



# Reagent Chemical & Research, Inc.

115 US HIGHWAY 202 • RINGOES, NEW JERSEY 08551

OFFICE: (908) 284-2800 • FAX: (908) 284-2113

April 22, 2016

Division of Environmental Protection  
Division of Air Quality  
Attn: Beverly McKeone  
601 - 57<sup>th</sup> Street  
Charleston, WV 25304



RE: Class 1 Administrative Amendment for Air Permit  
Due to Equipment Removal, Permit # R13-1789C

Dear Beverly McKeone:

In response to our recent telephone conversation, I am writing this letter to you to request a class 1 administrative amendment to the air permit for Reagent Chemical's facility located in Institute, WV.

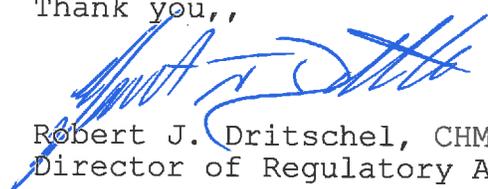
The current air permit, Permit # R13-1789C - Issue Date December 29, 2011, is the permit that Reagent Chemical would like to amend. A removal of a piece of equipment is the reason for the amendment request.

Reagent Chemical would like to remove the scrubber that services an area of inactivity at the facility. Reagent Chemical proposes to remove the scrubber identified in the permit as 2C. All other equipment that had been serviced by this scrubber, which includes some inactive storage tanks and process equipment, will be mothballed in place.

I have attached copies of the current permit pages 1 thru 3, with the proposed eliminations identified that would result from this removal of equipment. I have also enclosed an executed Permit Determination form with a permit modification narrative, a current MSDS, a plot plan and location map, the WV business certificate, a block diagram from the previous permit identifying the items to be removed from the permit and a block diagram for the proposed future emission points.

If you have any questions or comments, or if anything further is needed, please feel free to call me at (908) 284-2800.

Thank you,,

  
Robert J. Dritschel, CHMM  
Director of Regulatory Affairs



Attachment: Permit R13-1789B, pages 1-3.

Reagent Chemical & Research, Inc.  
CC: J. Macyszewicz - Reagent Chemical, Institute, WV  
Registered to ISO 9001; 2008  
Certificate # 10004240 QM08



WEST VIRGINIA  
DEPARTMENT OF ENVIRONMENTAL PROTECTION  
DIVISION OF AIR QUALITY  
601 57<sup>th</sup> Street, SE  
Charleston, WV 25304  
Phone: (304) 926-0475  
www.dep.wv.gov/daq

**PERMIT DETERMINATION FORM  
(PDF)**

FOR AGENCY USE ONLY: PLANT I.D. # \_\_\_\_\_  
PDF # \_\_\_\_\_ PERMIT WRITER: \_\_\_\_\_

1. NAME OF APPLICANT (AS REGISTERED WITH THE WV SECRETARY OF STATE'S OFFICE):  
**Reagent Chemical & Research, Inc.**

2. NAME OF FACILITY (IF DIFFERENT FROM ABOVE):  
**Reagent Chemical - Institute**

3. NORTH AMERICAN INDUSTRY CLASSIFICATION SYSTEM (NAICS) CODE:  
**424690**

4A. MAILING ADDRESS:  
**115 US Hwy 202 Ringoes, NJ 08551**

4B. PHYSICAL ADDRESS:  
**Rte 25 Institute, WV 25112**

5A. DIRECTIONS TO FACILITY (PLEASE PROVIDE MAP AS ATTACHMENT A):  
**Adequate to Route 25, next to Interstate 64 Institute exchange**

5B. NEAREST ROAD:  
**Route 25**

5C. NEAREST CITY OR TOWN:  
**Institute**

5D. COUNTY:  
**Kanawha**

5E. UTM NORTHING (KM):  
**4248**

5F. UTM EASTING (KM):  
**432**

5G. UTM ZONE:  
**17**

6A. INDIVIDUAL TO CONTACT IF MORE INFORMATION IS REQUIRED:  
**Robert Dritschel**

6B. TITLE:  
**Dir of Regulatory Affairs**

6C. TELEPHONE:  
**908-284-2800**

6D. FAX:  
**908-284-0374**

6E. E-MAIL:  
**bdritschel@reagentchemical.**

7A. DAQ PLANT I.D. NO. (FOR AN EXISTING FACILITY ONLY):  
**\_039 - \_00071**

7B. PLEASE LIST ALL CURRENT 45CSR13, 45CSR14, 45CSR19 AND/OR TITLE V (45CSR30) PERMIT NUMBERS ASSOCIATED WITH THIS PROCESS (FOR AN EXISTING FACILITY ONLY):  
**R13-1789C**

7C. IS THIS PDF BEING SUBMITTED AS THE RESULT OF AN ENFORCEMENT ACTION? IF YES, PLEASE LIST:  
**No**

8A. TYPE OF EMISSION SOURCE (CHECK ONE):  
 NEW SOURCE     ADMINISTRATIVE UPDATE  
 MODIFICATION     OTHER (PLEASE EXPLAIN IN 11B)

8B. IF ADMINISTRATIVE UPDATE, DOES DAQ HAVE THE APPLICANT'S CONSENT TO UPDATE THE EXISTING PERMIT WITH THE INFORMATION CONTAINED HEREIN?  
 YES     NO

9. IS DEMOLITION OR PHYSICAL RENOVATION AT AN EXISTING FACILITY INVOLVED?     YES     NO

10A. DATE OF ANTICIPATED INSTALLATION OR CHANGE:  
**06/01/2016**

10B. DATE OF ANTICIPATED START-UP:  
**Currently Operating**

11A. PLEASE PROVIDE A DETAILED PROCESS FLOW DIAGRAM SHOWING EACH PROPOSED OR MODIFIED PROCESS EMISSION POINT AS ATTACHMENT B.

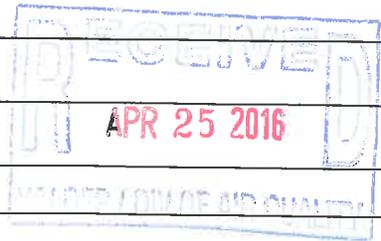
11B. PLEASE PROVIDE A DETAILED PROCESS DESCRIPTION AS ATTACHMENT C.

12. PLEASE PROVIDE MATERIAL SAFETY DATA SHEETS (MSDS) FOR ALL MATERIALS PROCESSED, USED OR PRODUCED AS ATTACHMENT D. FOR CHEMICAL PROCESSES, PLEASE PROVIDE A MSDS FOR EACH COMPOUND EMITTED TO AIR.

