to the required reaction temperature.

The heated gas passes the Hydrogenation Reactor andis cooled in the Feed / Effluent Exchanger. The tailgas is routed to the Tail Gas Scrubber to wash out any traces of unconverted sulphur and for further cooling.

The washing and cooling effect is performed by recycling the main portion of the scrubber bottom product through Recycle Water Cooler back to the top of the scrubber. The heat is removed by means of cooling water.

The formed condensate is drawn off from the cooling cycle level controlled at the bottom or Tail Gas Scrubber and sent to the Gas Condensate Drum.

The treated and cooled tailgas from the top of Tail Gas Scrubber is compressed to approx. 3 bara in the Recycle Gas Compressor and sent cooled by a water cooler to the regeneration section of the CO2/H2S Removal Unit 235.

Sulphur Degassing and Solidification

The sulphur separated from the process gas in the Sulphur Condenser I-III flow by gravity via Sulphur Locks to the first compartment of the Sulphur Pit.

By circulation and the intensive contact with stripping air, the total of the sulphur is degassed

Sulphur can be delivered in liquid form or in a separate working step as flakes.

$\overline{\mathbf{3}}$ **Methanol Synthesis and MTG**

Methanol Synthesis (Unit 331) 3.1

$3.1.1$ **Synthesis Loop**

The make-up gas coming from the Front-End Section is mixed with recovered hydrogen and compressed in MUG Compressor.

The synthesis loop recycle stream is mixed with the make-up gas coming from the MUG compressor and enters the circulator where it is compressed to synthesis inlet pressure.

The syngas mixture is first sent to the Feed Preheater, where it is heated to the reactor inlet temperature while cooling down the reactor product gas. The converter inlet temperature is controlled by bypassing feed shell side of the heat exchanger. The feed gas enters the MeOH converter and directly flows to the catalyst where methanol production reaction proceeds.

Crossing the isothermal zone, the reaction goes on while temperature is kept almost constant by releasing the reaction heat to heatexchanger plates in the reaction area where steam is raised.

The product gas leaves the converter passing the gas-gas exchanger where it is cooled. The methanol formed in the product gas is condensed in the product condenser (air and water cooler); in the Flash Drum I the liquid raw product is separated from the recycle gas, which is sent to the suction of the recycle compressor.

A gas stream is purged from the synthesis loop and it is sent to the common Hydrogen Recovery Unit of the overall plant.

The raw methanol from is let down in the Flash Drum II where most of the dissolved gases are flashed and separated from the liquid.

The flash gas is sent to the B.L.'s via a pressure control valve while the raw methanol is made available at Battery Limits via a level control valve.

$3.1.2$ **Steam Generation System**

The heat exchange plates inside the MeOH Converter act as a boiler with forced circulation.

BFW is added directly to the Steam Drum. The circulation in the exchanger plates is ensured by BFW pumps.

The saturated MP steam generated by recovering the reaction heat is sent pressure controled to the Battery Limits. The steam pressures can be adjusted to control the boiling temperature in the Steam Generation System separately for external and internal plates; accordingly, the heat transfer in the catalyst is strictly controlled.

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Methanol to Gasoline (MTG) 3.2

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$3.2.1$ **MTG Reaction Section (Unit 332)**

Methanol Vaporizing

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DME and MTG Reactors

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Regeneration System

Gas - Liquid - Liquid Separation

$3.2.2$ **Separation Section (Unit 333)**

Deethanizer

Gasoline Splitter / Methanol Recovery

$3.2.3$ **Heavy Gasoline Treatment Section (Unit 334)**

HGT Reactors

HGT Product Stripper

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Air Separation Unit (Unit 478), **Instrument and Plant Air System (Unit 684)**

For Air separation a commercially available Package Unit (ASU) will be used,. The number of parallel units will be adjusted to the total oxygen demand and ASU capacity respectively. A liquid storage for both oxygen and nitrogen will be foreseen. Medium pressure nitrogen and high pressure will be taken out of an appropriate process step of the ASU, so compression can be saved.

For Instrument and Plant Air supply two sources will be used. Part of the air required will be taken from the air compressor in the, another part will be supplied by an Instrument and Plant Air Compressor, delivered with Instrument and Plant Air System. By this it is supply of instrument and plant air is ensured even if air separation units are out of operation, e.g. to operate MTG section and utilities. Compressed air will be dried for instrument air and fed to the plant via Instrument Air Receiver. Furthermore compressed air will be fed to the plant without drying as plant via Plant Air Receiver.

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Utilities 5

5.1 Nitrogen / HP Gas System (Unit 586)

The Nitrogen/ HP gas system provides nitrogen on different pressure levels, as well as HP gas on different temperature levels to all consumers within the plant. For LP Level nitrogen already pressurized is taken from the ASU. The HP gas will be CO₂ taken from the Rectisol Unit235, which is compressed by $CO₂$ Compressor. It can be used directly with compressor outlet temperature of 120°C in the Coal Feeding Unit 1121/2 and after further heating by Blow Back Heater in the Gasification Unit 113.

For start-up Nitrogen from ASU is fed to the CO₂ Compressor, compressed and used as HP Gas.

Medium Pressure Nitrogen is fed from ASU to Nitrogen Compressor. Pressurized Nitrogen is sent to consumers via MP Nitrogen Buffer Vessel.