**13A. REGULATED AIR POLLUTANT EMISSIONS:**

⇒ FOR A NEW FACILITY, PLEASE PROVIDE PLANT WIDE EMISSIONS BASED ON THE POTENTIAL TO EMIT (PTE) FOR THE FOLLOWING AIR POLLUTANTS INCLUDING ALL PROCESSES.  
 ⇒ FOR AN EXISTING FACILITY, PLEASE PROVIDE THE PROPOSED CHANGE IN EMISSIONS BASED ON THE PTE OF ALL PROCESS CHANGES FOR THE FOLLOWING AIR POLLUTANTS.  
 PTE FOR A GIVEN POLLUTANT IS TYPICALLY BEFORE AIR POLLUTION CONTROL DEVICES AND IS COLLECTED BASED ON THE MAXIMUM DESIGN CAPACITY OF PROCESS EQUIPMENT.

POLLUTANT	HOURLY PTE (LB/HR)	YEARLY PTE (TON/YR) (HOURLY PTE MULTIPLIED BY 8760 HR/YR) DIVIDED BY 2000 LB/TON
PM		
PM <sub>10</sub>		
VOCs		
CO		
NO <sub>x</sub>		
SO <sub>2</sub>		
Pb		
HAPs (AGGREGATE AMOUNT)	No Change	
TAPs (INDIVIDUALLY)*		
OTHER (INDIVIDUALLY)*		



\* ATTACH ADDITIONAL PAGES AS NEEDED

**13B. PLEASE PROVIDE ALL SUPPORTING CALCULATIONS AS ATTACHMENT E.**

CALCULATE AN HOURLY AND YEARLY PTE OF EACH PROCESS EMISSION POINT (SHOWN IN YOUR DETAILED PROCESS FLOW DIAGRAM) FOR ALL AIR POLLUTANTS LISTED ABOVE INCLUDING INDIVIDUAL HAP'S (LISTED IN SECTION 112(b) OF THE 1990 CAAA), TAP'S (LISTED IN 45CSR27), AND OTHER AIR POLLUTANTS (E.G. POLLUTANTS LISTED IN TABLE 45-13A OF 45CSR13, MINERAL ACIDS PER 45CSR7, ETC.).

**14. CERTIFICATION OF DATA**

I, ROBERT DRITSCHEL (TYPE NAME) ATTEST THAT ALL THE REPRESENTATIONS CONTAINED IN THIS APPLICATION, OR APPENDED HERETO, ARE TRUE, ACCURATE, AND COMPLETE TO THE BEST OF MY KNOWLEDGE BASED ON INFORMATION AND BELIEF AFTER REASONABLE INQUIRY, AND THAT I AM A **RESPONSIBLE OFFICIAL**\*\* (PRESIDENT, VICE PRESIDENT, SECRETARY OR TREASURER, GENERAL PARTNER OR SOLE PROPRIETOR) OF THE APPLICANT.

SIGNATURE OF RESPONSIBLE OFFICIAL: \_\_\_\_\_

TITLE: DIRECTOR OF REGULATORY AFFAIRS

DATE: 04/22/2016

\*\*THE DEFINITION OF THE PHRASE 'RESPONSIBLE OFFICIAL' CAN BE FOUND AT 45CSR13, SECTION 2.23.

**NOTE: PLEASE CHECK ENCLOSED ATTACHMENTS:**

- ATTACHMENT A     ATTACHMENT B     ATTACHMENT C     ATTACHMENT D     ATTACHMENT E

RECORDS ON ALL CHANGES ARE REQUIRED TO BE KEPT AND MAINTAINED ON-SITE FOR TWO (2) YEARS.

THE PERMIT DETERMINATION FORM WITH THE INSTRUCTIONS CAN BE FOUND ON DAQ'S PERMITTING SECTION WEB SITE:

[www.dep.wv.gov/daq](http://www.dep.wv.gov/daq)

*West Virginia Department of Environmental Protection*

*Earl Ray Tomblin  
Governor*

*Division of Air Quality*

*Randy C. Huffman  
Cabinet Secretary*

# Class I Administrative Update Permit



**R13-1789C**

*This permit is issued in accordance with the West Virginia Air Pollution Control Act (West Virginia Code §§ 22-5-1 et seq.) and 45 C.S.R. 13 — Permits for Construction, Modification, Relocation and Operation of Stationary Sources of Air Pollutants, Notification Requirements, Temporary Permits, General Permits and Procedures for Evaluation. The permittee identified at the facility listed below is authorized to construct the stationary sources of air pollutants identified herein in accordance with all terms and conditions of this permit.*

Issued to:  
**Reagent Chemical & Research, Inc.**  
Institute  
039-00071

  
\_\_\_\_\_  
*John A. Benedict*  
Director

*Issued: December 29, 2011 • Effective: December 29, 2011*

This permit will supercede and replace Permit R13-1849B.

Facility Location: Institute, Kanawha County, West Virginia

Mailing Address: 115 US Highway 202, Ríngoos, NJ 08551

Facility Description: HCl Distribution Facility

NAICS Codes: 424690

UTM Coordinates: 432 km Easting • 4248 km Northing • Zone 17

Permit Type: Class I Administrative Update

Description of Change:

Decrease total hydrochloric acid throughput caused by elimination of the supply of the untreated HCl.

*Any person whose interest may be affected, including, but not necessarily limited to, the applicant and any person who participated in the public comment process, by a permit issued, modified or denied by the Secretary may appeal such action of the Secretary to the Air Quality Board pursuant to article one [§§ 22B-1-1 et seq.], Chapter 22B of the Code of West Virginia. West Virginia Code §22-5-14.*

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*The source is not subject to 45CSR30.*

IN ACCORDANCE WITH THE PERMIT APPLICATION AND ITS AMENDMENTS, THIS PERMIT IS LIMITED AS FOLLOWS:

A. SPECIFIC REQUIREMENTS

REMOVE

1. Maximum emissions to the atmosphere from Emission Point ID No. 1E following the hydrogen chloride (HCl) water scrubber (Control Device ID No. 2C) and carbon adsorption system (Equipment ID No. 1C) shall not exceed the following:

Chemical	Maximum Emission Rate	
	Hourly (lb/hr)	Annual (TPY)
Hydrogen Chloride (HCl)	0.01	0.01

2. Maximum emissions to the atmosphere at Emission Point ID No. 2E following the HCl water scrubber (Control Device ID No. 4C) shall not exceed the following:

Chemical	Maximum Emission Rate	
	Hourly (lb/hr)	Annual (TPY)
Hydrogen Chloride (HCl)	0.01	0.01

3. The HCl water scrubber (Control Device ID No. 4C) shall be maintained and operated so as to reduce HCl emissions by 99.8% by weight or more.

REMOVE

4. The concentration of HCl acid vapor vented through Emission Point ID No. 1E shall not exceed 14.18 milligrams per dry standard cubic meter (mg/dscm).

5. The concentration of HCl acid vapor vented through Emission Point ID 2E shall not exceed 14.18 mg/dscm.

REMOVE

6. The HCl scrubber (Control Device ID No. 1C) utilized to control emissions of HCl vapors shall be operated at a minimum liquor flow rate of 2 gallons per minute.

7. The HCl scrubber (Control Device ID No. 4C) utilized to control emissions of HCl vapors shall be operated at a minimum liquor flow rate of 3 gallons per minute.

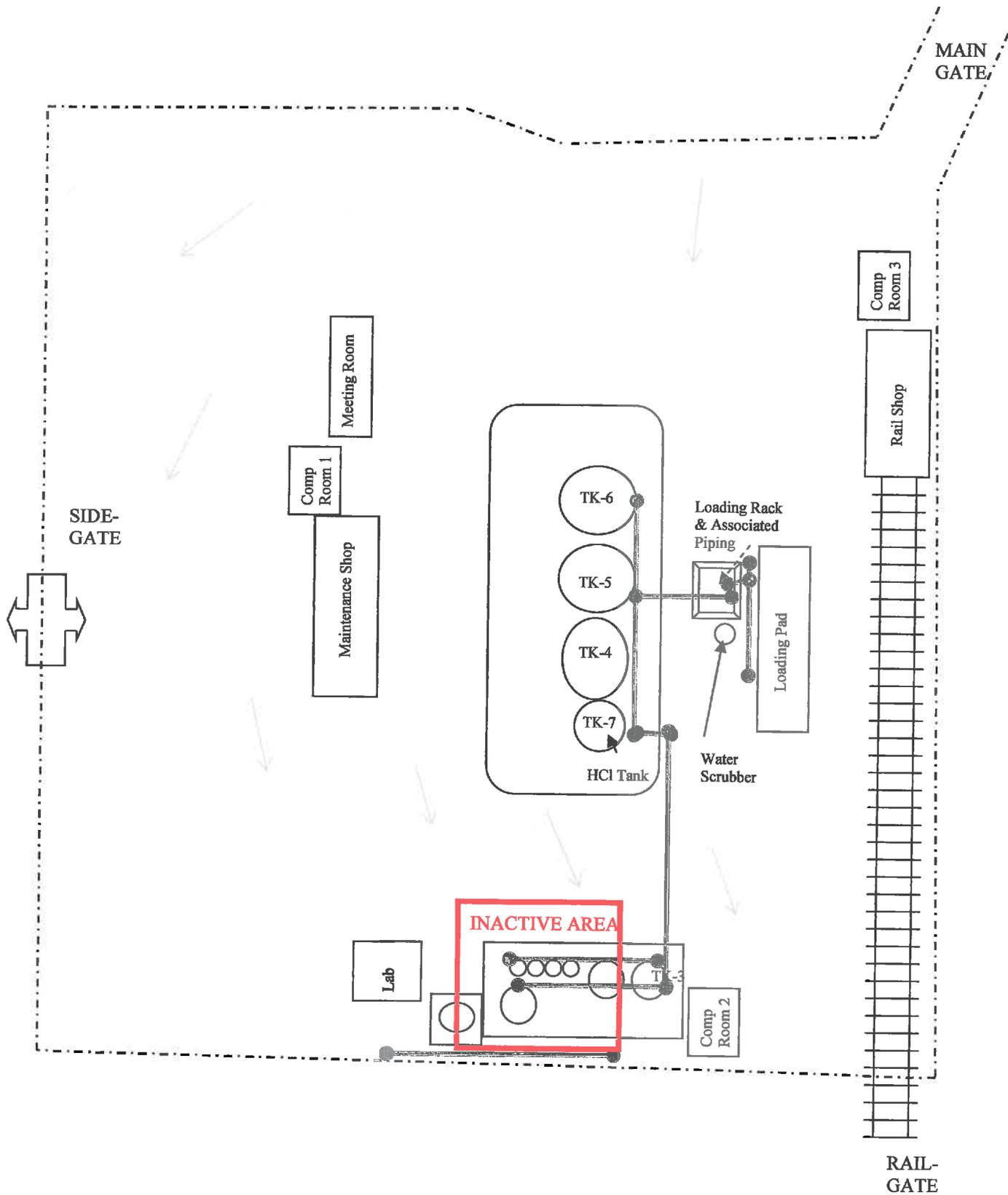
REMOVE

8. The maximum fill rate/throughput of untreated (raw) HCl to tank TK-1B shall not exceed 45 gallons per minute.

9. The maximum fill rate/throughput of treated HCl to tank TK-7 shall not exceed 125 gallons per minute.

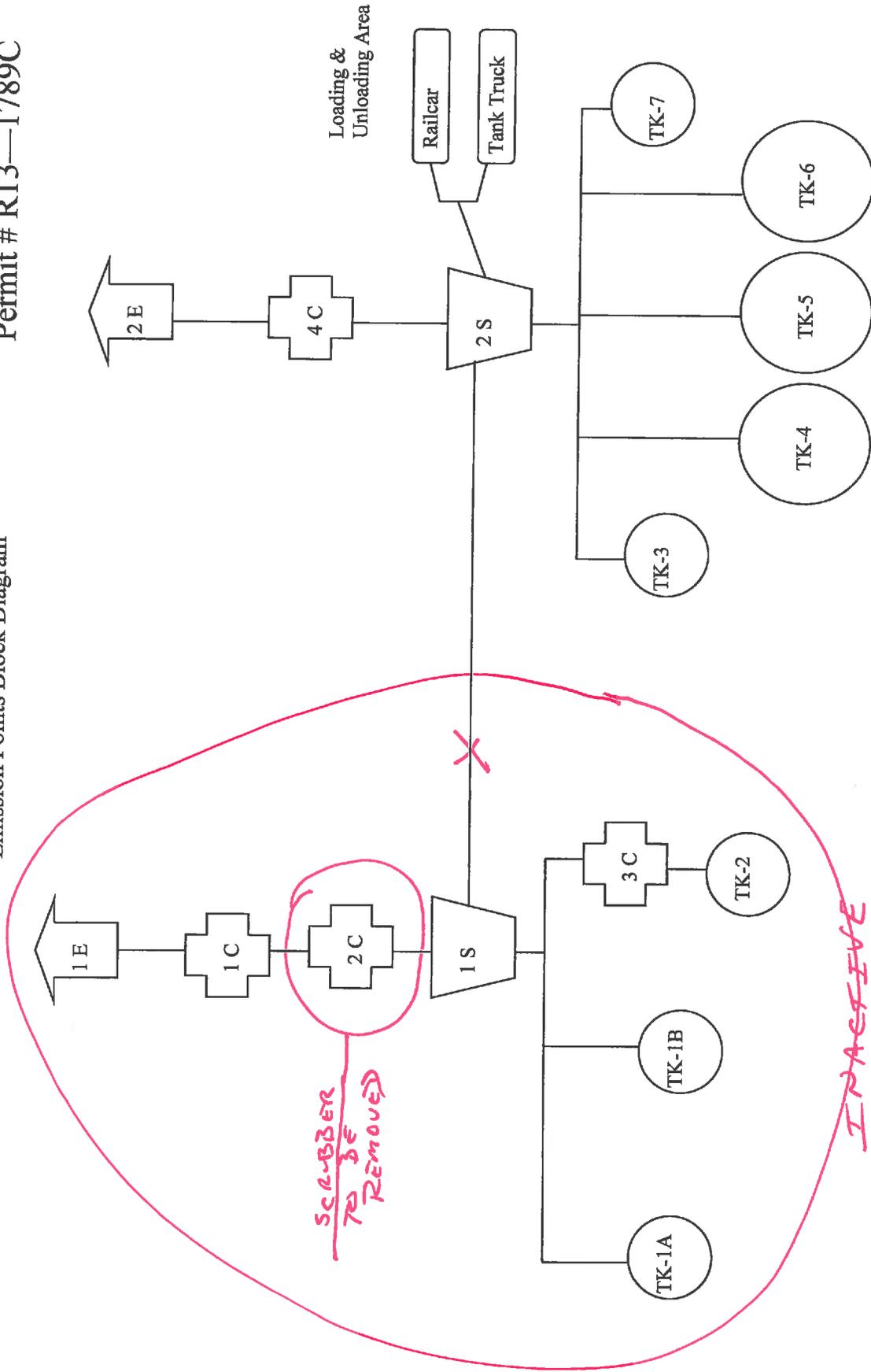
10. The throughput of off-site treated HCl to be unloaded for storage to the facility from tank trucks and railcars shall not exceed 20,000,000 gallons/year (based on 1 jumbo railcar/week, 5 days per week, and 52 weeks/yr). Annual throughput shall be based on a 12-month rolling total.