5.2 **Chemical Distribution (Unit 619)**

A 20 wt% caustic solution is used to minimize winterising requirements. Since caustic is normally marketed at higher concentrations, the caustic is diluted with demin water from B.L. during the filling of the caustic storage tank via the caustic mixer. To unload the caustic the caustic fill pump is used. In case winterisation is required, caustic will be circulated permanently by Caustic Injection Pump via the Caustic Preheater. Supply of caustic to the plant is also done by Caustic Injection Pump.

Hydrochloric acid in a 15 wt% acid solution is used in the gasification slurry system to reach the necessary pH-concentration. Since acid is normally marketed at higher concentrations, on site dilution is foreseen with demin water. The demin water used for dilution is routed via the scrubber to the Acid Storage Tank to wash out acid traces from the vent gases leaving the tank during loading time.

To unload the acid the acid fill pump is used. In case winterisation is required, acid will be circulated permanently by Acid Injection Pump via the Acid Preheater. Supply of Acid to the plant is also done by Acid Injection Pump.

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5.3 Flare System (Unit 665)

For flare gases there are foreseen separate headers in the gasification plant, MTG plant and for sour gases. For sour gas an extra flare stack is provided. The other flare gases are sent via Flare Condensate Knock Out Drum to the Flare Stack. Both flare stacks are equipped with mol sieves to prevent back flow of air into the system. Both flare stacks get nitrogen blanketing. At the top of the stacks are Pilot Burners

Condensate will be transferred by Flare Condensate Pump to waste water treatment.

5.4 **Fire Fighting System (Unit 699)**

Fire fighting will be done mainly by two ring headers, one for the gasifcation ring header (which needs enhanced pressure because of its height, and one for the other parts of the plant. Both ring headers are equipped with hydrants.

Fire fighting water will be held available in the Fire Water Tank. A Jockey Pump will keep the system under pressure. Fire Water Pump (electric drive and diesel engine drive) will supply the ring headers with fire fighting water. Pressure for the gasification ring header is enhanced by Fire Water Booster Pump.

Further Fire Fighting equipment such as estinguishers etc. will be considered during further engineering.

PSA System (Unit 335) 6

6.1 General

The pressure swing adsorption process is based on physical adsorption phenomena, whereby highly volatile compounds with low polarity as represented by hydrogen or helium, are practically non-adsorbable compared to molecules such as CO2, CO, N2 and hydrocarbons. Hence most impurities in a hydrogen-containing stream can be selectively adsorbed and high-purity hydrogen product is obtainable.

The pressure swing adsorption process is working between two pressure levels:

- Adsorption of impurities is carried out at high pressure to increase the partial \bullet pressure and, therefore, the loading of the impurities on the adsorbent material.
- Desorption or regeneration takes place at low pressure to reduce the residual \bullet loading of the impurities as much as possible, in order to achieve a high product purity, high delta loading adsorption I desorption and subsequently a high hydrogen recovery.

The process works at ambient temperature. There is no heat required for the regeneration. Changes in temperature are caused only by heat of adsorption and desorption and depressurization. This feature results in an extremely long lifetime of the adsorbent material as no hydrothermal effect will deactivate the adsorbent.

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Adsorption and Regeneration Cycles 6.2

Adsorption

The feed gas flows through the adsorbers in upward direction. The impurities are selectively adsorbed - water, heavy hydrocarbons, light hydrocarbons, CO and nitrogen - from bottom to the top. High-purity hydrogen flows to the product line. The adsorbers on adsorption are on staggered cycles resulting in a highly flexible purification unit which is not influenced by fluctuations of the composition, temperature and pressure of the feed gas.

The proposed PSA system allows a high performance by maximum utilization of the hydrogen stored in an adsorber at the end of adsorption for pressure equalization, repressurization and purging of other adsorbers.

Regeneration

After the adsorption step, the adsorber is regenerated in four basic steps:

- The adsorber is depressurized to a low-pressure level co-current to the feed \bullet flow. The co-current depressurization uses the hydrogen stored in the adsorber to repressurize and purge other adsorbers.
- The adsorber is depressurized in the counter-current direction to tail gas pressure (blow-down step) to remove the impurities from the adsorbent.
- The adsorber is purged at tail gas pressure with pure hydrogen to desorb the residual impurities from the adsorbent.
- The adsorber is re-pressurized to adsorption pressure with pure hydrogen coming from adsorbers on the depressurization step.

Pressure Equalization

In order to recover most of the hydrogen stored in an adsorber at the end of the adsorption step, several equalizations are performed.

6.3 **Hydrogen Product**

High purity hydrogen according to specification is discharged to the hydrogen product line from the top of the adsorber vessels presently on adsorption. Its pressure is equivalent to feed gas pressure minus pressure loss across the PSA unit.

6.4 **Tail Gas**

The tail gas system homogenizes variations in tail gas composition, flow and pressure. It consists of the tail gas drum(s) and appropriate control devices.

The control of the tail gas is performed as follows: The tail gas flows out of the tail gas system under flow control. The set point of the flow controller is provided by the process control system taking into account feed flow, purge gas flow and tail gas system conditions.

The tail gas stream is compressed and recycled to the CO Shift unit.

MTG Process Flow Diagrams

 $\frac{1}{2}$

REDACTED - CLAIM OF CONFIDENTIALITY 12-08-08

NW, (2)
Na Nul, 1969, 2079