11. Railcar air pad venting of HCl laden air to the HCl water scrubber (Control Device ID No. 4C) to reduce railcar pressure (25 psig maximum) to 0 psig (atmospheric pressure) shall take a minimum of two (2) hours to complete.



INSTITUTE, WV FACILITY  
Emission Points Block Diagram

Permit # R13—1789C



SCRUBBER  
TO BE  
REMOVED

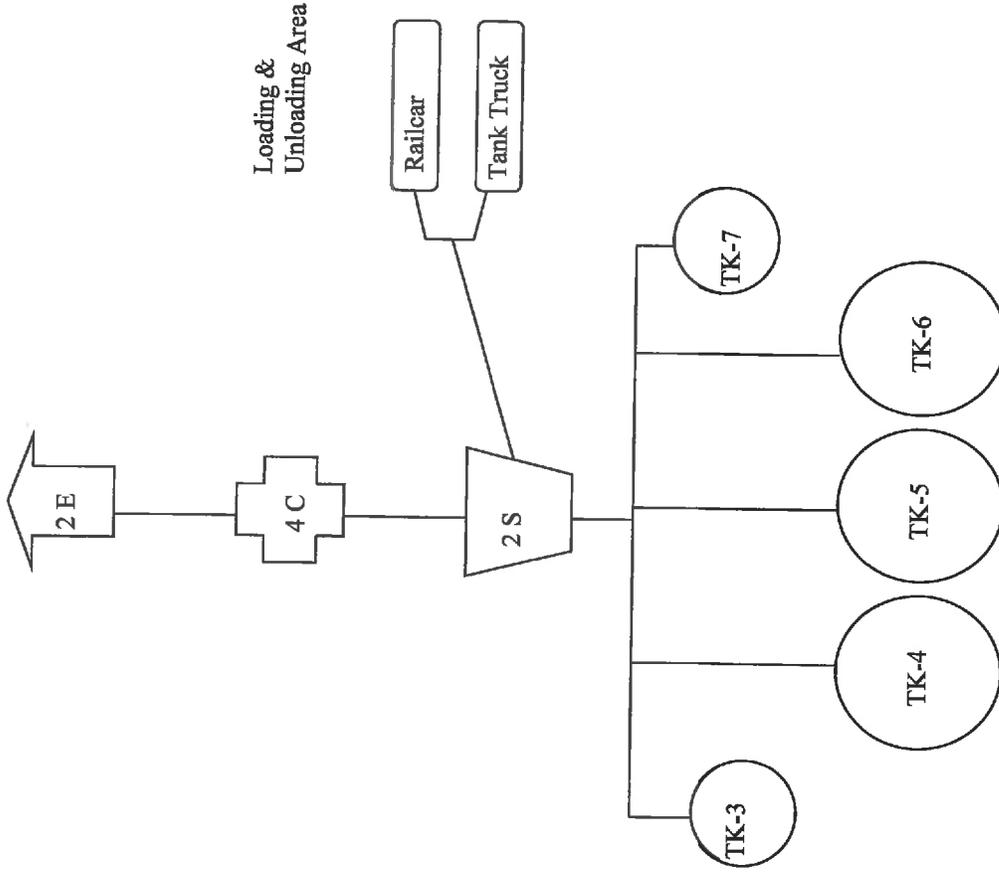
INACTIVE  
AREA

Request  
Removal  
From  
Permit

- 1S & 2S = Combined Source
- 1C = Carbon Drum
- 3C = Carbon Pot
- 2C & 4C = Wet Scrubbers
- 1E & 2E = Emission Points

INSTITUTE, WV FACILITY  
Emission Points Block Diagram

Amendment Requested



TK-# - Storage Tanks  
2S = Combined Source  
4C = Wet Scrubber  
2E = Emission Points

## Process Description for Institute, Facility

### Operation 1 - Railcar Unloading

The contents of railroad tank cars, 35.21% Hydrochloric Acid solution, is pumped directly into a storage tank, using a 3 inch transfer line. At the same time, 26.74 CFM of air displaced from the storage tank is transferred directly back to the railcar being emptied. The loading/unloading lines terminate with the vent lines to the packed tower water scrubber (4C).

### Operation 2 - Tank Truck Loading

5000 gallons of 35.21% Hydrochloric Acid solution, is pumped at a rate of 200 gpm directly into the tank truck using a 3 inch transfer line. At the same time, 26.74 CFM of air displaced from the tank truck is transferred directly back to the storage tank being emptied. Although the loading/ unloading lines terminate with the vent lines to the packed tower water scrubber, (4C) essentially there are no vapors emitted to the atmosphere during this operation.

**Safety Data Sheet**

**GHS-Compliant**

May be used to comply with  
OSHA's Hazard Communication Standard  
29 CFR 1910.1200. Standard must be  
consulted for specific requirements.



REAGENT CHEMICAL & RESEARCH, INC.  
115 US Hwy 202 Ringoes, NJ 08551

**PRODUCT IDENTITY**

Hydrochloric Acid, 20° or 22° Baume

Safety Data Sheet Revision Date - June 11, 2015

**Section 1 - Identification**

Product Name	CAS #
Hydrochloric Acid	7647-01-0
Synonym	Chemical Formula
Muriatic Acid	HCl
Chemical Name	Chemical Family
Hydrochloric Acid Solution	Inorganic Acid
Product Use	
Acidification, pH Adjustment	
Manufacturer/Supplier Name	Address
Reagent Chemical & Research, Inc.	115 US Hwy 202 Ringoes, NJ 08551
General Information	Country
1-908-284-2800	United States
Emergency Telephone	Transportation Emergency Number
1-409-899-3400	CHEMTREC 1-800-424-9300

**Section 2 - Hazards Identification**

**GHS Classification:**

HEALTH	PHYSICAL
Serious Eye Damage - Category 1	Corrosive to Metals - Category 1
Skin Corrosion - Category 1 B	
Sensitization, Respiratory - Category 1	
Specific Target Organ Toxicity (single exposure) - (Respiratory System) - Category 2	
Specific Target Organ Toxicity (repeated exposure) - (Respiratory System) - Category 2	

**GHS Label Elements:**

**SYMBOLS: corrosion, health hazard**



**Signal Word: DANGER**

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**Section 2 - Hazards Identification (continued)**

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**GHS Label ELEMENTS:**

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***Hazard Statements***

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Causes severe skin burns & eye damage

May cause allergic or asthmatic symptoms or breathing difficulties if inhaled

May cause damage to organs (respiratory system) if inhaled

May cause damage to organs (respiratory system) through prolonged or repeated exposure

May be corrosive to metals

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***Precautionary Statements***

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**PREVENTION**

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Do not breathe dusts/fume/gas/mist/vapors/spray

Wash face, hands and exposed skin thoroughly after handling

Wear protective gloves/protective clothing/eye protection/face protection

In case of inadequate ventilation, wear respiratory protection

Do not eat, drink or smoke when using this product

Keep only in original container

**RESPONSE**

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IF SWALLOWED: Rinse mouth. DO NOT induce vomiting

IF ON SKIN(or hair): Take off immediately all contaminated clothing. Rinse skin with  
water/shower. Wash contaminated clothing before reuse.

IF INHALED: Remove person to fresh air and keep comfortable for breathing.

Immediately call emergency medical professional or Poison Control Center

Specific treatment (See Section 4)

If in eyes: Rinse cautiously with water for several minutes. Remove contact lenses,  
if present and easy to do.

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Absorb spillage to prevent material damage

**STORAGE**

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Store locked up

Store in corrosive resistant container/container with resistant inner liner

**DISPOSAL**

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Dispose of contents/container in accordance with federal and state regulations

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### Section 3 - Composition / Information on Ingredients

Component Description	Percent	CAS #
Hydrogen Chloride	26.00 - 37.00	7647-01-0
Water	63.00 - 74.00	7732-18-5

### EXPOSURE LIMITS/REGULATORY INFORMATION

Substance	PEL	TLV	STEL	TWA	CEILING
Hydrogen Chloride	C-7 mg/m3	C-2 ppm	50 ppm	N/D	5 ppm
Water	N/D	N/D	N/D	N/D	N/D

N/D - Not Determined      C = Ceiling Level

### Section 4 - First Aid Measures

#### General

If a known exposure occurs or is suspected, immediately initiate the recommended procedures below. Simultaneously contact a physician, or the nearest Poison Control Center. Inform the person contacted of the type and extent of exposure, describe the victim's symptoms and follow the advice given. For additional information, call day or night, Reagent Chemical (409) 899-3400 or Chemtrec (800) 424-9300.

#### Inhalation

Remove from contaminated atmosphere. If breathing has ceased, clear the victim's airway and start mouth-to-mouth artificial respiration, which may be supplemented by the use of a bag-mask respirator, or a manually-triggered, oxygen supply capable of delivering 1 liter/second or more. If the victim is breathing, oxygen may be administered from a demand-type or continuous-flow inhalator, preferably with a physician's advice. Contact a physician immediately.

#### Eye Contact

Immediately flush the eyes with large quantities of running water for 15 minutes. Hold the eyelids apart during the flushing to ensure rinsing of the entire surface of the eyes and lids with water. DO NOT attempt to neutralize with chemical agents. Obtain medical attention as soon as possible. Oils or ointments should not be used. Continue the flushing for an additional 15 minutes if the physician is not available.

#### Skin Contact

Immediately remove contaminated clothing under a safety shower. Flush all affected areas with large amounts of water for 15 minutes. DO NOT attempt to neutralize with chemical agents. Obtain medical advice.

#### Ingestion

DO NOT induce vomiting. Immediately give large quantities of water or milk, if available. If vomiting does occur, give fluids again. Never give anything by mouth to an unconscious person. Call a physician or the nearest Poison Control Center.

#### Medical Conditions Generally Aggravated by Exposure

Hydrogen Chloride will aggravate breathing disorders

#### Note to Physician

Attending Physician should treat exposed patients symptomatically

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## **Section 5 - Fire Fighting Measures**

### **Extinguishing Method**

Not Applicable, use water to dilute spills and to flush them away from ignition sources.

### **Unusual Fire and Explosion Hazards**

Non-flammable, but Hydrochloric Acid reacts with metals.

### **Special Firefighting Procedures**

Non-flammable, but Hydrochloric Acid reacts with all metals, except gold and

platinum, with rapid evolution of Hydrogen which is flammable and explosive in air.

Firefighters exposed to Hydrochloric Acid vapors should wear Scott Air-Pak, or

equivalent. Hydrogen Chloride vapors are extremely irritating to the respiratory

tract and may cause breathing difficulty.

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## **Section 6 - Accidental Release Measures**

### **Steps to be Taken in Case Material is Released or Spilled**

Spills or discharges into the environment involving large quantities of Hydrochloric

Acid should be controlled and cleaned-up according to a pre-determined, affirmative

written Spill Prevention and Control Program. For assistance in developing a SPCP

contact your nearest Reagent Sales Office. Refer to Section 15 for spill/release

reporting information.

Spills should be handled immediately by neutralization and dilution of the spilled

product by the use of Soda Ash (Sodium Carbonate), Lime (Calcium Hydroxide), or

Limestone (Calcium Carbonate) with large amounts of water. For an interior (inside

a closed space) spill be aware that the use of Soda Ash, Lime and Limestone will

evolve heat and carbon dioxide and that ample ventilation must be provided.

### **Waste Disposal**

Under Federal RCRA, it is the responsibility of the user of products to determine,

at the time of disposal, whether the product falls under RCRA as a hazardous waste.

This is because product uses, transformations, mixtures, etc. may render the

resulting end-product hazardous.

### **Container Disposal**

Containers should be cleaned of residual product before disposal. Empty containers

should be disposed of in accordance with all applicable laws and regulations.

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## **Section 7 - Handling and Storage**

### **Handling**

Chemical goggles and full face shield must be worn at all times by personnel

exposed to or handling Hydrochloric Acid. The use of a NIOSH approved cartridge

respirator or a Scott Air-Pak should be used by all personnel exposed.

### **Storage**

Store containers in a cool, dry location away from direct sunlight, sources of

intense heat, or where freezing may occur. Store material in acid-proof container.

Keep container tightly closed when not in use. Keep container away from incompatible

materials. All loading, unloading, and storage equipment must be inspected prior to

any transfer operations are initiated.

## Section 7 - Handling and Storage (continued)

### General Comments

Impervious clothing, gloves, footwear and head gear must be worn at all times

by personnel exposed to or handling Hydrochloric Acid.

### Precautions to be Taken in Handling and Storage

Make sure all personnel involved in housekeeping and spill clean-up follow good

Industrial Hygiene practices and wear proper protective equipment.

## Section 8 - Exposure Controls / Personal Protection

### EXPOSURE LIMITS

Substance	PEL	TLV	STEL	TWA	CEILING
Hydrogen Chloride	C-7 mg/m <sup>3</sup>	C-5 ppm	50 ppm	N/D	5 ppm
Water	N/D	N/D	N/D	N/D	N/D

N/D - No Data Available      C = Ceiling Level

### Respiratory Protection

Maintain airborne contaminate levels below listed guidelines. Use with adequate

ventilation. Use a mechanical fan or vent area to scrubber. Use NIOSH approved

respiratory protection if exposure limits are exceeded.

Ventilation	Local Exhaust	Special
	If PEL exceeded	Vent fumes to appropriate scrubber
	Mechanical (General)	Other
	If PEL exceeded	Not Applicable

### Skin Protection

Wear neoprene rubber gloves to minimize skin contact. Additional protection may be

necessary to prevent skin contact including use of apron, face shield, boots or full

body protection. A safety shower should be located in the work area.

### Eye Protection

Splash goggles or safety glasses. Face shields are recommended. Eye-wash stations

should be available where eye contact can occur.

### Other Protection

Use body protection appropriate for task. An apron or other impermeable body

protection is suggested. Full body chemical protection is recommended for

emergency response procedures.

## Section 9 - Physical and Chemical Properties

Boiling Point	230 F	Specific Gravity (H <sub>2</sub> O = 1)	1.13 - 1.19
Vapor Pressure (mm Hg)	50 - 60 mm	Freezing Point	-.12 F to -63 F
Vapor Density (AIR = 1)	No Data Available	Density	9.48 - 9.61
pH	< 1	Odor Threshold	0.25 - 10 ppm
Flash Point	Not Flammable	Evaporation Rate	No Data Available
Flammability	Not Flammable	Flammability Limits	Not Flammable
Auto Ignition Temperature	Not Flammable	Partition Coefficient	No Data Available
Viscosity (at 15 C)	2.3 mPa.s	Decomposition Temperature	No Data Available
Solubility in Water			

miscible

### Appearance and Odor

Clear/Slightly yellow with a sharp pungent odor

**Section 10 - Stability and Reactivity**

Stability	Unstable		Conditions to Avoid Hydrochloric Acid is extremely reactive. Avoid contact with metal surfaces and oxidizing agents.
	Stable	X	

**Incompatibility (Materials to Avoid)**

Hydrochloric Acid is chemically stable when properly contained and handled. It is a strong mineral acid and reacts with many metals and metal oxides and hydroxides to form the equivalent metal chloride. It reacts with zeolites and other silicious compounds to form Hydrosilicic Acid; it reacts with carbonates to form Carbon Dioxide and Water. It is oxidized by Oxygen or electrolysis to form Chlorine, a lethal, poisonous gas. It reacts with alkaline compounds to form a neutral salt. It is a hydrolyzing agent for carbohydrates, esters and other compounds.

It's reaction with most metals will produce Hydrogen, an explosive gas. Violent reactions will result when Hydrochloric Acid Reacts with acetic anhydride, 2-aminoethanol, ammonium hydroxide, calcium phosphide, chlorosulfonic acid, ethylene diamine, ethylene imine, oleum (fuming sulfuric acid), perchloric acid, beta propiolactone, propylene oxide, sodium hydroxide, sulfuric acid, uranium phosphide and vinyl acetate. This listing is not all-inclusive.

**Hazardous Decomposition or By-products**

Extreme heat may cause the product to decompose, producing toxic fumes which may include chlorine compounds.

Hazardous Polymerization	May Occur		Conditions to Avoid Extreme heat and contact with incompatible materials
	Will Not Occur	X	

**Section 11 - Toxicological Information**

Route(s) of Entry:	Inhalation?	Skin?	Ingestion?
	Yes	Yes	Yes

**Health Hazards (Acute and Chronic)**

Hydrogen Chloride, both as a gas and in a solution as Hydrochloric Acid, is a corrosive substance and can cause severe and painful burns on contact with any part of the body or if taken internally. The mucous membranes of the eyes and the upper respiratory tract are especially susceptible to the injurious effects of high atmospheric concentrations of Hydrogen Chloride. The gas or vapor is so penetrating and pungent that when high concentrations do occur, those exposed should immediately leave the contaminated area.

Carcinogenicity:	NTP?	IARC Monographs?	OSHA Regulated?
	No Data Available	No Data Available	No Data Available

**Signs and Symptoms of Exposure**

Exposure to Hydrochloric acid may cause severe burns at the contact points

**Medical Conditions Generally Aggravated by Exposure**

Exposure to fumes may aggravate dermatitis and breathing disorders.

## Section 11 - Toxicological Information (continued)

### Specific Target Organ Toxicity (Single Exposure)

Respiratory System - May cause respiratory injury/irritation

### Specific Target Organ Toxicity (Repeated Exposure)

Respiratory System - May cause respiratory injury/irritation

### Toxicology

Hydrogen Chloride

### Inhalation Data

Human LCLo - 1300 ppm/30 min

Rat LC<sub>50</sub> - 4701 ppm/30 min

### Oral (rabbit)

LD<sub>50</sub> - 900 mg/kg

### Oral (rat)

LD<sub>50</sub> - 700 mg/kg

### Dermal (rabbit)

LD<sub>50</sub> - 5010 mg/kg

### Germ Cell Mutagenicity

No Data Available

### Skin Corrosion/Irritation

Causes severe skin burns and eye damage pH <1

### Serious Eye Damage/Irritation

Causes severe eye damage pH <1

### Respiratory or Skin Sensitization

Corrosive to respiratory tract with concentrated or repeated exposures

## Section 12 - Ecological Information

### Ecological Toxicity

Animals exposed to hydrochloric acid solution will experience tissue damage, burns and may be killed. Plants contaminated with hydrochloric acid solutions of low pH may be adversely effected or destroyed. High concentrations have been shown to be detrimental to aquatic life. A release into a body of water will kill fish and other aquatic life.

### Other Ecological Information

Hydrochloric acid is stable and found naturally in the environment. All work practices should be aimed at eliminating environmental contamination.

### Chemical Fate Information

Hydrochloric acid is naturally occurring in the environment.

### Other Regulatory Information

No other regulatory information is available on this product.

## Section 13 - Disposal Considerations

As sold, this product, when discarded or disposed of, is a hazardous waste according to Federal regulations (40 CFR 261). It is listed as Hazardous Waste Number D002, listed due to its corrosivity. The transportation, treatment and disposal of this waste material must be conducted in compliance with 40 CFR 262, 263, 264, 268 and 270.

Disposal can occur only in properly permitted facilities. Refer to state and local statutes for any additional requirements, as they may differ from Federal laws.

### Waste Disposal

Under Federal RCRA, it is the responsibility of the user of products to determine, at the time of disposal, whether the product falls under RCRA as a hazardous waste.

This is because product uses, transformations, mixtures, etc. may render the resulting end-product hazardous.

### Container Disposal

Containers should be cleaned of residual product before disposal. Empty containers should be disposed of in accordance with all applicable laws and regulations.

**Section 14 - Transport Information**

**Regulated Material**

Hydrochloric Acid is defined as hazardous by the US DOT and Transport Canada

North American Emergency Response Guide Book

ID # 1789 Guide #157 2008 & 2012 Revision

**DOMESTIC SHIPPING INFORMATION**

Proper Shipping Name	Hydrochloric Acid	Hazard Classification	Corrosive
UN/NA Identification	UN 1789	Hazard Class	Class 8
DOT Labels Required	Corrosive	Packaging Group	II

**INTERNATIONAL SHIPPING INFORMATION**

Proper Shipping Name	Hydrochloric Acid	Hazard Classification	Corrosive
UN/NA Identification	UN 1789	Hazard Class	Class 8
Labels Required	Corrosive	Packaging Group	II

**Section 15 - Regulatory Information**

**U.S. Federal Regulations**

**Comprehensive Environmental Response and Liability Act of 1980 (CERCLA):**

Chemical Name: Hydrochloric Acid CAS # 7647-01-0 RQ - 5000 lbs

**Toxic Substances Control Act (TSCA):**

All components of this product are included on the TSCA inventory

**OSHA Hazard Communication Standard Classification:**

Corrosive as defined by the OSHA Hazard Communication Standard.

**Clean Water Act (CWA):**

Chemical Name: Hydrochloric Acid CAS # 7647-01-0 Listed as Hazardous

No chemical components listed as Priority pollutants or Toxic pollutants

**Clean Air Act (CAA):**

Hydrochloric acid, CAS 7647-01-0, is listed as a hazardous air pollutant (HAP)

**US Environmental Protection Agency Risk Management Plan (RMP) Regulated:**

No, Hydrochloric acid solution under 37% is not regulated

**Superfund Amendments and Reauthorization Act (SARA) Title III Information:**

SARA Section 302: Hydrochloric Acid CAS # 7647-01-0 TPQ 5000 lb EPCRA RQ

SARA Section 313: Hydrochloric Acid CAS # 7647-01-0

**National Sanitation Foundation Limits (ANSI/NSF Standard 60):**

Maximum Drinking Water Use Concentration - 40 mg/l

Scale and Corrosion Control at Maximum 40 mg/l

**State Regulations**

**California Safe Drinking Water Act (Prop 65) Listing:**

No ingredients listed in this section

**California Right to Know Act:**

Chemical Name: Hydrochloric Acid CAS # 7647-01-0

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**Section 15 - Regulatory Information (continued)**

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**New Jersey Right to Know Act:**

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Chemical Name: Hydrochloric Acid CAS # 7647-01-0

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Chemical Name: Water CAS # 7732-18-5

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**Massachusetts Right to Know Act Substance List (MSL)::**

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Chemical Name: Hydrochloric Acid CAS # 7647-01-0

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**Pennsylvania Right to Know Act Hazardous Substance List:**

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Chemical Name: Water CAS # 7732-18-5

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Chemical Name: Hydrochloric Acid CAS # 7647-01-0

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**International Regulations****Canadian Domestic Substance List (DSL) Inventory Listing:**

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Chemical Name: Hydrochloric Acid CAS # 7647-01-0

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**Canadian Ingredient Disclosure List**

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Chemical Name: Hydrochloric Acid CAS # 7647-01-0

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**Canadian Workplace Hazardous Materials Information System (WHMIS):**

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Class E: Corrosive material

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This product has been classified according to the hazard criteria of the CPR  
and the MSDS contains all of the information required by the CPR

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**European Inventory of Existing Chemicals (EINECS):**

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Chemical Name: Hydrochloric Acid EINECS # 2315957

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**EU Labeling in Accordance with EC Directives:**

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Hazard Symbols: C

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**EU Risk (R) and Safety (S) Phrases:**

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R23/24/25: Toxic by inhalation, in contact with skin and if swallowed

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R37/38: Irritating to respiratory system and skin

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R41: Risk of serious damage to eyes

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S36/37: Wear suitable protective clothing and gloves

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S45: In case of accident or if you feel unwell, seek medical advice immediately

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S53: Avoid exposure - obtain special instructions before use

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S61: Avoid release to the environment. Refer to safety data sheet

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**Japanese Minister of International Trade and Industry (MITI) Inventory Listing:**

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Chemical Name: Hydrochloric Acid SECTION STRUCTURE # 1-324

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**Australian Inventory of Chemical Substances (AICS) Listing:**

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Chemical Name: Hydrochloric Acid CAS # 7647-01-0

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**US Census Bureau - Foreign Trade Identification**

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Chemical Name: Hydrochloric Acid HTS & Schedule B # 2806.10.0000

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**Section 16 - Other Information**

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Created By Product Safety - 6/1/98	MSDS Revision Date June 11, 2015
MSDS Revision Number Revision # 011	Revision Indicator Hazard Statement Alignment
MSDS Contact Robert Dritschel 908-284-2800	
Does Product Contain, or is Manufactured with, CFC's? No	
National Fire Protection Association (NFPA) Ratings: Health - 3 Flammability - 0 Instability - 0 Other Hazard Information - ACID	
Hazardous Material Identification System (HMIS): Health - 3 Flammability - 0 Physical Hazard - 0 Protective Equipment - X	
North American Emergency Response Guide Book ID # 1789 Guide #157 2008 & 2012 Revision	

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**Disclaimer of Liability**

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**WEST VIRGINIA  
STATE TAX DEPARTMENT  
BUSINESS REGISTRATION  
CERTIFICATE**

ISSUED TO:  
**REAGENT CHEMICAL & RESEARCH INC  
W VA ROUTE 25  
INSTITUTE, WV 25330-0000**

**BUSINESS REGISTRATION ACCOUNT NUMBER: 1012-5273**

This certificate is issued on: 07/20/2010

*This certificate is issued by  
the West Virginia State Tax Commissioner  
in accordance with Chapter 11, Article 12, of the West Virginia Code.*

*The person or organization identified on this certificate is registered  
to conduct business in the State of West Virginia at the location above.*

*This certificate is not transferrable and must be displayed at the location for which issued.  
This certificate shall be permanent until cessation of the business for which the certificate of registration  
was granted or until it is suspended, revoked or cancelled by the Tax Commissioner.*

*Change in name or change of location shall be considered a cessation of the business and a new  
certificate shall be required.*

**TRAVELING STREET VENDORS:** Must carry a copy of this certificate in every vehicle operated by them.  
**CONTRACTORS, DRILLING OPERATORS, TIMBER/LOGGING OPERATIONS:** Must have a copy of  
this certificate displayed at every job site within West Virginia.

atL006 v.3  
L0680298240

MAILED ORIGINAL  
TO DAVE M.  
7/28/10

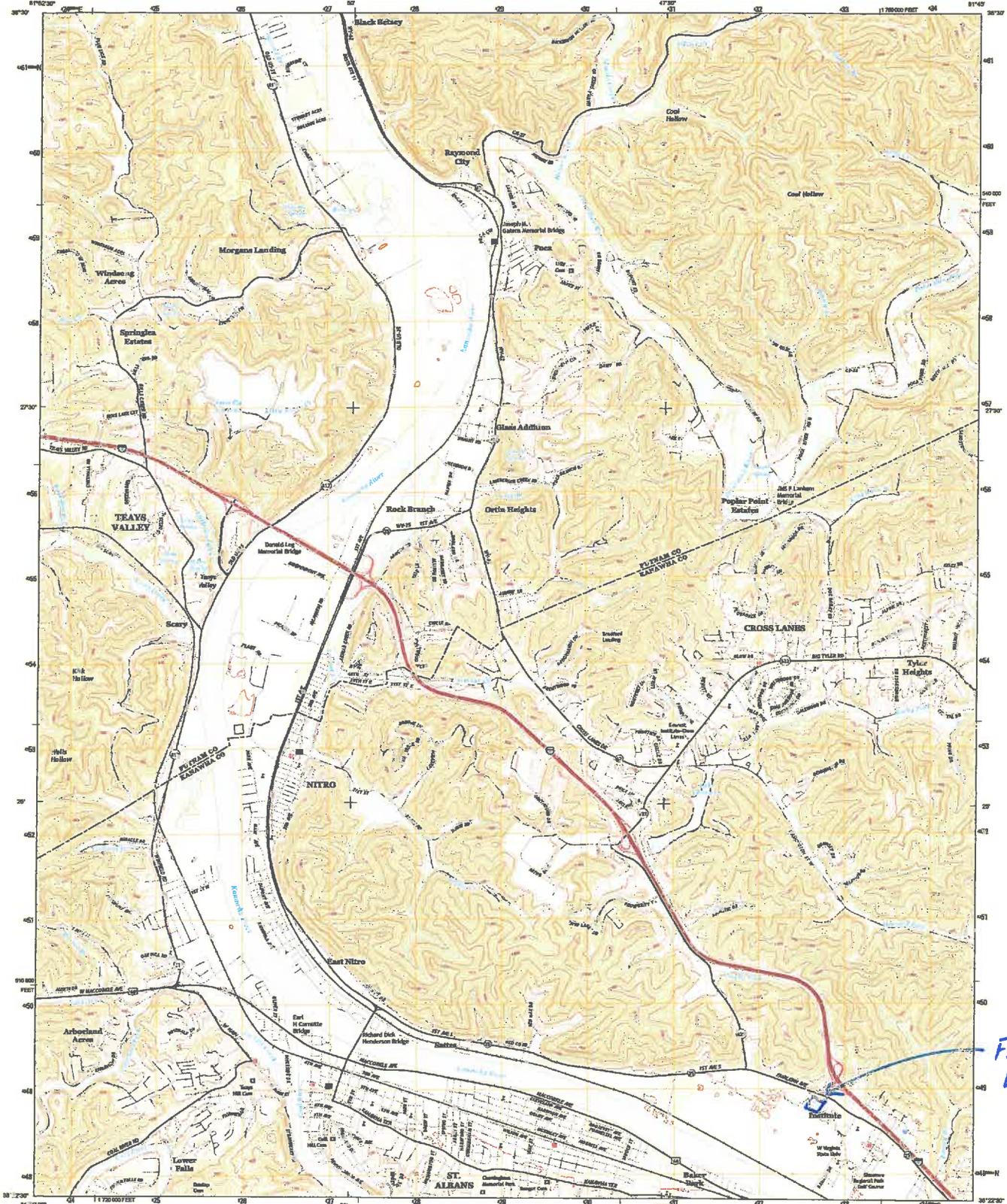


Google earth



**Facility Location**

**Reagent Chemical & Research, Inc.  
Route 25  
Institute, WV**



Facility  
Location

Produced by the United States Geological Survey  
North American datum of 1983 (NAD83)  
World Geodetic System of 1984 (WGS84), Projection and  
1:250,000 scale grid. Contour Interval: 20 Feet. Zone 17N  
of UTM-foot (GCS) - West Virginia Coordinate System of 1983  
datum.

This map is not a legal document. Readings may be  
generalized for this map scale. Private lands visible government  
reservations may not be shown. Other pertinent features  
including private lands.

Imagery: ..... NAD83, July 2011  
Base: ..... NAD83, 1983  
Hydrography: ..... NAD83, 2010  
Contours: ..... National Elevation Dataset, 2010  
Boundaries: ..... Multiple sources by the metadata file 1972 - 2015



ROAD CLASSIFICATION

Expressway	Local Connector
Secondary Hwy	Local Road
Interstate	400
Interstate Spur	US Route
	Main Road

1	2	3	1 Wetland
2	3	4	2 Swamp
3	4	5	3 Shrubland
4	5	6	4 Open Water
5	6	7	5 Pastureland
6	7	8	6 Cropland
7	8		7 Urban
8			8 Developed